



Novel impedance-spectroscopy process analytical technology for freeze drying process development

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Analytical challenges and solutions for medicines manufacturing towards Industry 4.0 event
KTN meeting at CPI, Darlington, 26 February 2020 13:55 – 14:15

Knowledge Transfer Network
Medicines Manufacturing
Challenge Community



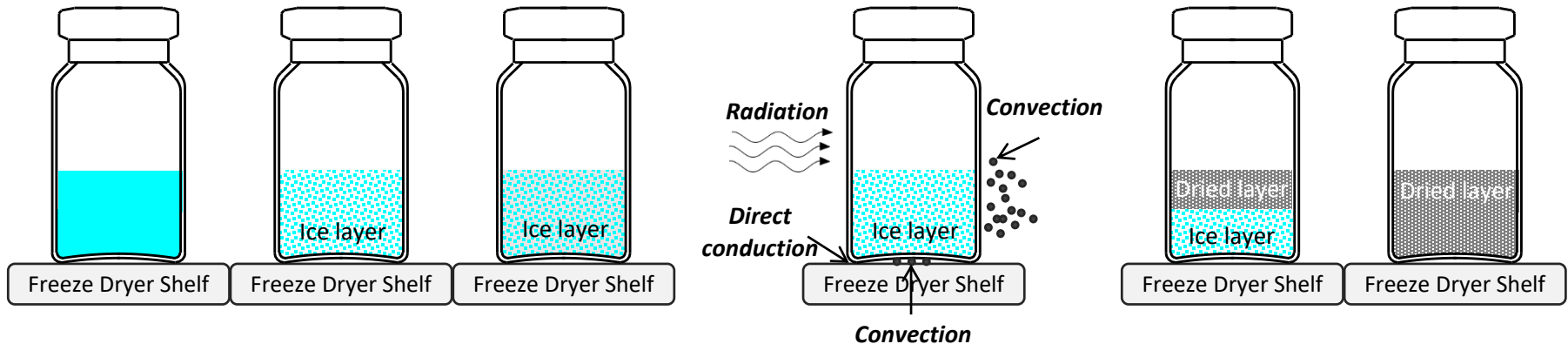
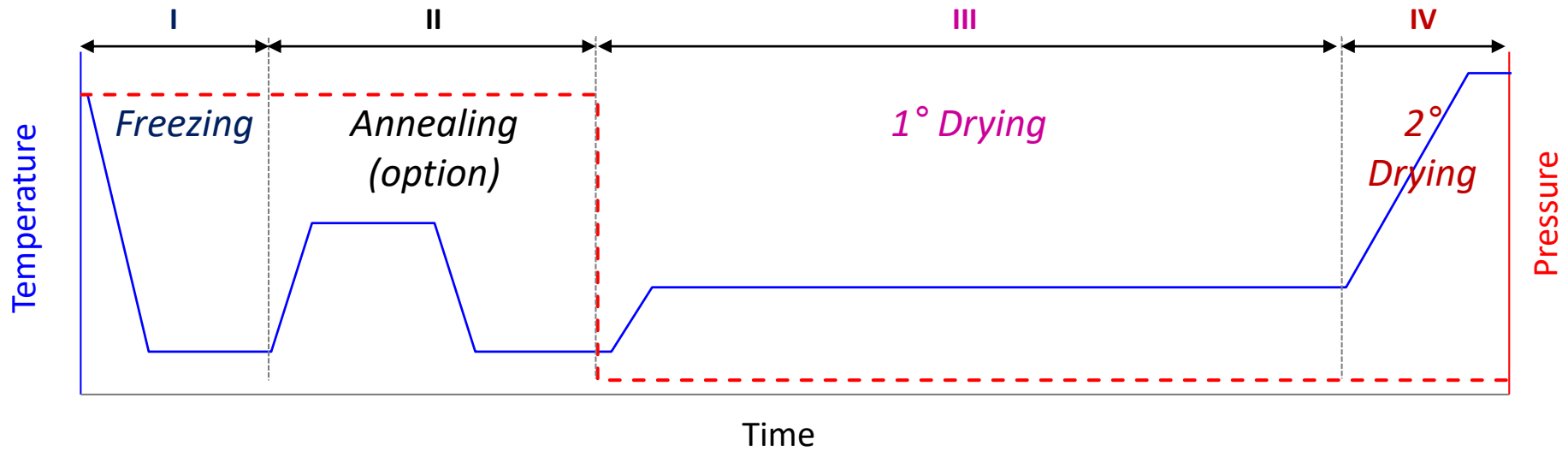
Global Sales of Top 10 Lyophilized Drug Products

The prevalence of stability challenges for complex APIs and biologics has resulted in more pharmaceutical and biotech manufacturers turning to lyophilization resulting in 13.5% annual growth in freeze-drying over the last five years.

Product Name	API	Indication	Owner	Estimated 2018 Product Sales
Herceptin IV	Trastuzumab	Cancer	Genentech	\$7.2B
Keytruda	Pembrolizumab	Cancer	Merck and Co.	\$7.2B
Remicade	Infliximab	Rheumatoid Arthritis, Crohn's Disease	Janssen Biotech	\$6.4B
Botox	Daxibotulinumtoxin A	Various	Allergan	\$3.6B
Carimune NF	Immunoglobulin	Immunodeficiency	CSL Behring	\$3.3B
Xolair	Omalizumab	Asthma	Genentech	\$3B
Orencia	Abatacept	Rheumatoid Arthritis	Bristol-Myers Squibb	\$2.9B
Cosentyx	Secukinumab	Plaque Psoriasis	Novartis AG	\$2.8B
Avonex	Interferon beta-1a	Relapsing MS	Biogen	\$2.4B
Velcade	Bortezomib	Cancer	Takeda	\$2.3B

<https://lubrizolcdmo.com/blog/lyophilization-of-pharmaceuticals-an-overview/>

Freeze-drying



Presentation Brief

1. Analytical techniques for medicines manufacturing
2. How the techniques used are being or can be applied in actual real-life processes
3. Where the measurement strategies replace/bring efficiencies in the control strategy for medicines above traditional testing strategies

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- Impedance spectroscopy
 - Z-FDM : Suitable for small scale formulation development



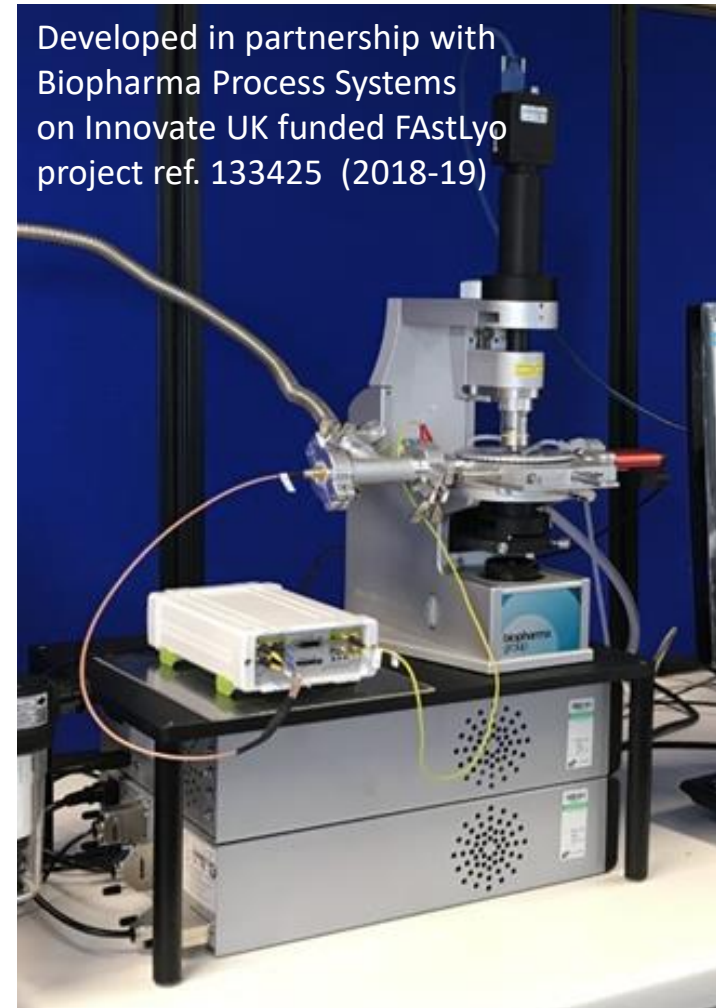
- TVIS : Scale-able to process development



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Application : objective assessment of thermal processes (crystallization, glass transition, collapse temperature) for **formulation Development**



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Application : In-vial assessment of multiple process parameters for **process development and scale up**

TVIS (Through-Vial Impedance Spectroscopy) was developed in partnership with GEA Pharma Systems on Innovate UK funded LyoDEA project (2010-13)



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Traditional testing – Formulation Dev.

