

 **qepler**
PHARMACEUTICAL
LYOPHILIZATION
SUMMIT
2021

An introduction to TVIS PAT

Through-Vial Impedance Spectroscopy

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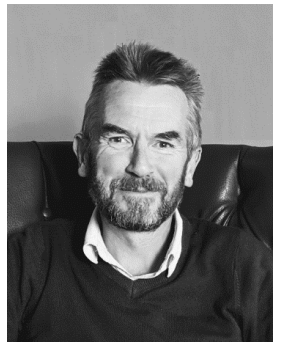


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July 29 | 1st DAY

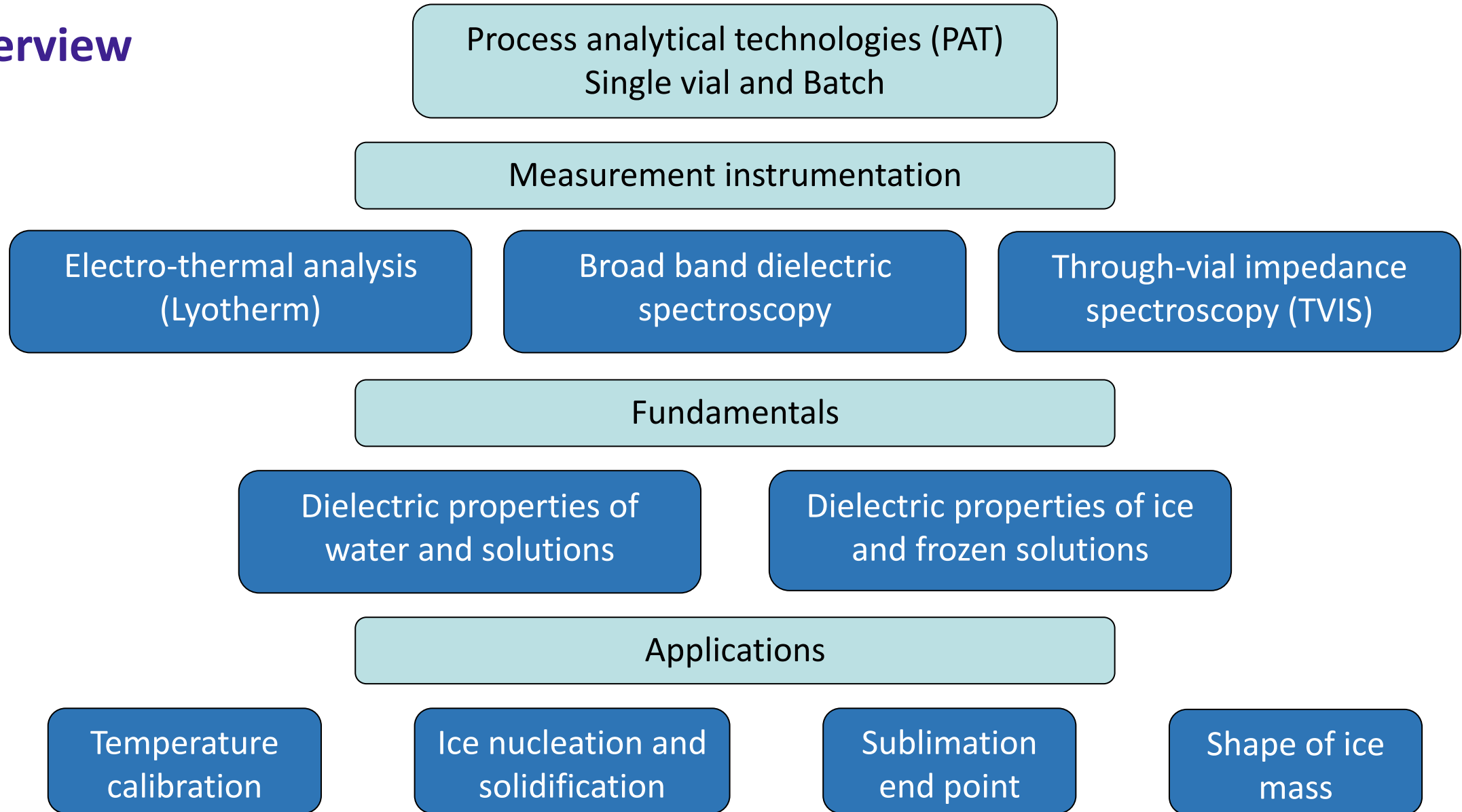
Central European Summer Time (CEST, Prague, UTC/GMT +2 hours)

In collaboration with

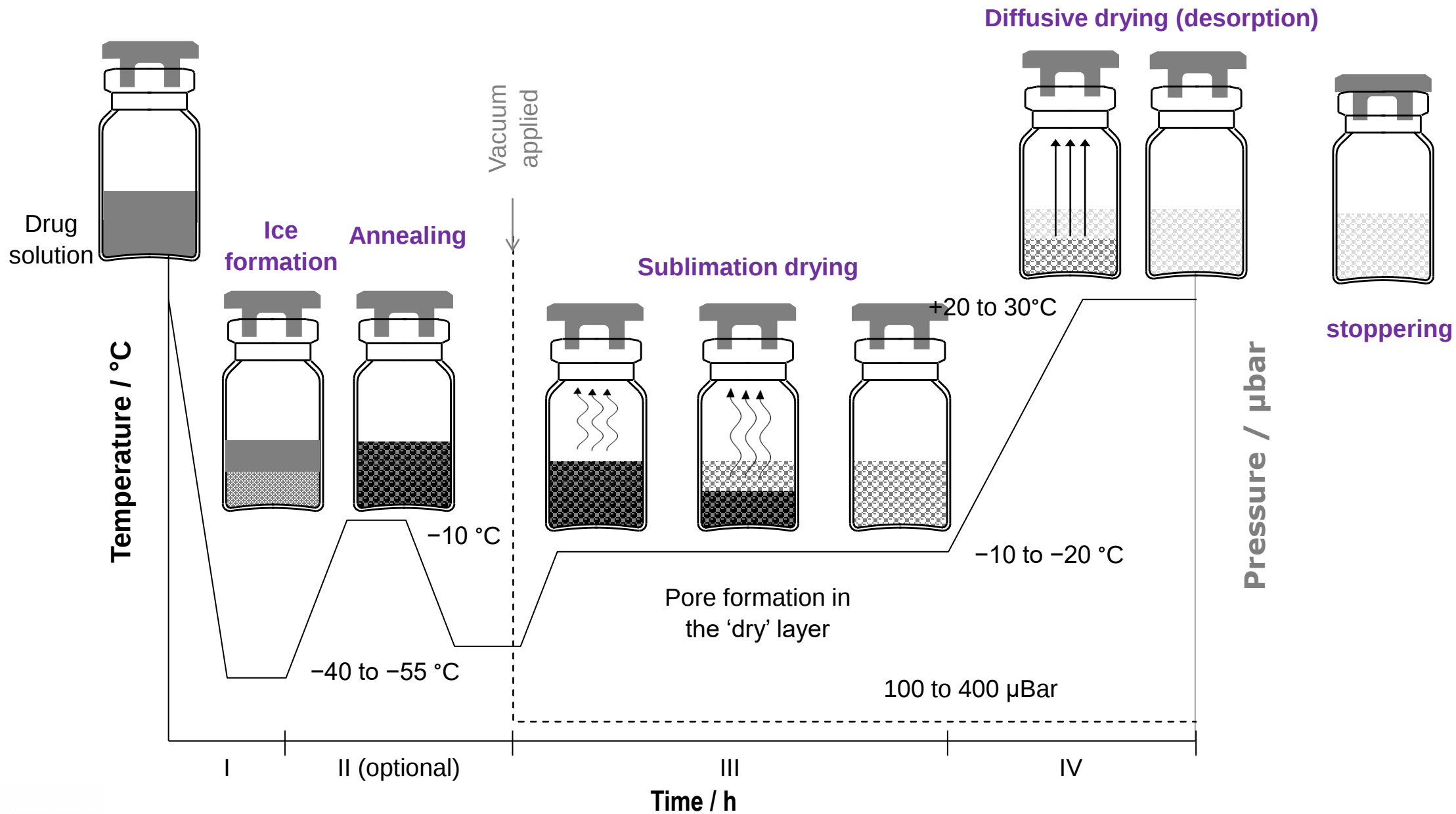


DMU LyoGroup

Overview



Lyophilization Process





- (A) shelf temperature (ramp)
- (B) chamber pressure
- (C) condenser temperature

- Ice nucleation temperature (T_n)
- Rate of change of the product temperature (dT_p/dt):
- Phase behaviour of solid/solute fraction
- Critical temperature (e.g. collapse), T_c
- Vial heat transfer coefficient
- Porosity of the 'dry' fraction of the product that develops during primary drying ($R_p = 1/\text{Porosity}$)
- Ice interface temperature (T_i) $< T_c$
- Primary and secondary drying end points

Process analytical technologies (PAT) for freeze-drying

Process Analytical Technology (PAT)

PAT, as defined by the ICH, is “a system for **designing, analysing and controlling** the manufacturing through timely measurement (during the process) of critical quality and performance attributes of raw and in-process materials and the process with the goal of ensuring final product quality”

ICH, 2009. International Conference on Harmonisation of Technical Requirements for Registration of Pharmaceuticals for Human Use
Topic Q8(R2): Pharmaceutical Development.

Single vial techniques

- Temperature probes
 - TC, RTD
 - Wireless : Tempris[®]; TrackSense[®]
- Microbalance (CHRIST GmbH)
- Heat flux transducers (MicroFD, Millrock)

Batch techniques

- Pressure rise test (PRT)
 - Comparative pressure measurement (CPM)
- } Primary drying end point
-
- Manometric temperature measurement (MTM)
 - Time domain laser absorption spectroscopy (TDLAS)
- } Predictions of product temperature, drying rate



Louis Rey 1960s

STUDY OF THE FREEZING AND DRYING OF TISSUES AT VERY LOW TEMPERATURES

L. R. REY

Laboratoire de Zoologie-Physiologie, Ecole Normale Supérieure, Paris

THE outstanding interest of the freeze-drying technique is that it provides a possibility for long-term preservation of highly alterable structures. The method has had a great number of applications in the field of histology and histochemistry since the pioneer work of Altmann (1890) and Gersh (1932).