- Relies on separate technologies, i.e.
 - Freeze-drying microscope for T_c
 - DSC for T_g and T_{eu}
- Disadvantages
 - Multiple techniques
 - FDM is subjective
 - One sample at a time
 - Not saleable to High-through-put

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Traditional testing – Process Dev.

- Indirect estimates of critical temperatures based on calculations that have assumptions.
- Many tools provide 'batch average' information
 - does not provide the resolution needed to understand the process variations across batch of vials.

Through Vial Impedance Spectroscopy

Single Vial PAT



Non- perturbing to packing of vials



Temperature calibration

- using nearest neighbour vial(s)

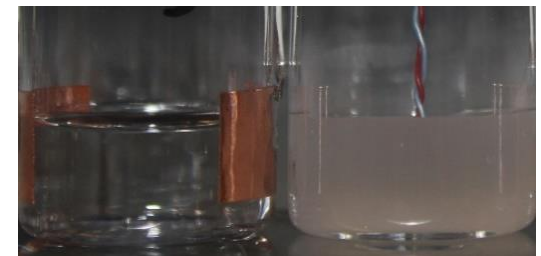
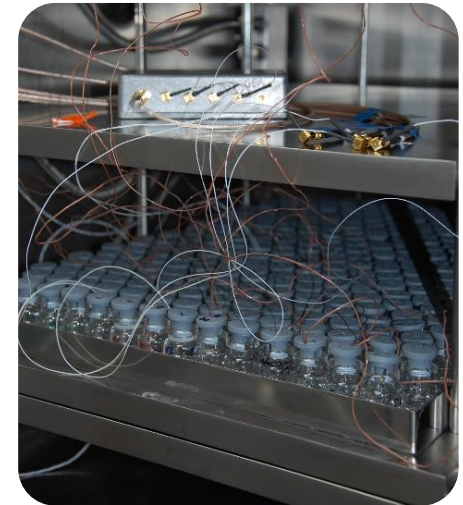


Low thermal mass of electrodes

- no interference with heat transfer & drying rates



Multichannel



Non-sample invasive

- no impact on ice nucleation

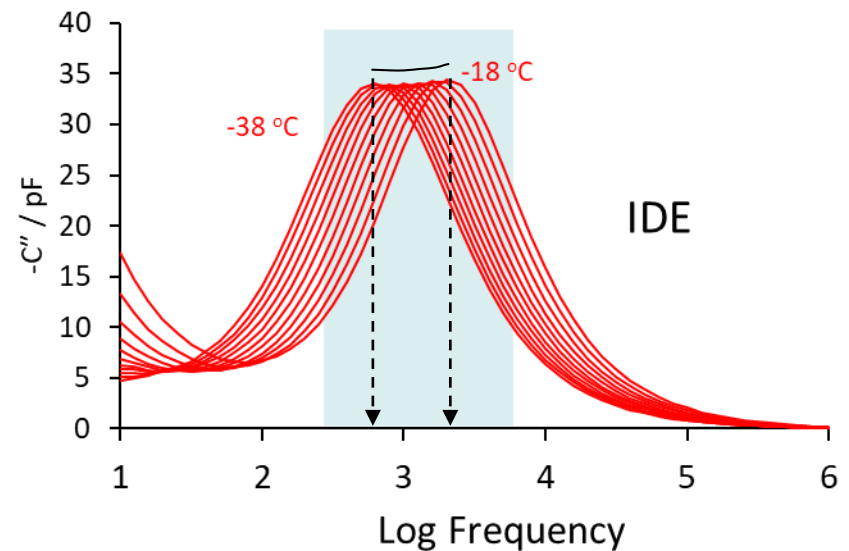
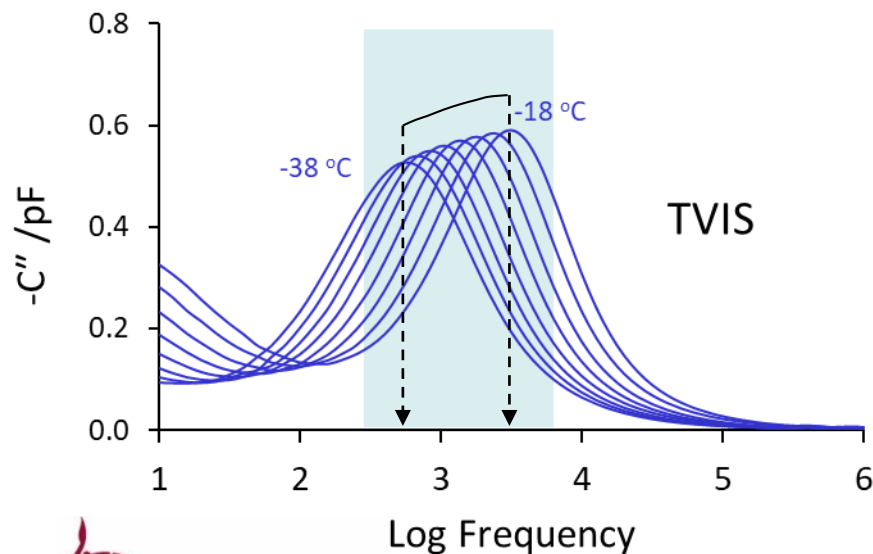
Dielectric relaxation of ice



5 mL water in
10 mL glass TVIS vial
(1 pair of 10/19 mm
height/width electrodes)



2 μL water over IDE
(90 pairs of gold interdigitated electrodes)



Freezing step

Nucleation
onset

Solidification
end point

Solidification
period

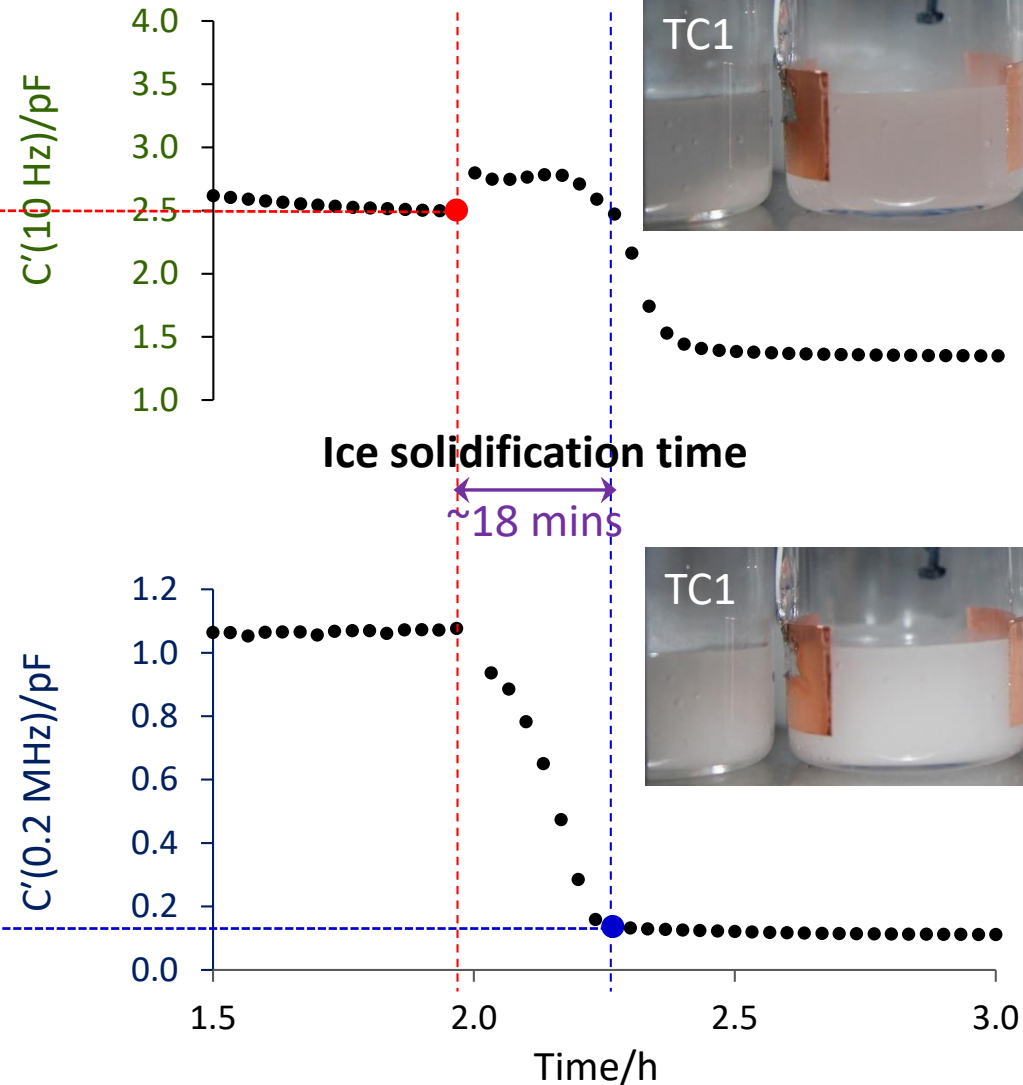
Nucleation
temperature

Solidification period

Ice nucleation

- The difference between these two times if the ice solidification time
- Knowing the height of the product in the vial one can then estimate an average solidification rate

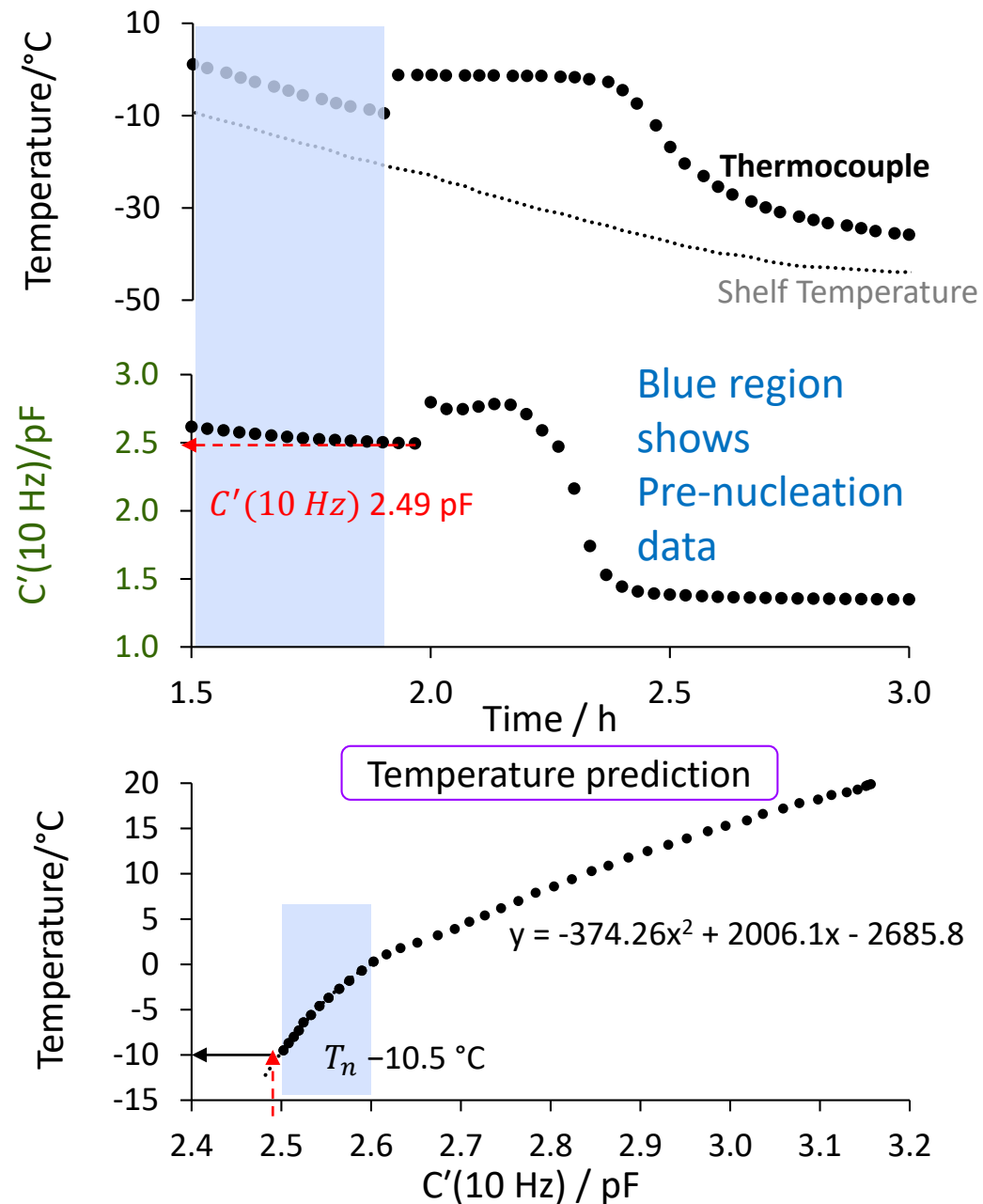
Solidification end point




$$\text{Average solidification rate } (R_{av}) = \frac{\text{Ice height } (L)}{\text{solidification time } \Delta t}$$

Nucleation Temperature

- In case the TVIS vial nucleates before TC vial, the nucleation temperature in the TVIS vial can be inferred directly from TC temperatures in the nearest neighbor vials
- However, if TVIS vial nucleates later than TC vial, the nucleation temperature can be predicted by fitting a curve to the plot of the average temperature from thermocouple vials against TVIS parameter (i.e. $C'(10\text{ Hz})$)
- The ice nucleation temperature of sample (5 %w/v sucrose) was found to be $-10.5\text{ }^{\circ}\text{C}$ in the case of this particular TVIS vial (other vials will differ owing to the stochastic nature of ice formation).