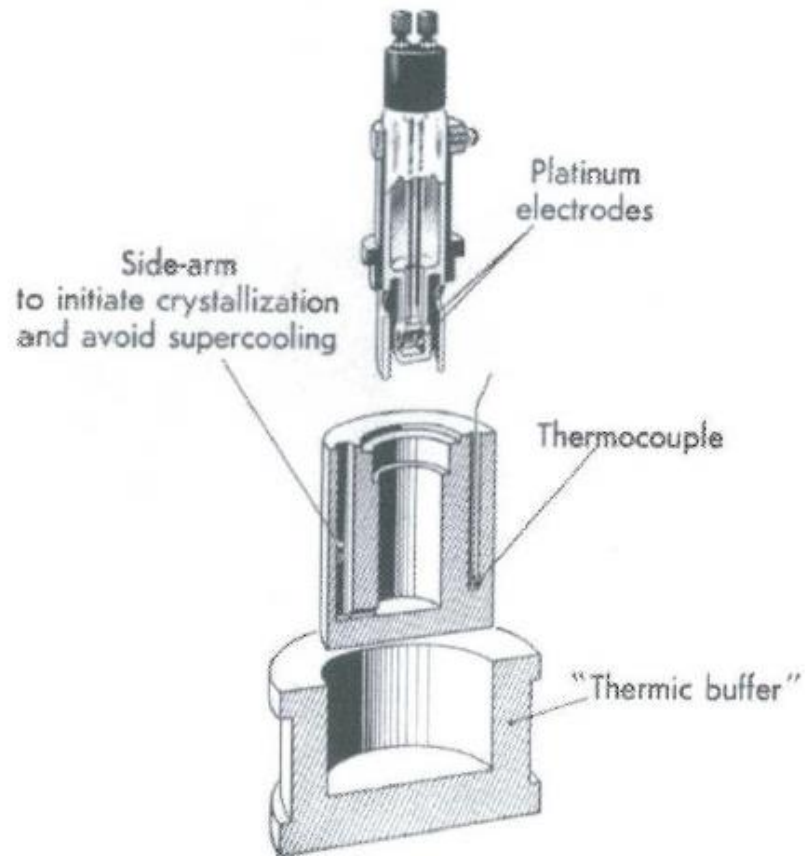
Many authors have thought, too, that it would be possible to try to preserve life in frozen and dried animal tissues. It has been shown long ago, by Becquerel (1936) that, as far as the cells of certain plants and lower animals are concerned, highly dehydrated material could be kept over very long periods of time and would resist extreme conditions of temperature.

Unfortunately, animal cells are not very resistant and they are not able to lose a great amount of water. Luyet and Hartung (1941), Rostand (1946), Parkes (1951) and Polge, Smith and Parkes (1949), have shown that different chemical substances and especially glycerol offer very good protection against cold injury. We have done similar experiments in our laboratory, and we have studied the protective action of glycerol by the use of tissue-culture and organ-culture techniques (Rey, 1957, b and d).

In a second group of researches, our intention was to find out the conditions of lyophilization which would be able to give the best preservation of the fine cytological structures of the tissue-cells, and which, if possible, would preserve life.

At first, and according to what we said in 1957 at the Royal Society, the freezing must be done in a very precise way, because there is no use drying with great care a tissue which has undergone denaturation in the course of freezing. It is likely that a glycerol preimpregnation would be necessary. However, we have had much evidence in our work (1957 d) and we agree perfectly well with Meryman (1957) that the most harmful period of the action of low temperatures on living tissues is during thawing. One could think then, that, if we avoid thawing by previous drying, it would be possible not to use glycerol impregnation, provided that the freezing period is short. This is a theoretical point of view, compromise solutions are not excluded and we can reasonably

40



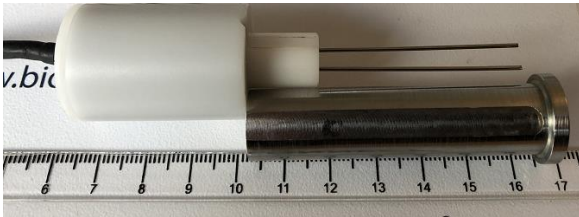
Rey (1960) Study of freezing and drying of tissues at very low temperatures. In Parkes and Audrey eds. Recent research in freezing and drying, Blackwell, Oxford.

Lyotherm

Biopharma Process Systems

Impedance analysis ($Z\sin\phi$) at a single frequency (1 kHz) with differential thermal analysis (DTA)

- Pin electrode (pair)

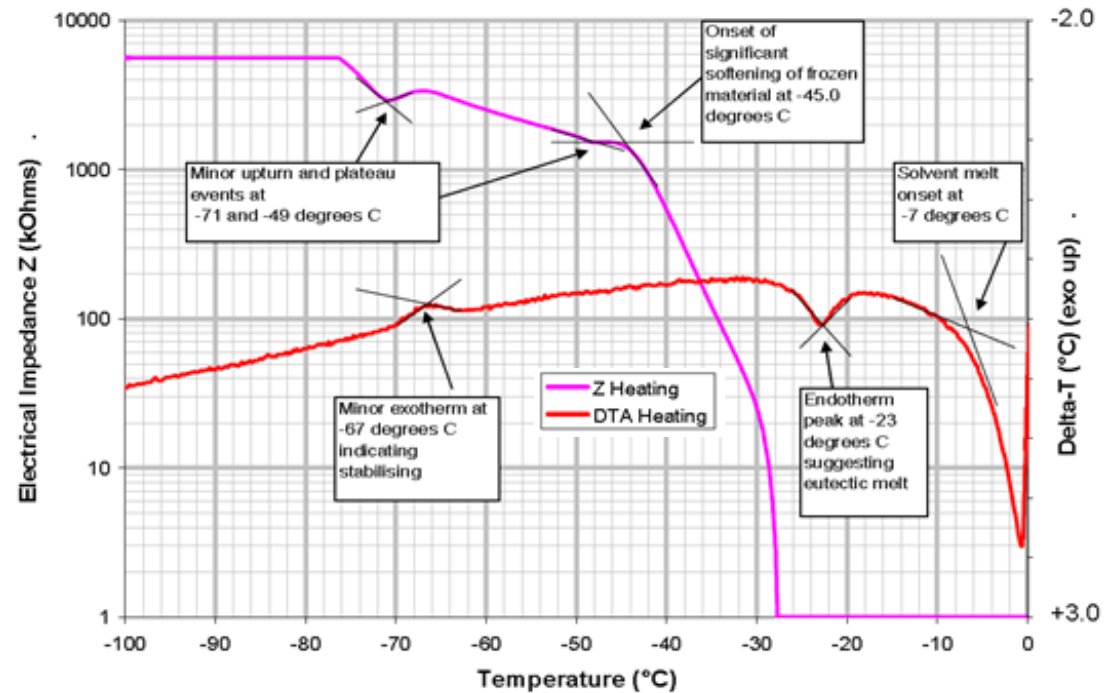


- Integrated within cryostat



Designed to measure temperatures relevant to freeze-dried formulations

- glass transition (T_G')
- ice melting (T_M) and eutectic (T_{EU})



Ward & Matejtschuk, 2010 in *Freeze Drying/ Lyophilization of Pharmaceutical & Biological Products* 3rd ed. Rey, L & May JC eds, Informa Press, New York

Measurement within glass vials

Freeze-thawing and freeze-drying

Spectroscopy Systems (sub Hz to 10 MHz)