Annealing



Structural
modification

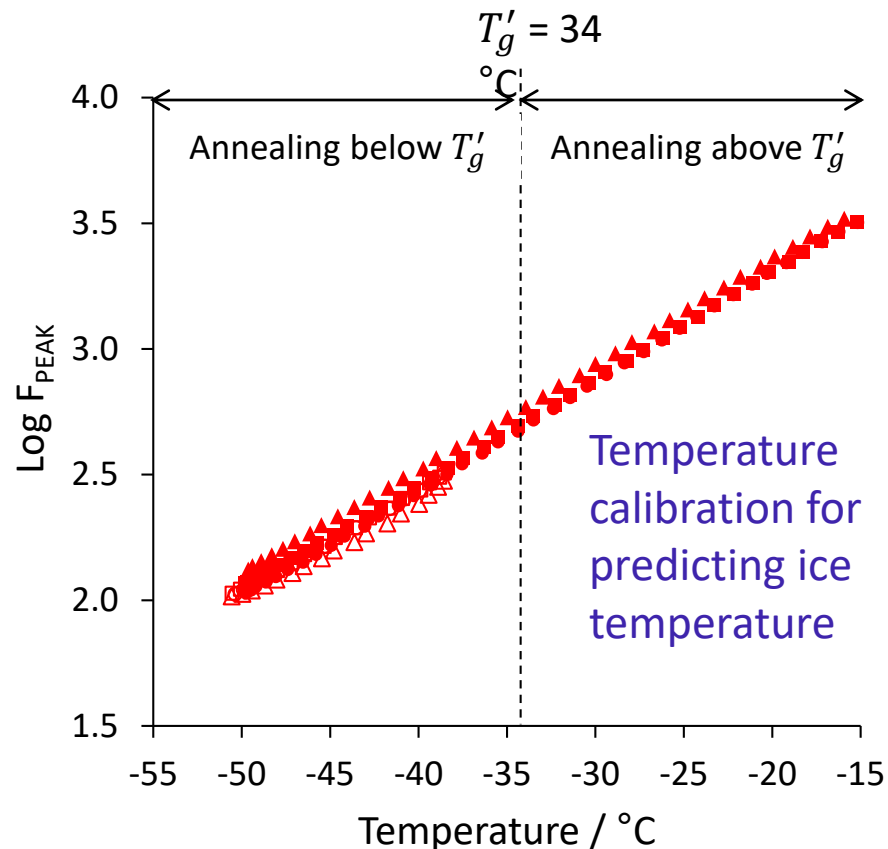
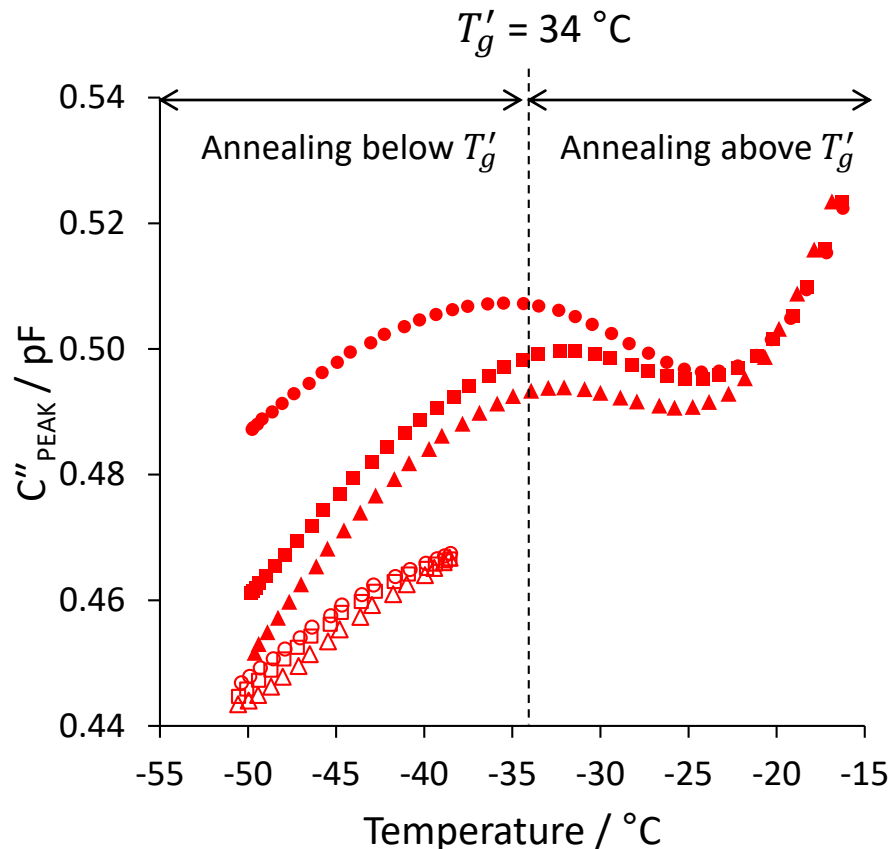
Glass transition

Crystallization

Completion of
annealing

Structural Modification

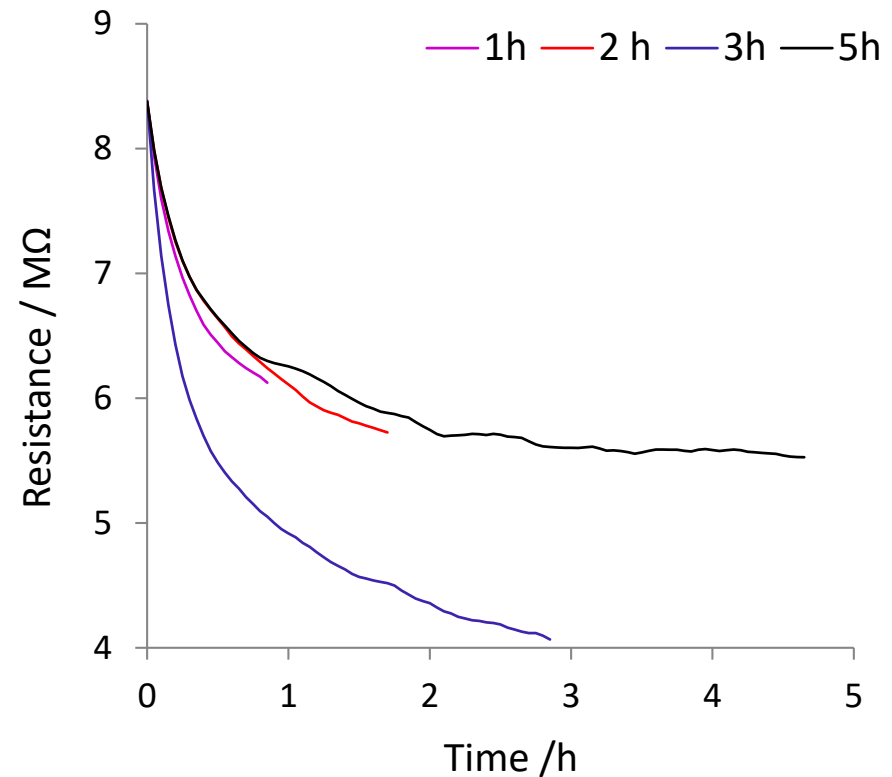
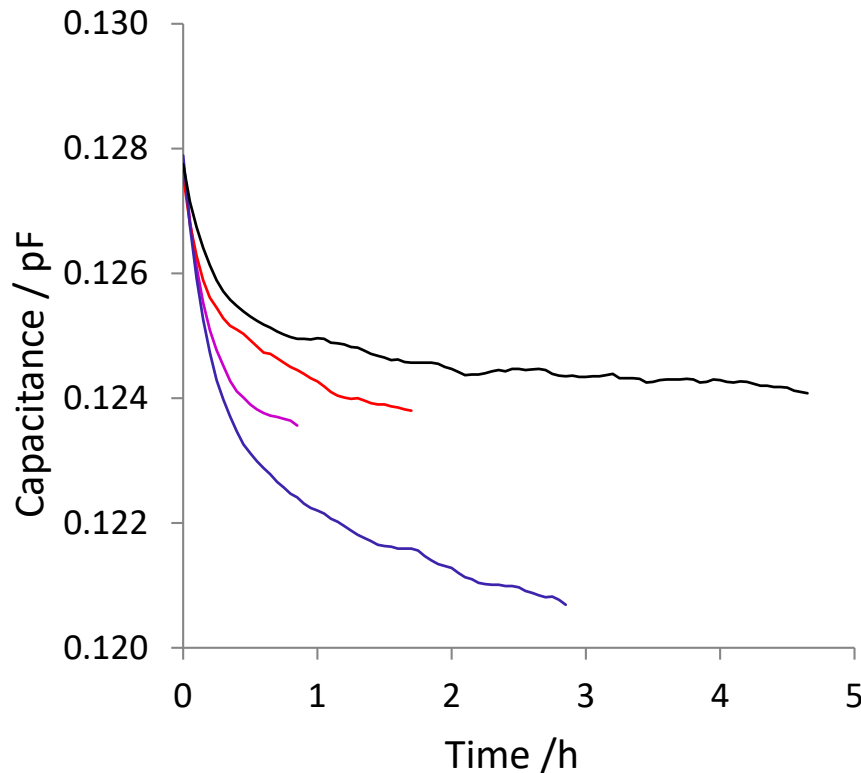
5% Sucrose solution



Closed symbols demonstrated the data when the sample was heated above its glass transition of freeze concentration ($T'_g = -34^{\circ}\text{C}$) whereas the product which annealed below transformation points of -35°C were represented by *open symbols*. Circle, square and triangle are 1st, 2nd and 3rd re-heating accordingly

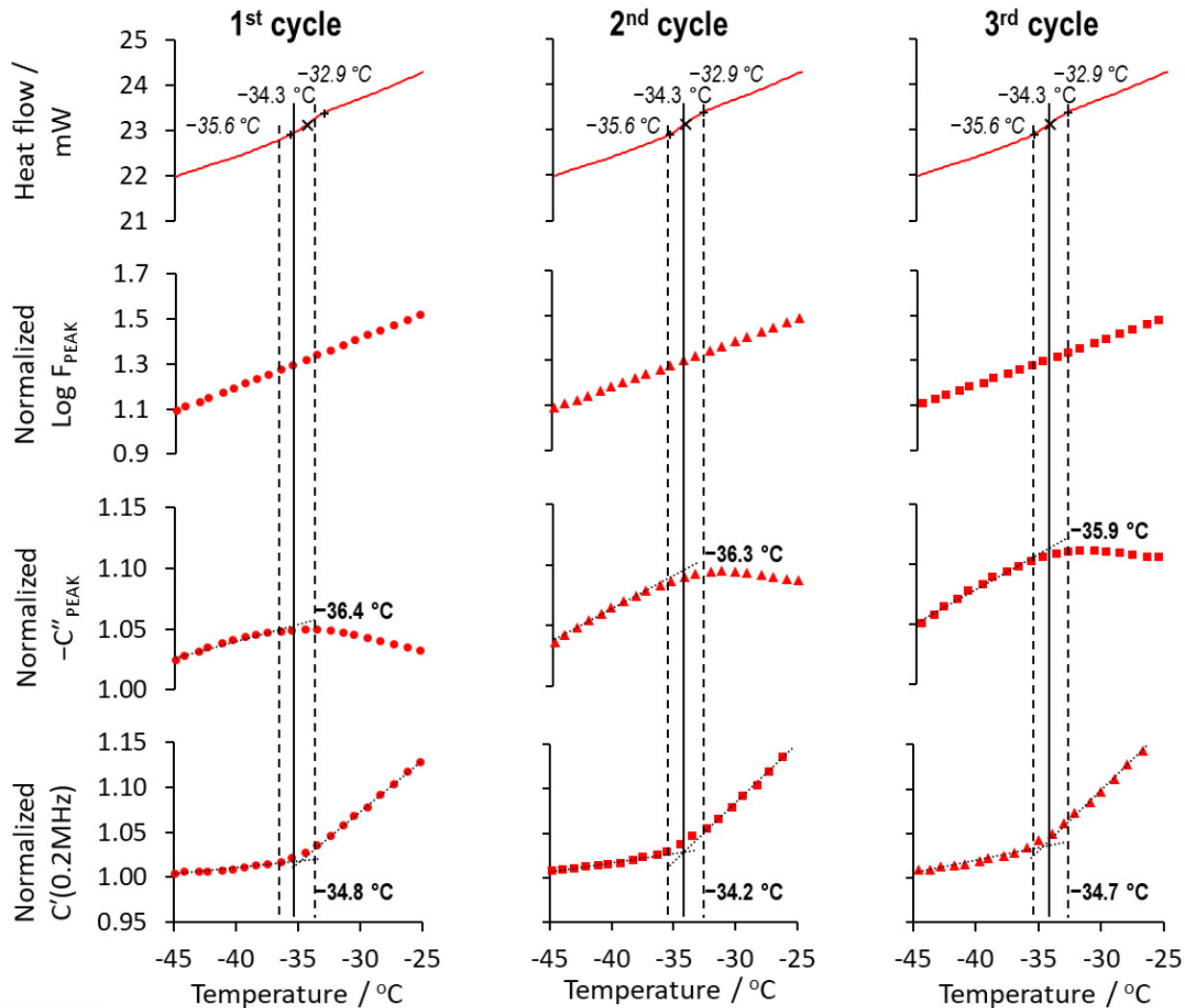
Investigations of annealing times

10% Maltodextrin




- The capacitance of the formulation changes minimally while the resistance changes significantly and plateaus at 3-4 h
- After 3h annealing hold time, both the capacitance and drying time changes insignificantly

Glass transition 5% Sucrose solution



Freezing and annealing of mannitol-sucrose



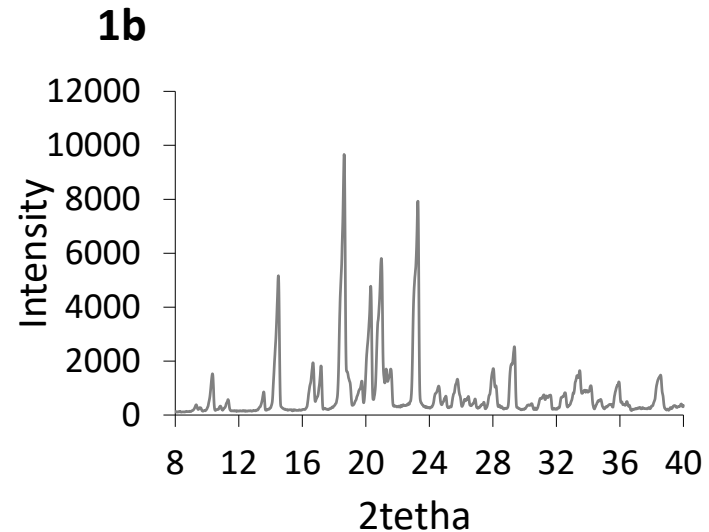
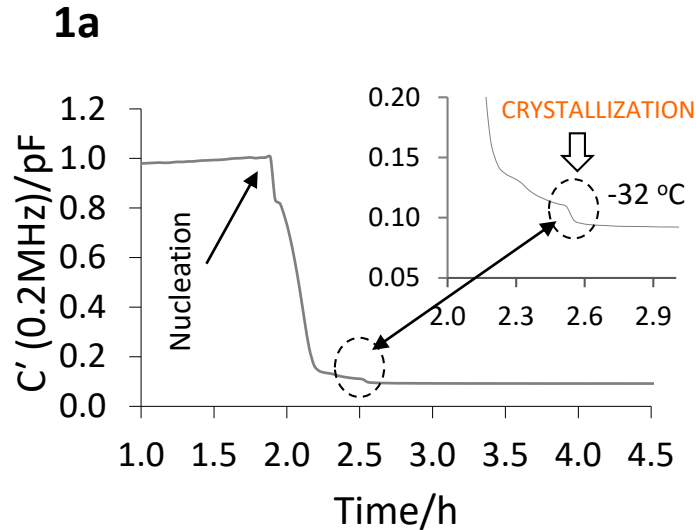
Freezing man-suc
solution

Crystallization in
man-suc formulations

Re-crystallization
events

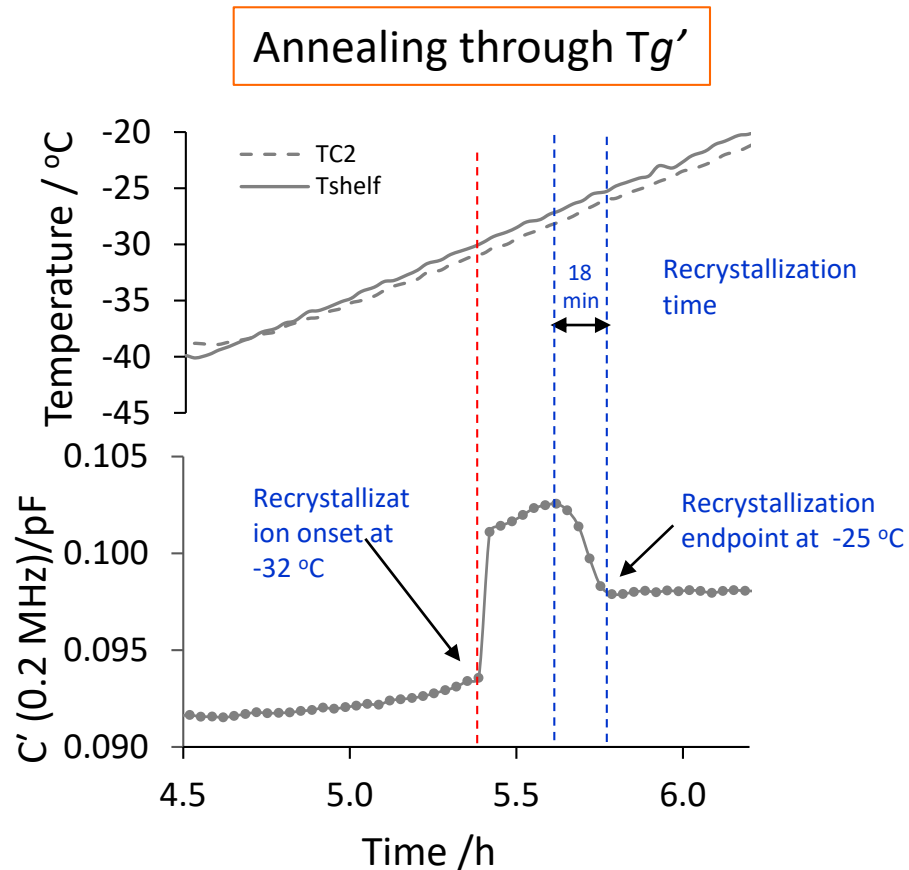
Change in crystal
structures

Crystallization events at freezing step for 5% mannitol



TVIS and XRD results for solution of 5% mannitol showing crystallization detected by TVIS