Alpha-series
dielectric
spectrometer

Temperature
controller



Novocontrol
BDS40

cryostat

ZGS active sample Cell



2-electrode sample cell

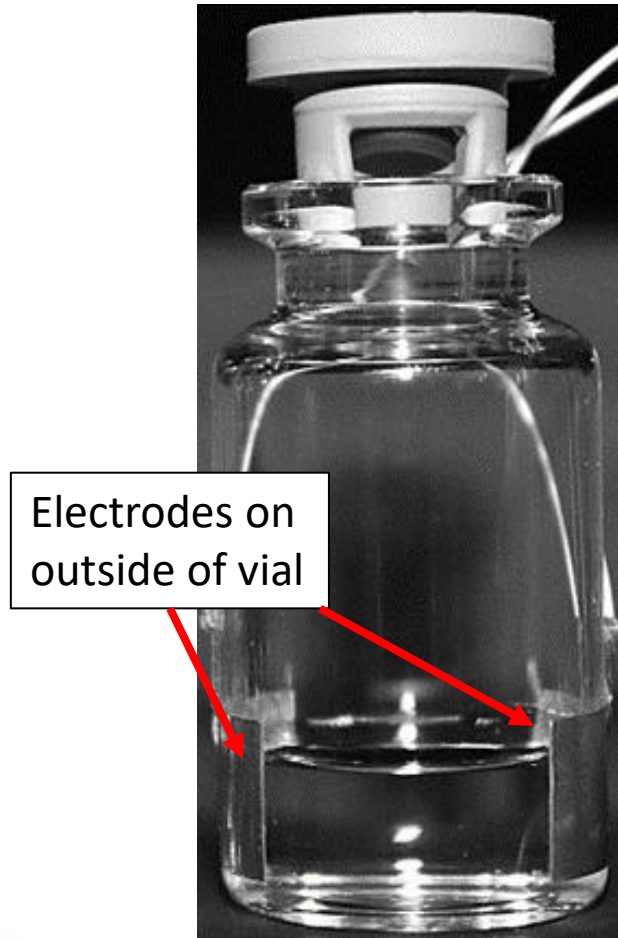


TVIS vial on new cradle
to be placed in the cryostat
of Novocontrol BDS

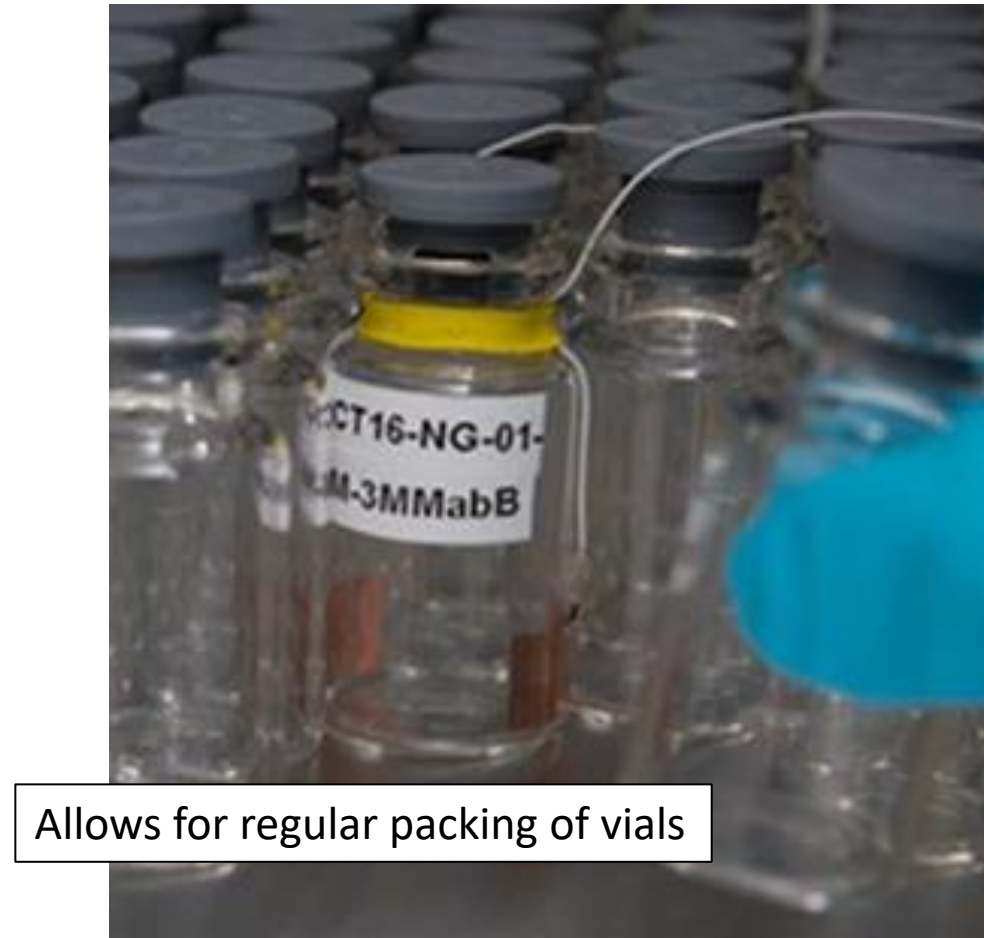


Non-invasive measurements: Lyosense LyoDEA TVIS

Individual vials

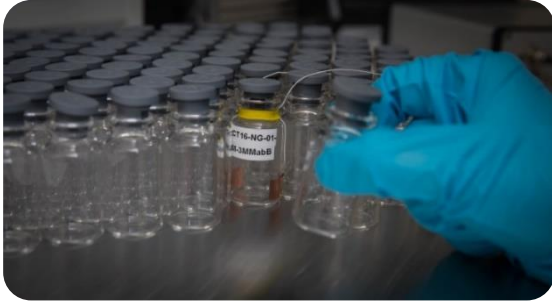


In process



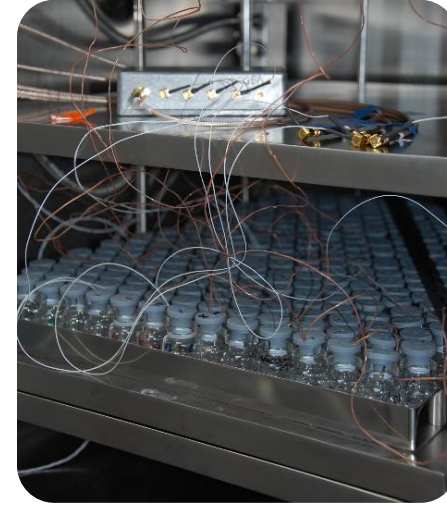
Through Vial Impedance Spectroscopy

Non- perturbing to packing of vials



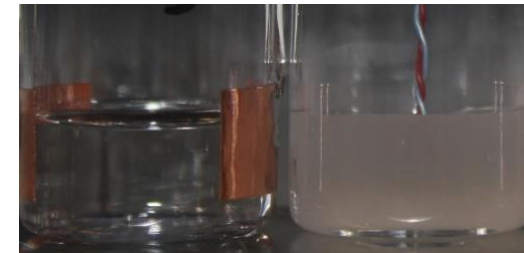
Thin flexible cables
(0.533 mm)