Re-crystallization on annealing step for 5% mannitol



On annealing to -20 °C
 C' (0.2 MHz) shows
mannitol recrystallization with
onset at -32 °C
end point at -25 °C

Primary drying

Product
temperature

Sublimation
rate

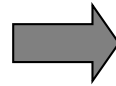
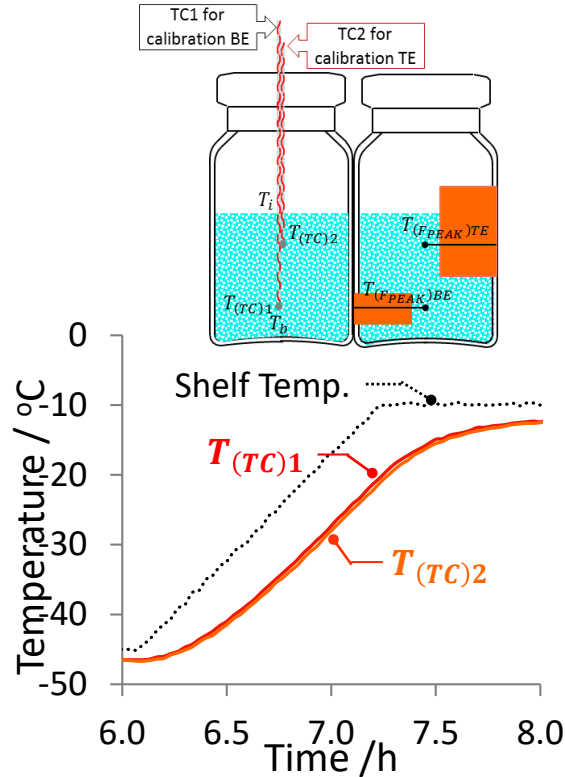
K_v

Primary
drying
endpoint

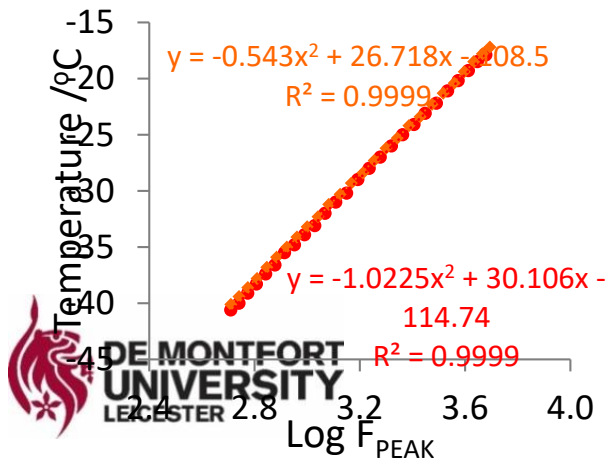
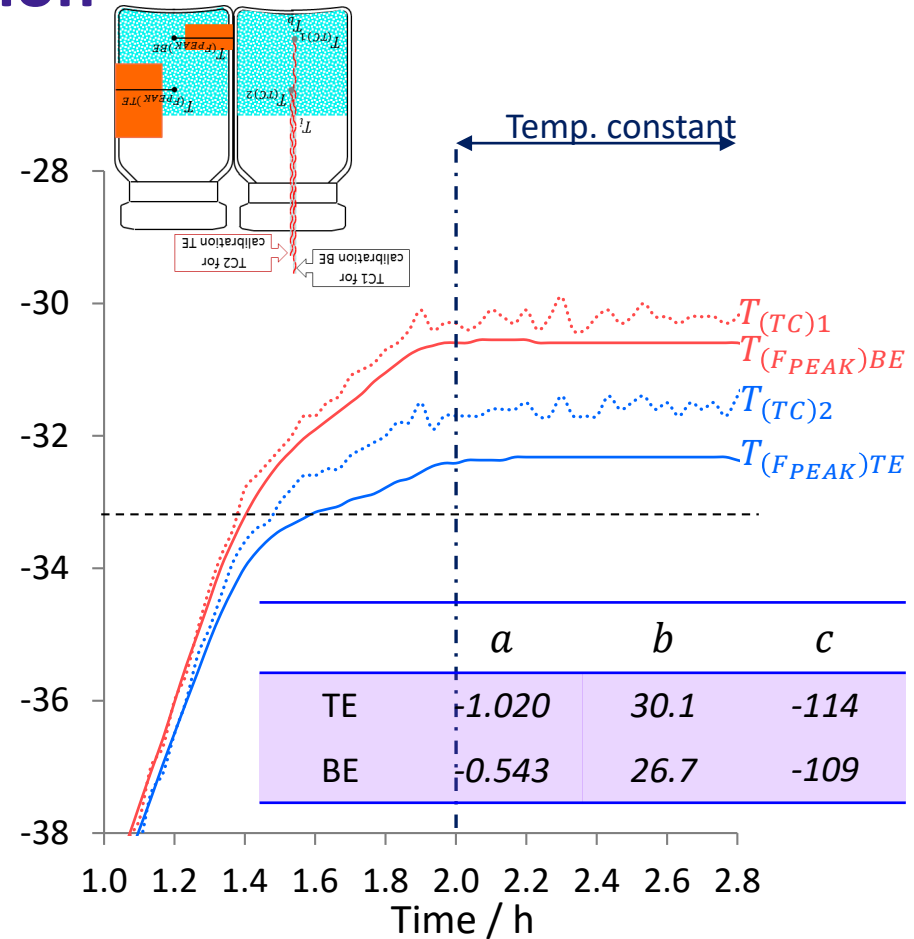
R_p

Collapse
event

Product temperature prediction



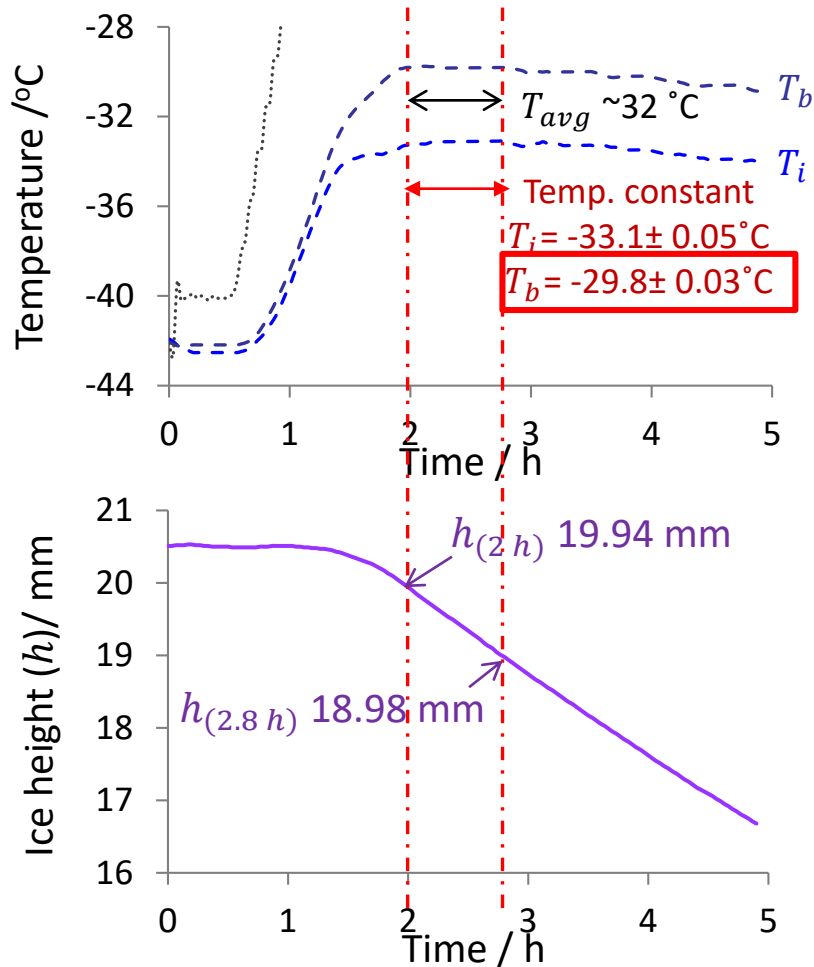
Temperature / °C



The product temperature predicted by TVIS can demonstrate the temperature gradient across ice cylinder height