- Stoppering unaffected



Low thermal mass of
electrodes

- no interference with heat transfer & drying rates

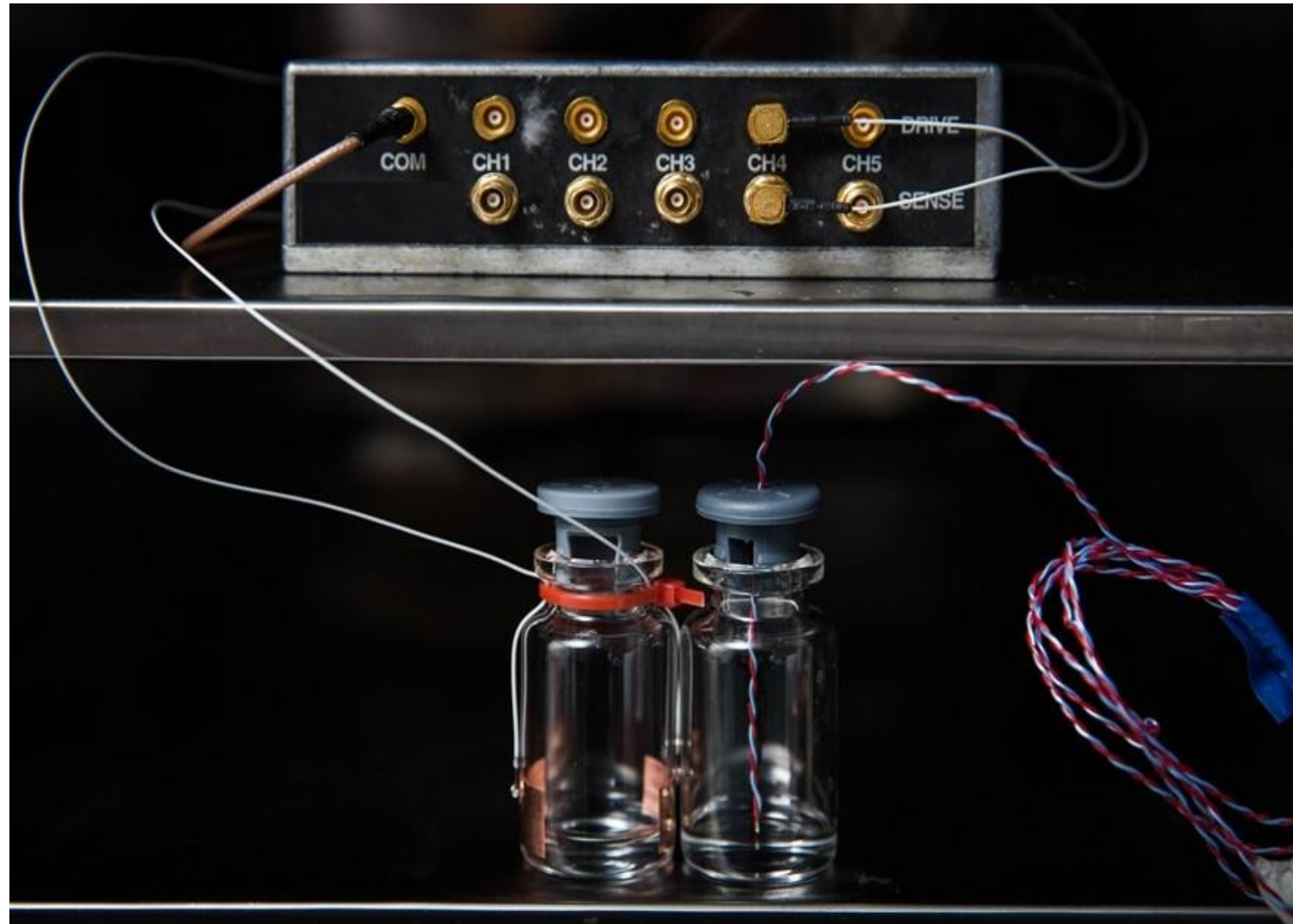


Non-sample invasive

- no impact on ice nucleation

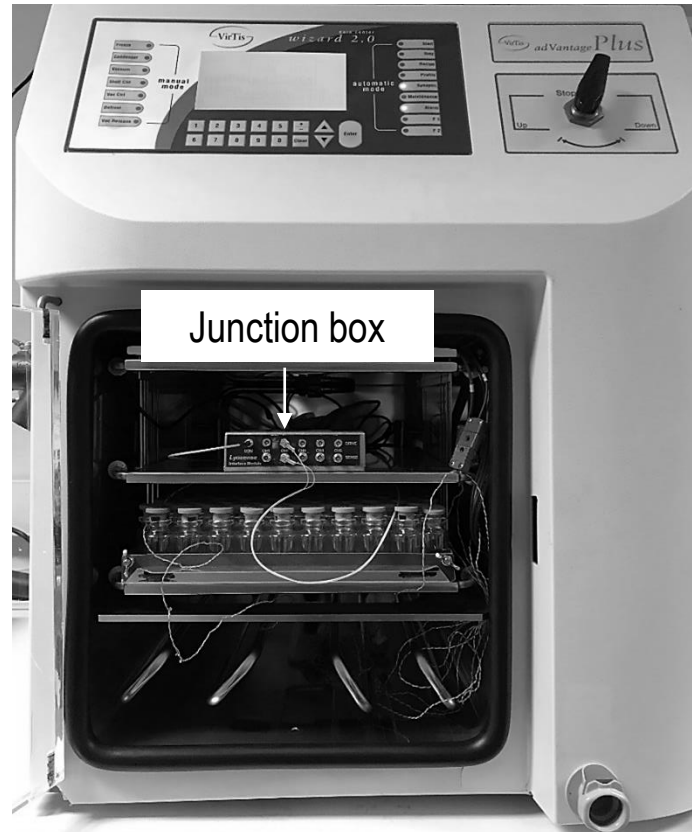


Junction box

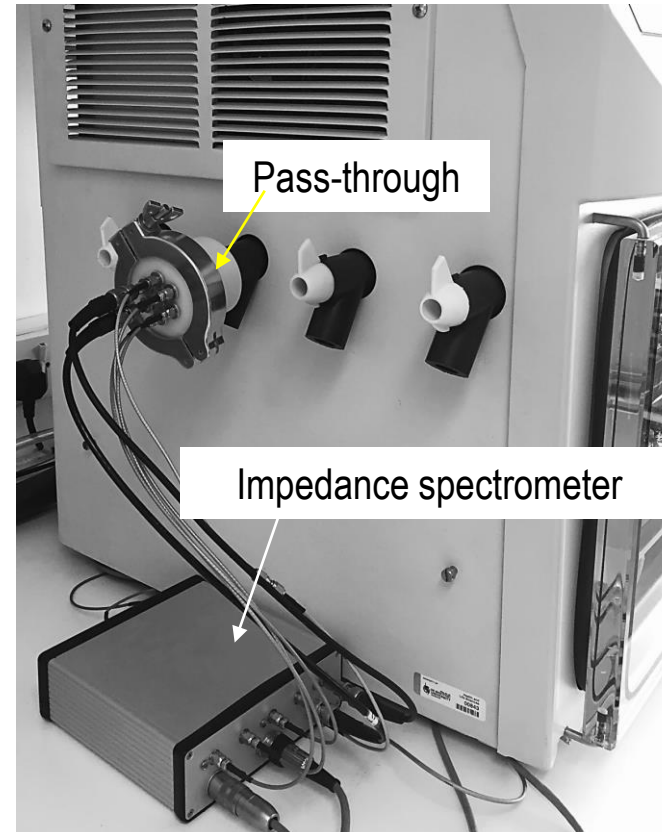


System configuration

a

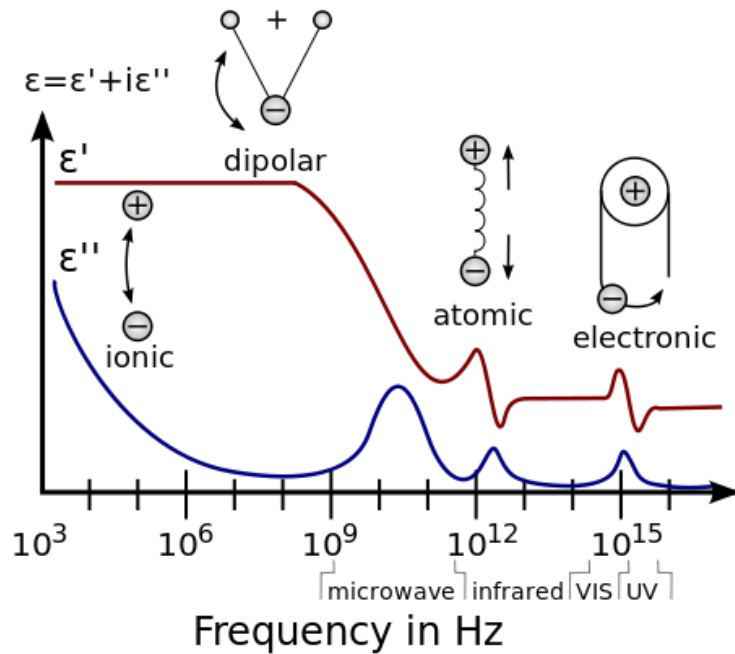


b

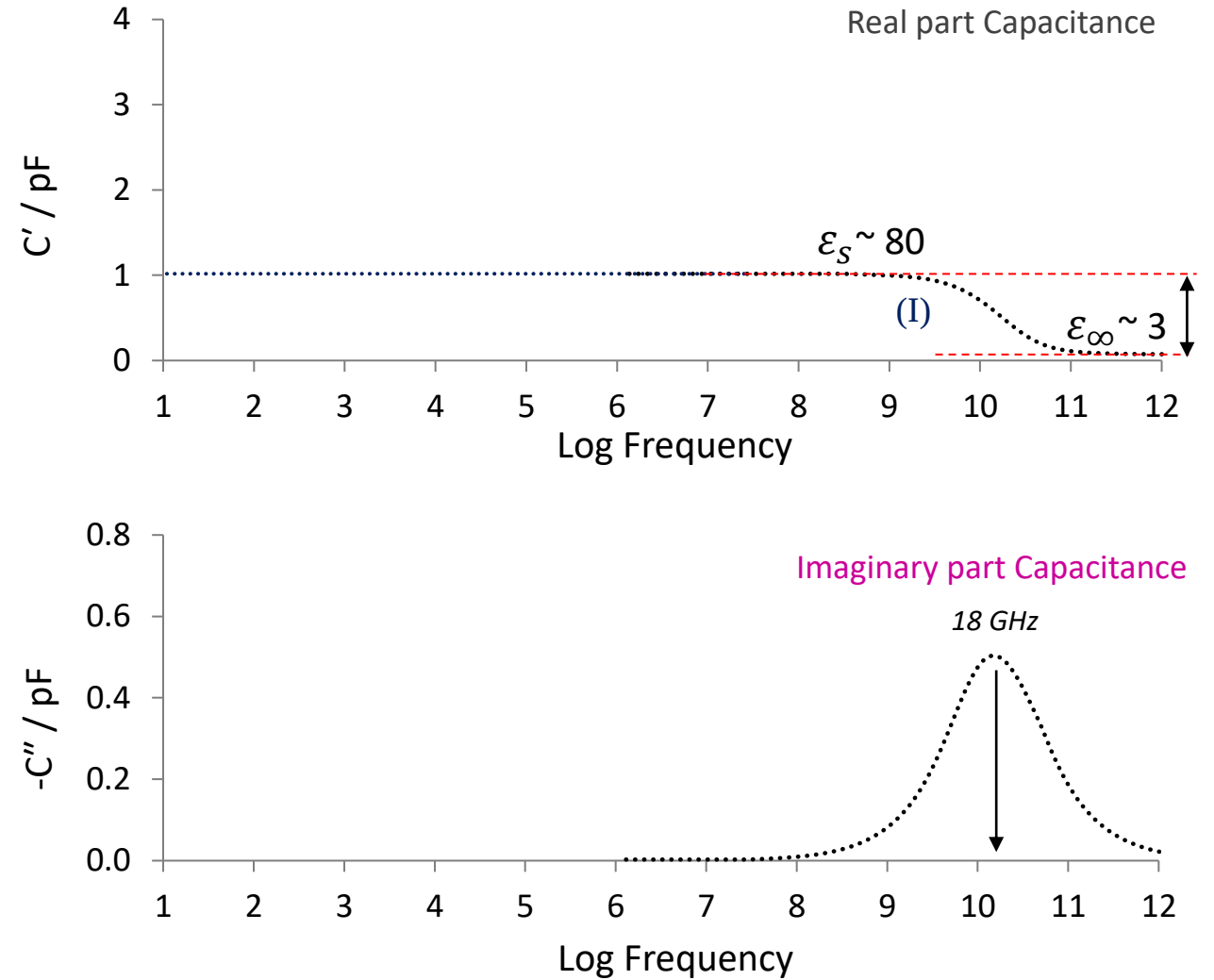


Dielectric Loss Mechanisms :

- I. The polarization of the water dipole in liquid water at 20 °C, with a dielectric loss peak frequency of ~ 18 GHz

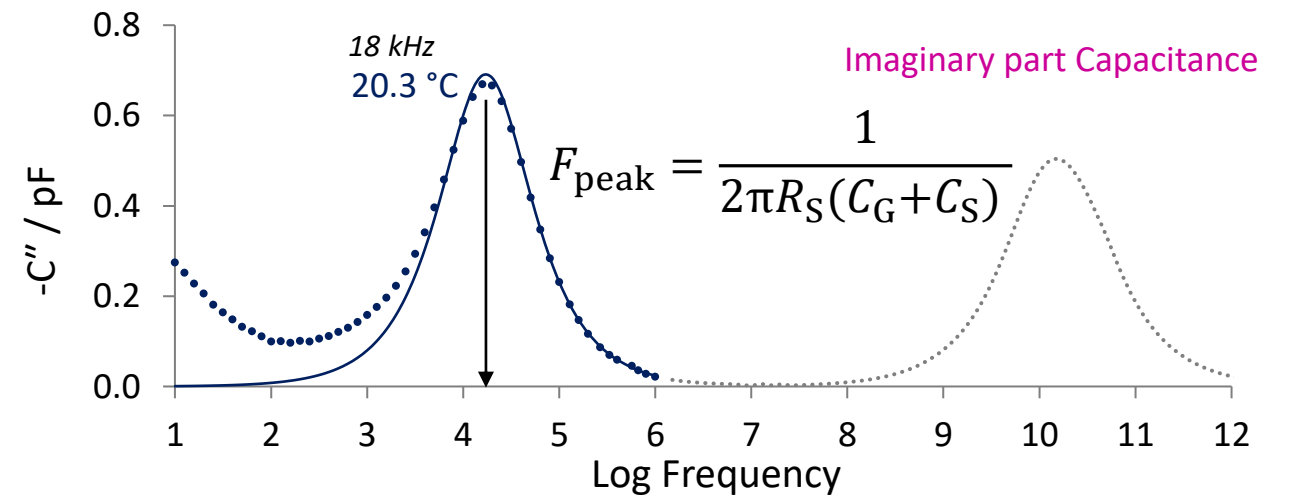
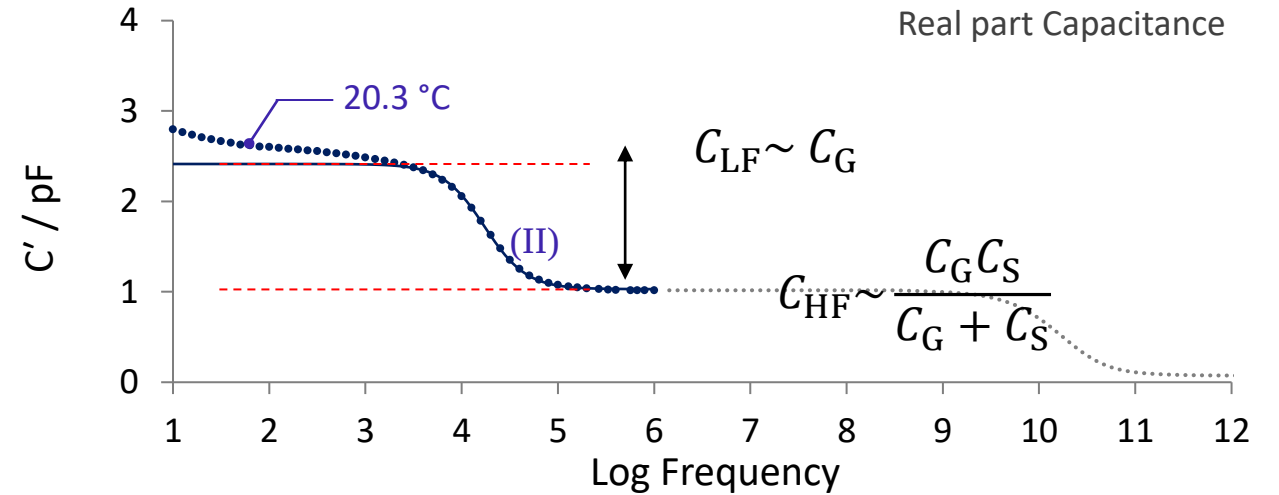
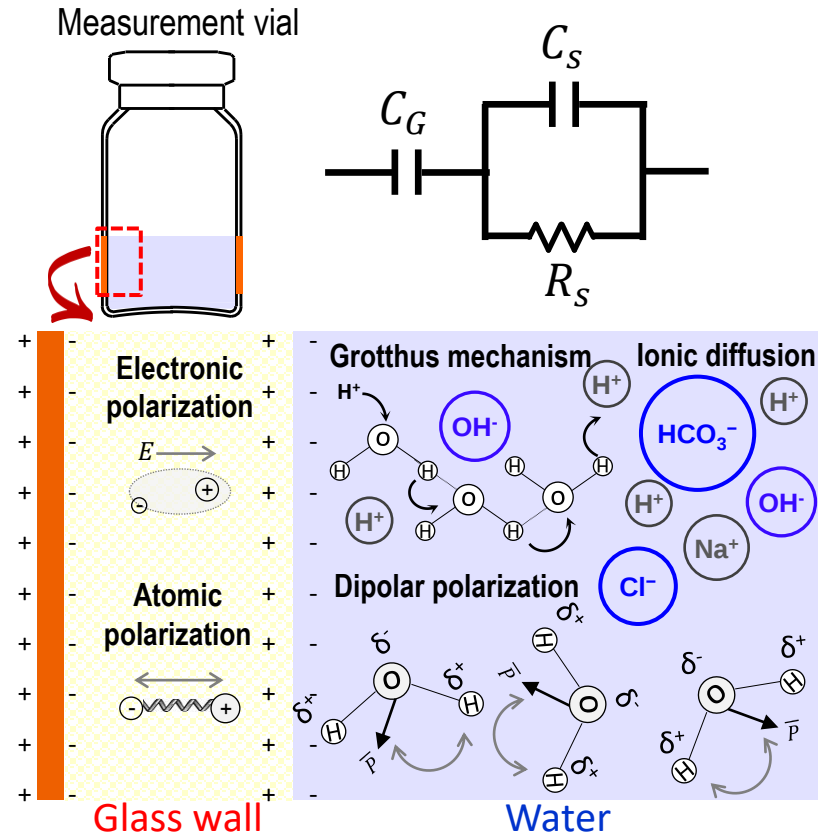