BACK to Drying

Drying Rate Estimation Pure ice



- Drying rate during the steady state

$$\text{Drying rate } \left(\frac{\Delta m}{\Delta t} \right) = \rho_i \cdot A \cdot \frac{h_{(t_1)} - h_{(t_2)}}{t_2 - t_1}$$

Ice density (ρ_i) at -32°C = $0.920\text{ g}\cdot\text{cm}^{-3}$

(Calculated ice temperature between T_i & T_b)

Internal vial diameter (VC010-20C) = 2.21 cm

Cross-section area (A) = 3.80 cm^2

Ice height at 2 h ($h_{(2\text{ h})}$) = 19.94 mm

Ice height at 2.8 h ($h_{(2.8\text{ h})}$) = 18.98 mm

TVIS parameters used for determination:

$$\frac{\Delta m}{\Delta t} = 0.42\text{ g}\cdot\text{h}^{-1}$$

$$T_b = -29.8^\circ\text{C}$$

$$\begin{aligned} \text{Drying rate} &= 0.920\text{ g}\cdot\text{cm}^{-3} \times 3.80\text{ cm}^2 \times \frac{(19.94 - 18.98) \times 10^{-1}\text{ cm}}{(2.8 - 2.0)\text{ h}} \\ &= \mathbf{0.42\text{ g}\cdot\text{h}^{-1}} \end{aligned}$$

Heat Transfer Coefficient (K_v) Determination

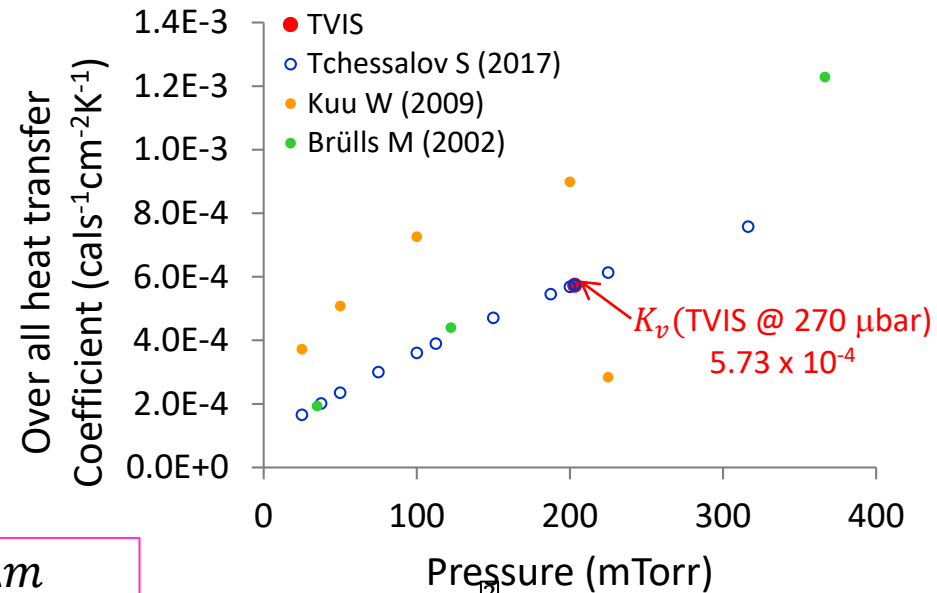
Pure ice

Parameters	TVIS
Drying rate at steady state (g/h) (2-2.8 h into primary drying)	0.42
Shelf Temperature, T_s (K)	273.3
Vial's base Temperature, T_b (K)	243.3

$$L \frac{\Delta m}{\Delta t} = A_e K_v (T_s - T_b) \Rightarrow$$

$$K_v = \frac{L \frac{\Delta m}{\Delta t}}{A_e (T_s - T_b)}$$

L is the latent heat of sublimation of ice (2844 J·g⁻¹ or 679.7 cal·g⁻¹) and A_e is external cross-sectional area of the base of the TVIS vial (4.62 cm²)



$$K_v(270 \text{ bar}) = \frac{L \frac{\Delta m}{\Delta t}}{A_e (T_s - T_b)}$$

$$= \frac{679.7 \text{ cal} \cdot \text{g}^{-1} \times 0.42 \text{ g} \cdot \text{h}^{-1}}{4.62 \text{ cm}^2 \times (273.3 - 243.3) \text{ K}}$$

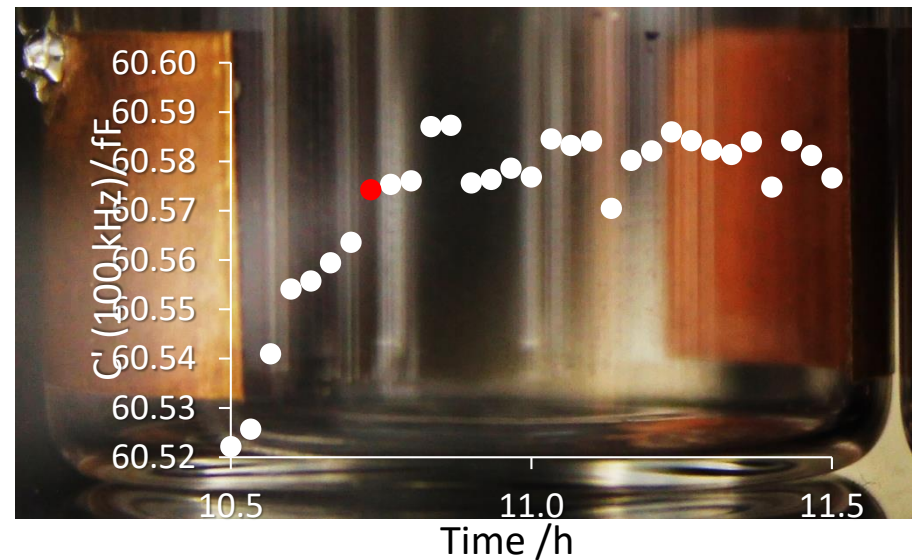
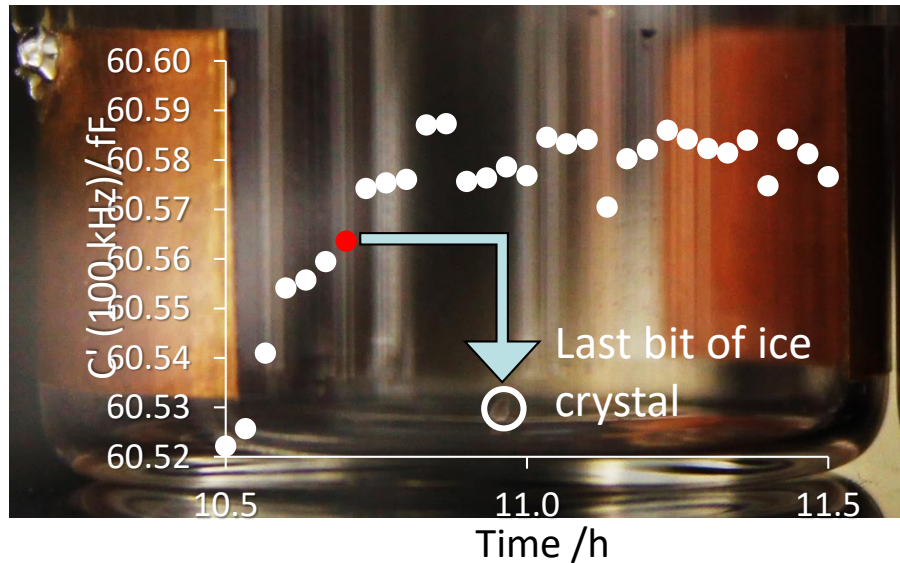
$$= 2.06 \text{ cal} \cdot \text{h}^{-1} \cdot \text{cm}^{-2} \cdot \text{K}^{-1}$$

$$= 5.73 \times 10^{-4} \text{ cal} \cdot \text{s}^{-1} \cdot \text{cm}^{-2} \cdot \text{K}^{-1}$$

$$K_v(270 \mu\text{bar}) = 5.73 \times 10^{-4} \text{ cal} \cdot \text{s}^{-1} \cdot \text{cm}^{-2} \cdot \text{K}^{-1}$$

Primary Drying Endpoint

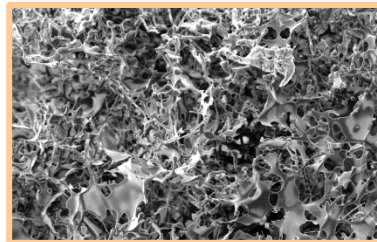
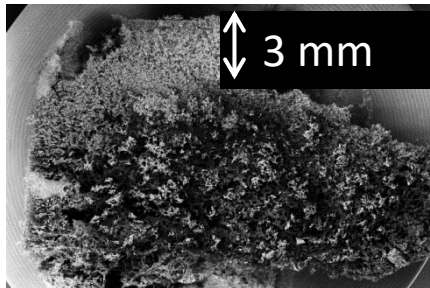
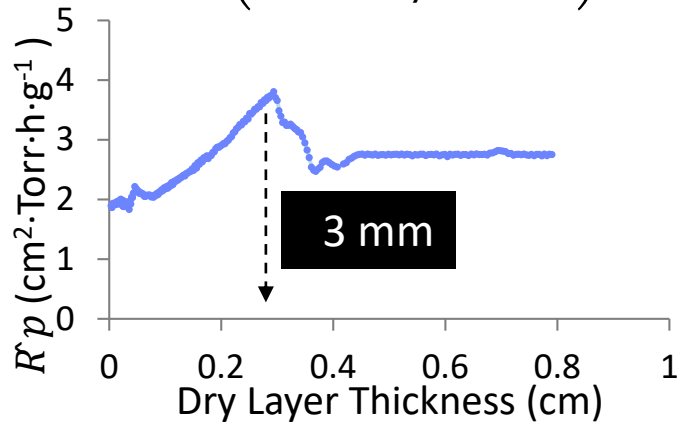
Pure ice



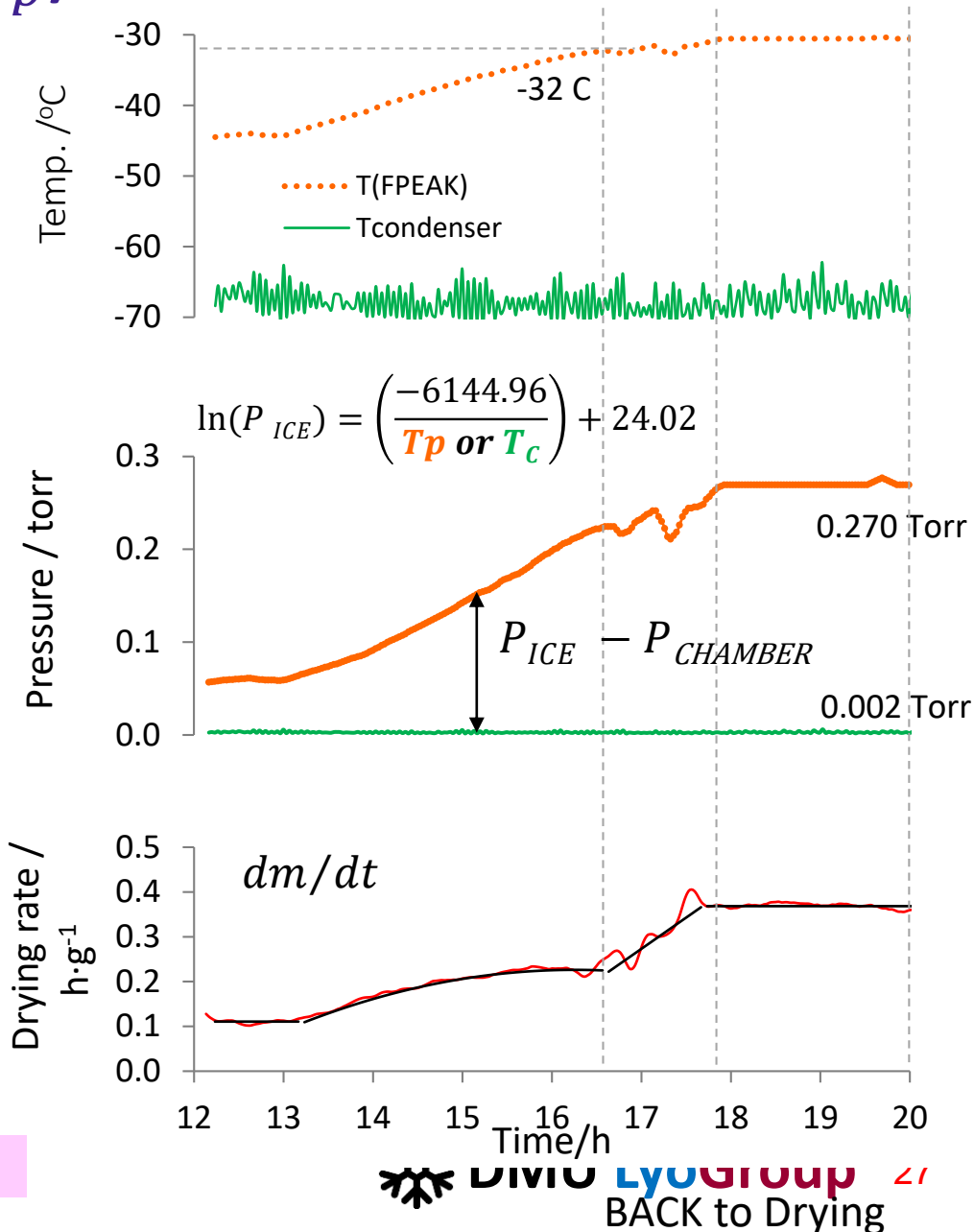
- End point determine by high frequency real part capacitance (i.e. 100 kHz for ice)
- Endpoint where $C' (100 \text{ kHz})$ reaches plateau

Dried product resistance (R_p) 5% Lactose

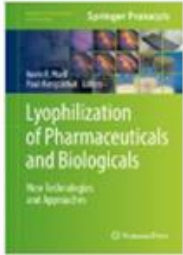
$$\hat{R}p = \left(\frac{P_{ICE} - P_{CHAMBER}}{dm/dt} \right) \cdot A_p$$



Middle layer micro-collapse



Further Reading



[Lyophilization of Pharmaceuticals and Biologicals](#) pp 241-290 | [Cite as](#)

Through Vial Impedance Spectroscopy (TVIS): A Novel Approach to Process Understanding for Freeze-Drying Cycle Development

Authors

[Authors and affiliations](#)

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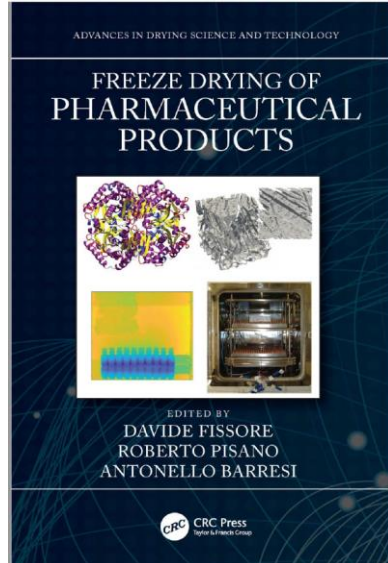
- Introduction to TVIS theory
- Description of the measurement principles
- Dielectric loss and relaxations mechanisms (liquid and frozen states)

Further Reading

Chapter 5 Through Vial Impedance Spectroscopy (TVIS) A New Method for Determining the Ice Nucleation Temperature and the Solidification End point

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£118.00

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