<https://en.wikipedia.org/wiki/Permittivity>



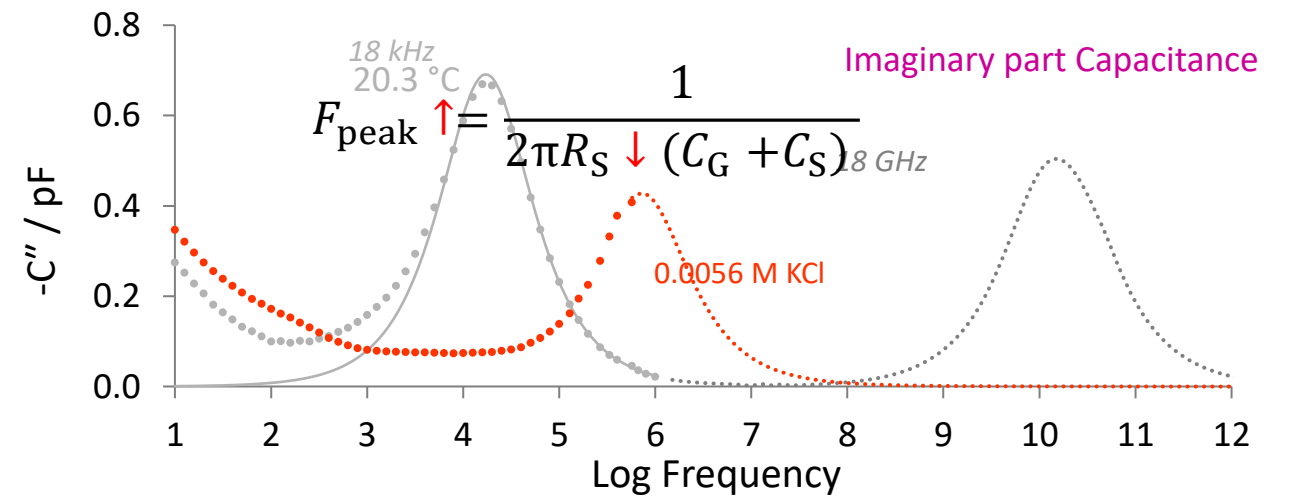
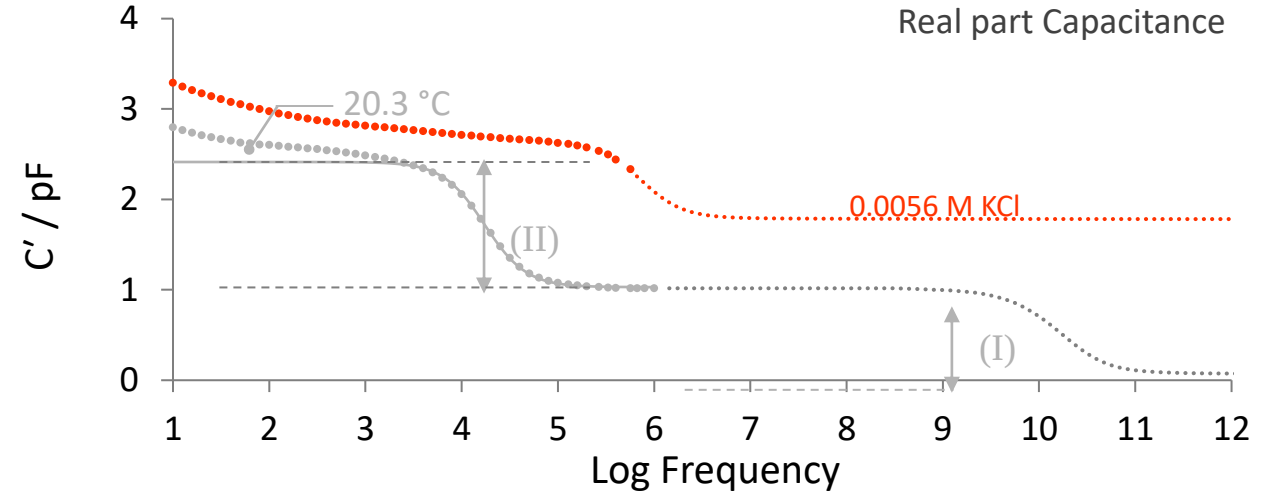
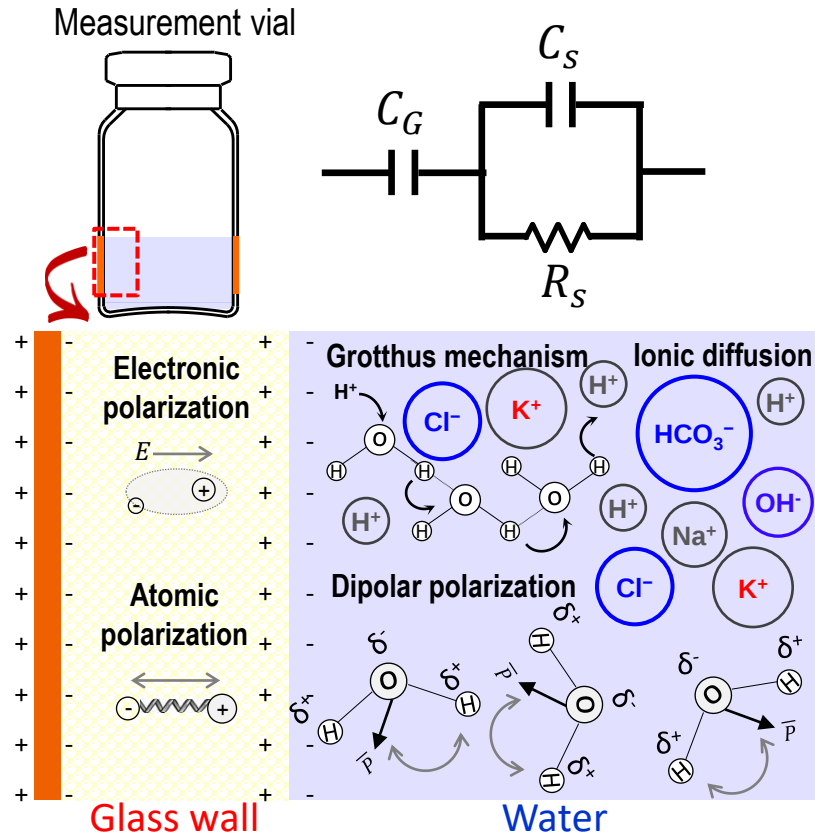
Dielectric Loss Mechanisms : Glass vial containing water

- II. **Maxwell-Wagner (MW)*** polarization of the glass wall of the TVIS vial at +20 °C, with a dielectric loss peak frequency of 17.8 kHz



Dielectric Loss Mechanisms

- II. Maxwell-Wagner (MW) polarization of the glass wall of the TVIS vial at +20 °C, with a dielectric loss peak frequency of 17.8 kHz

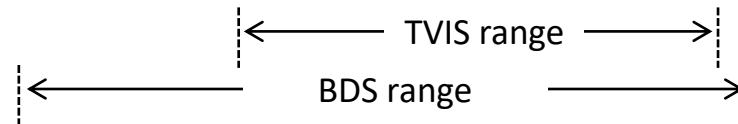
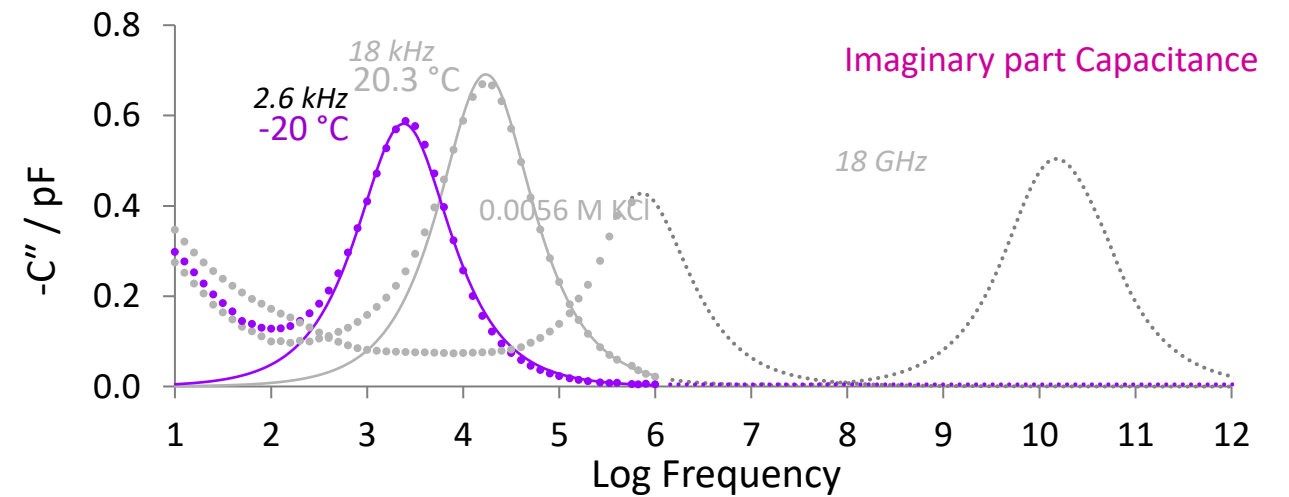
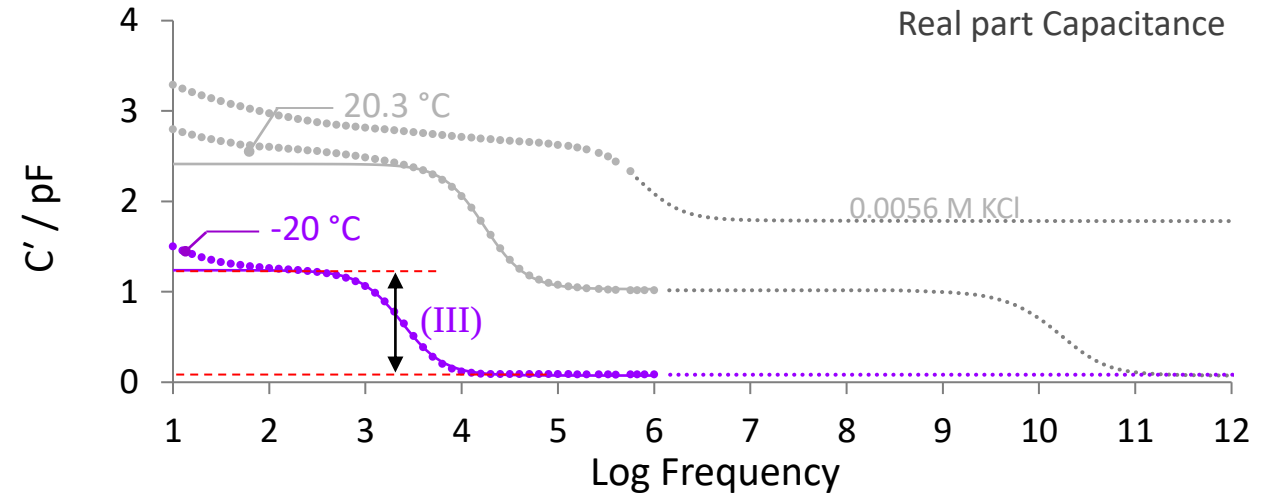
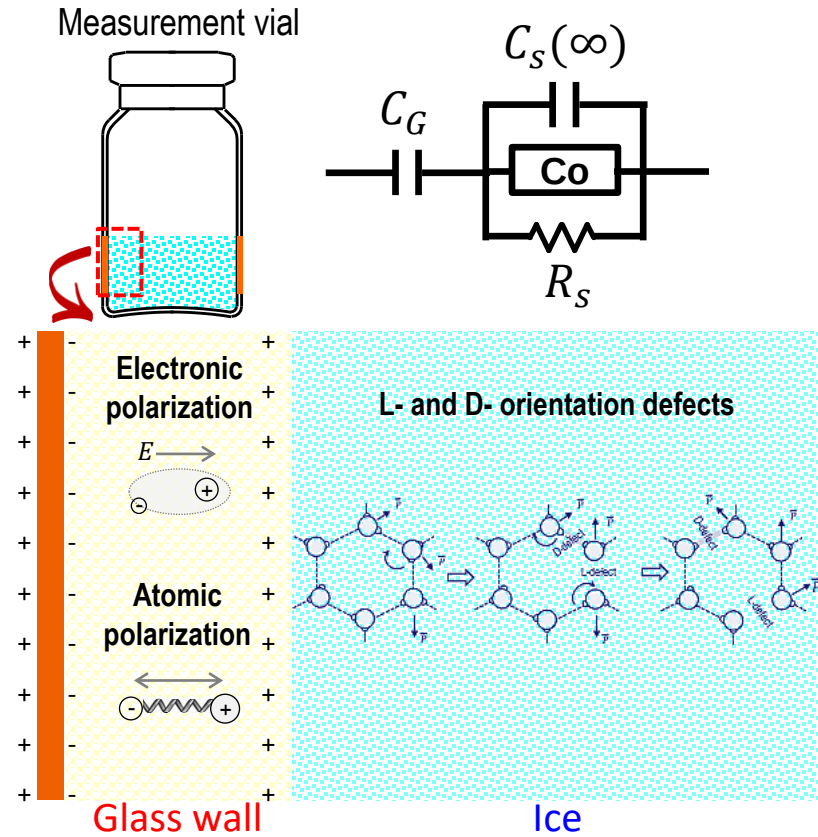


Dielectric properties of ice



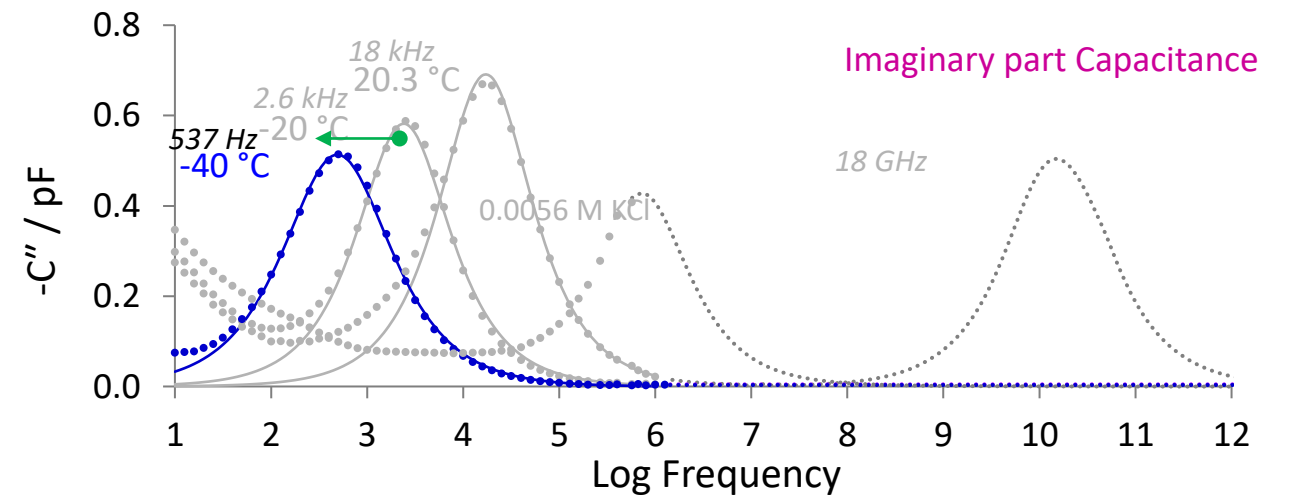
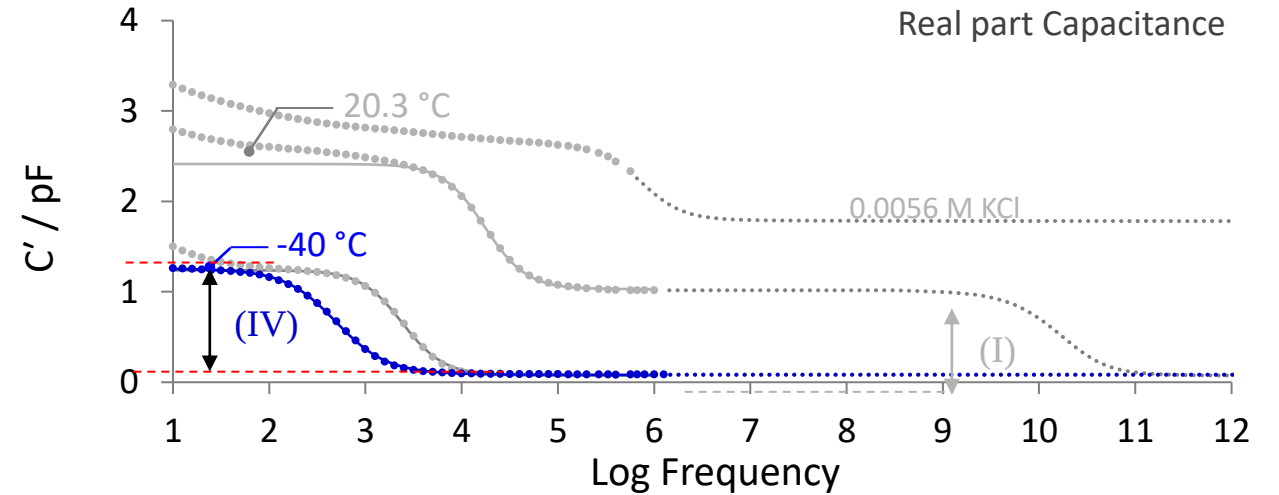
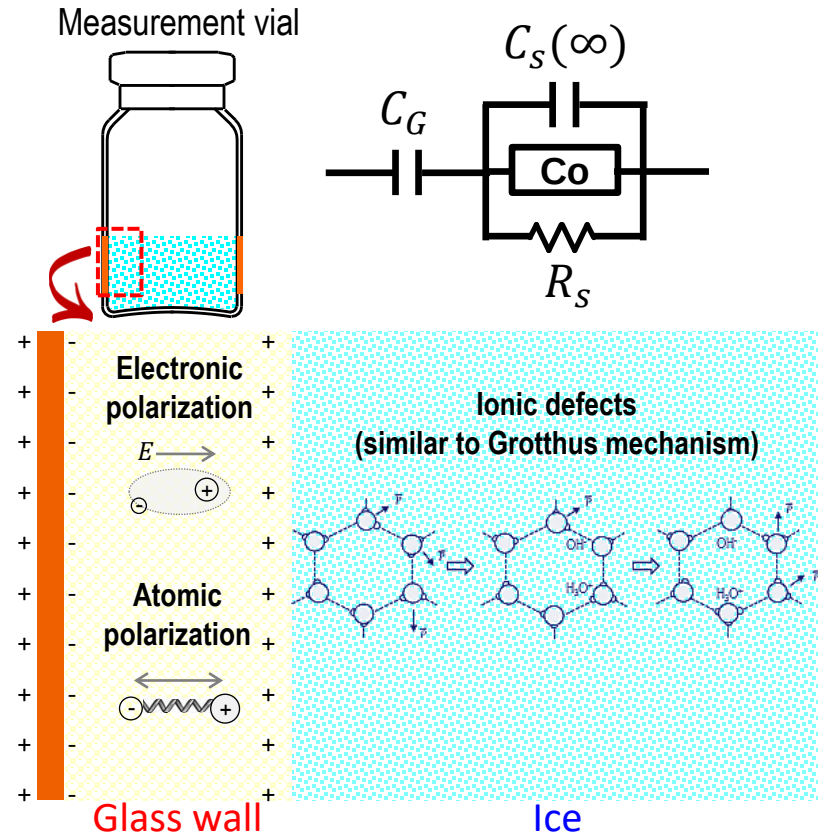
Dielectric Loss Mechanisms

- III. The dielectric polarization of ice at $-20\text{ }^{\circ}\text{C}$, with a dielectric loss peak frequencies of 2.57 kHz



Dielectric Loss Mechanisms

- IV. The dielectric polarization of ice at $-40\text{ }^{\circ}\text{C}$ with a dielectric loss peak frequencies of 537 Hz.



TVIS Applications



CASE STUDY 1

TVIS temperature calibration

Method 1 : Triangulation

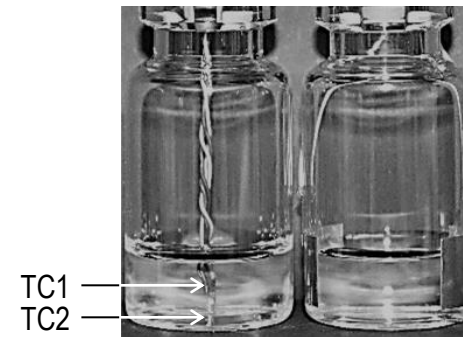
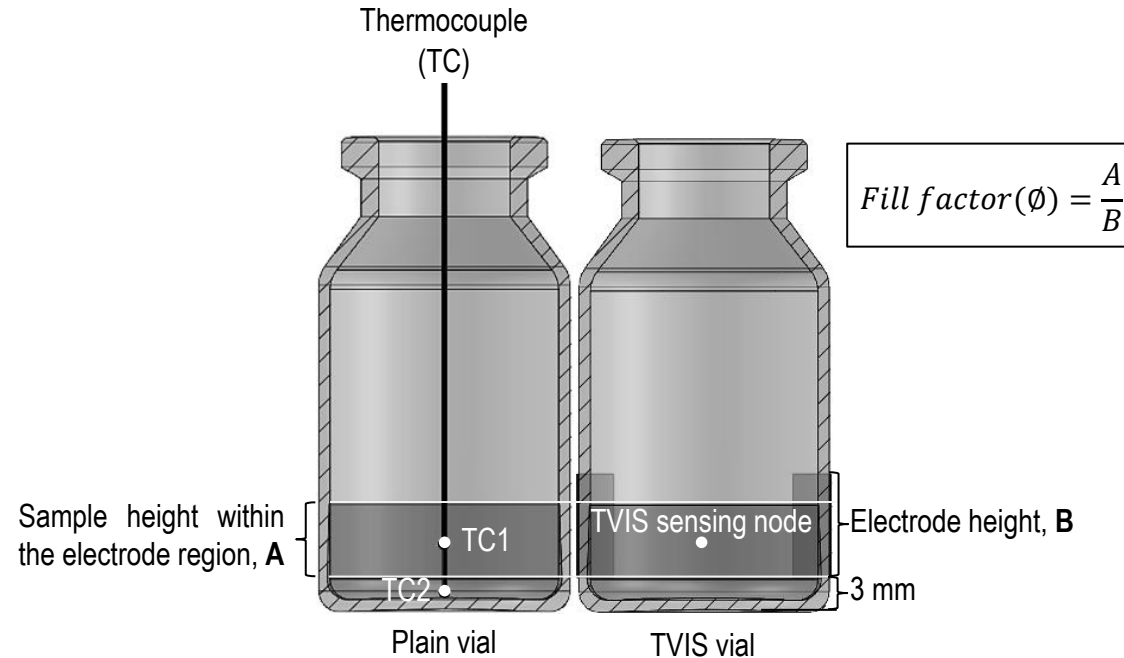
Method 2 : Tempris®



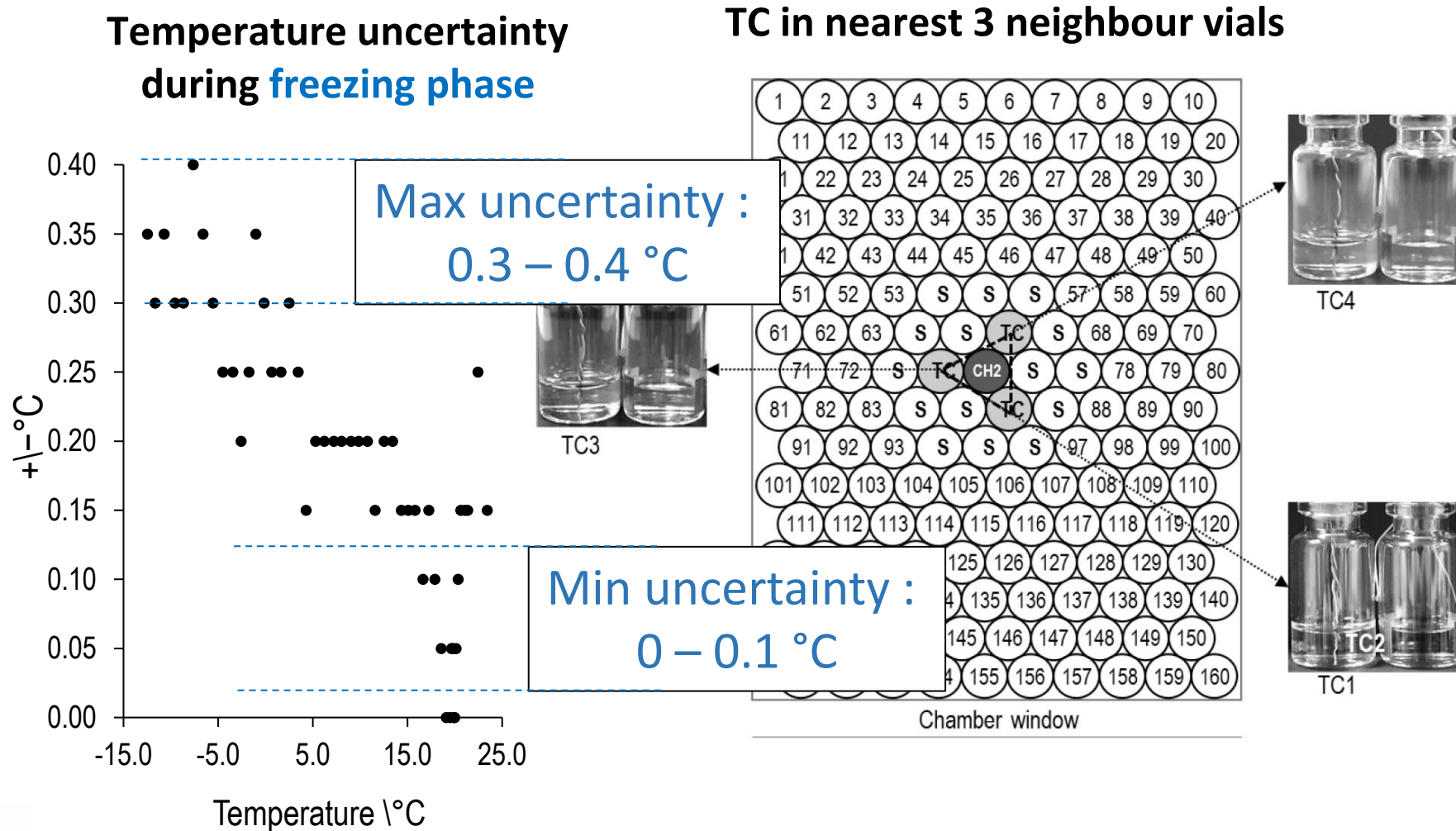
1. Triangulation method

Placing a thermocouple at the TVIS sensing node allows for the calibration of the temperature inside the TVIS vial to a precision of +/- 0.4 C
(see next two slides)

TC in nearest neighbour vial

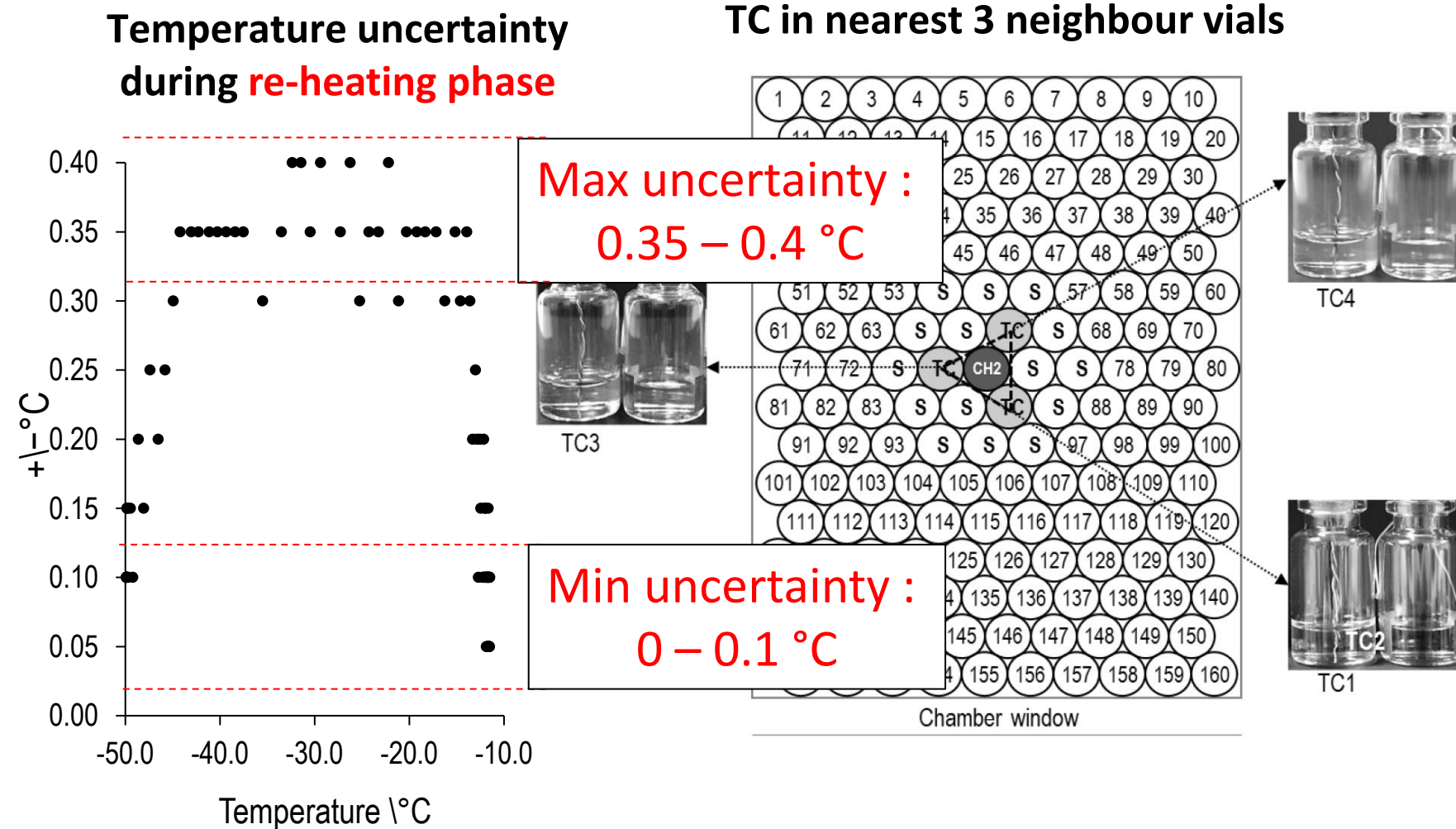


1. Triangulation method



Temperature calibration for the TVIS vial:

1. Triangulation method

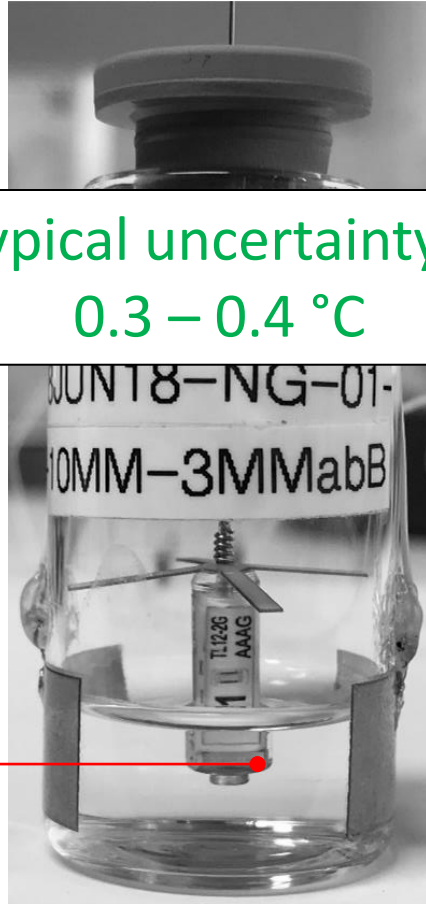


Temperature calibration for the TVIS vial:

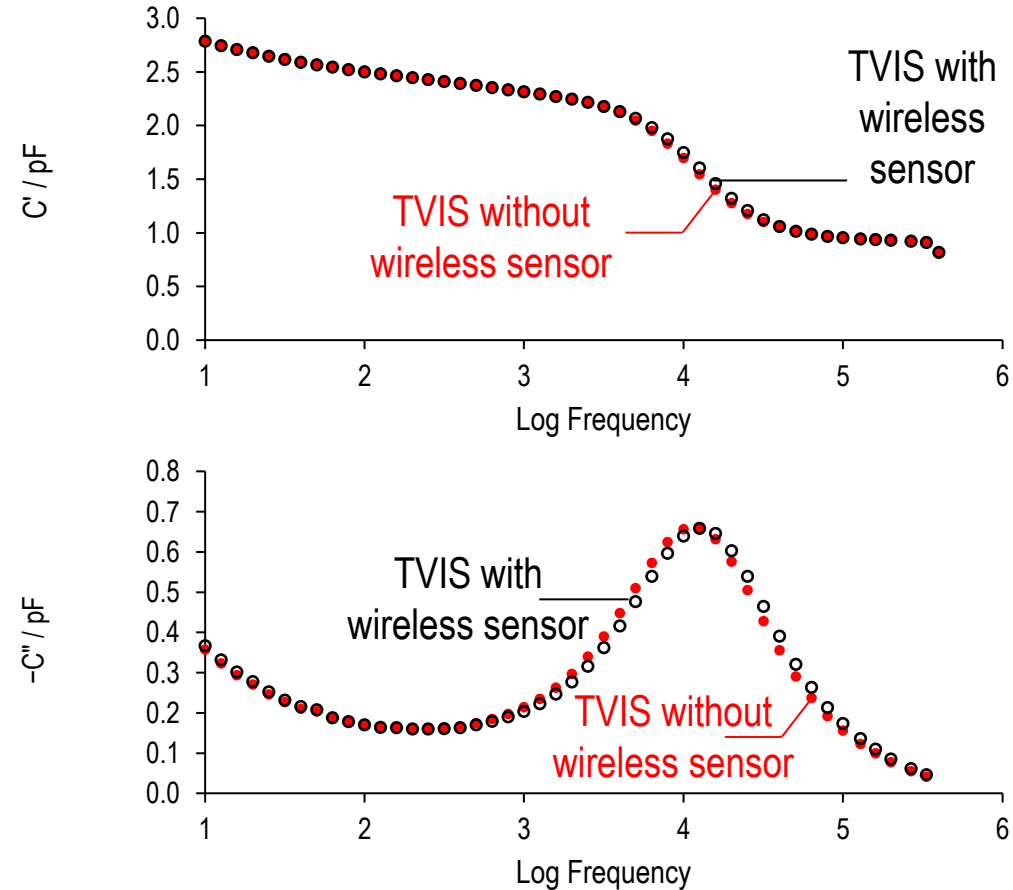
2. Tempris® method

Jeeraruangrattana (2020) PhD Thesis <https://dora.dmu.ac.uk/handle/2086/20278>

Typical uncertainty :
0.3 – 0.4 °C

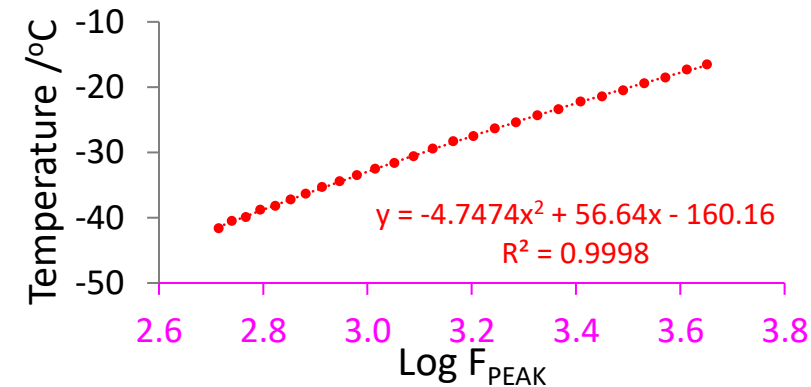
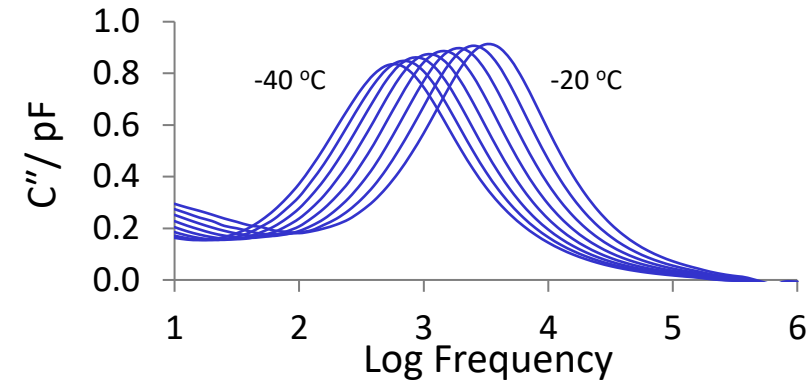
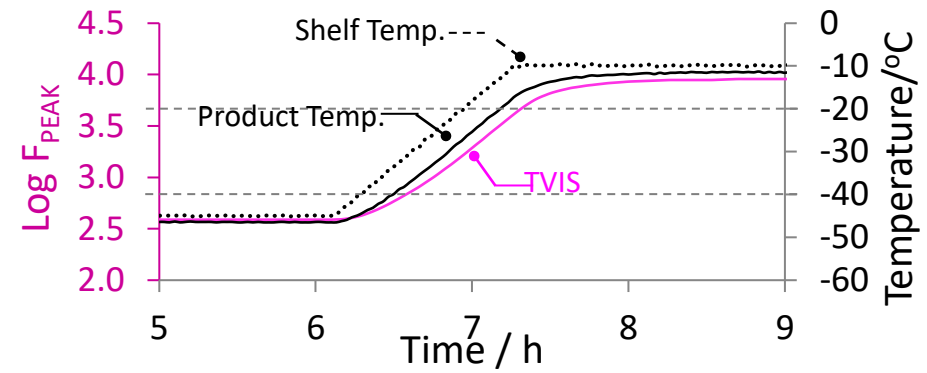
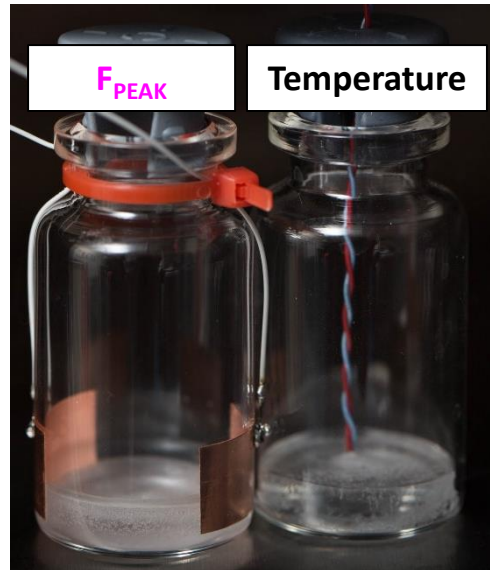


TVIS spectra



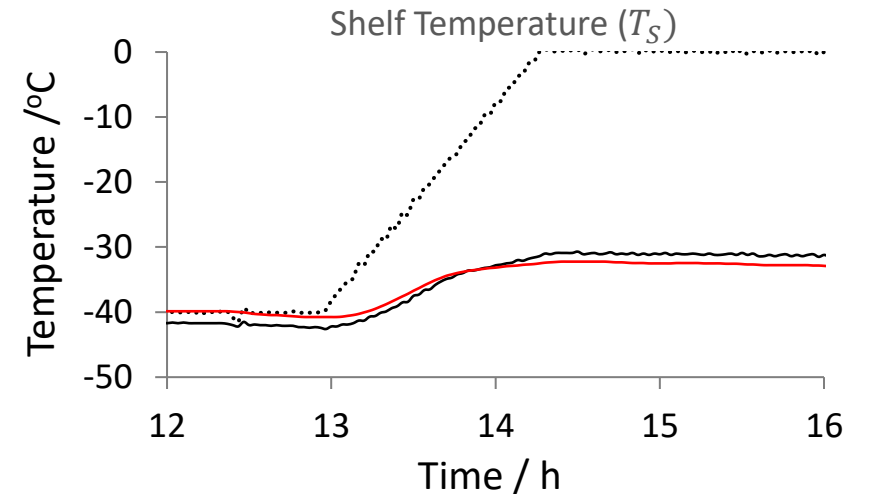
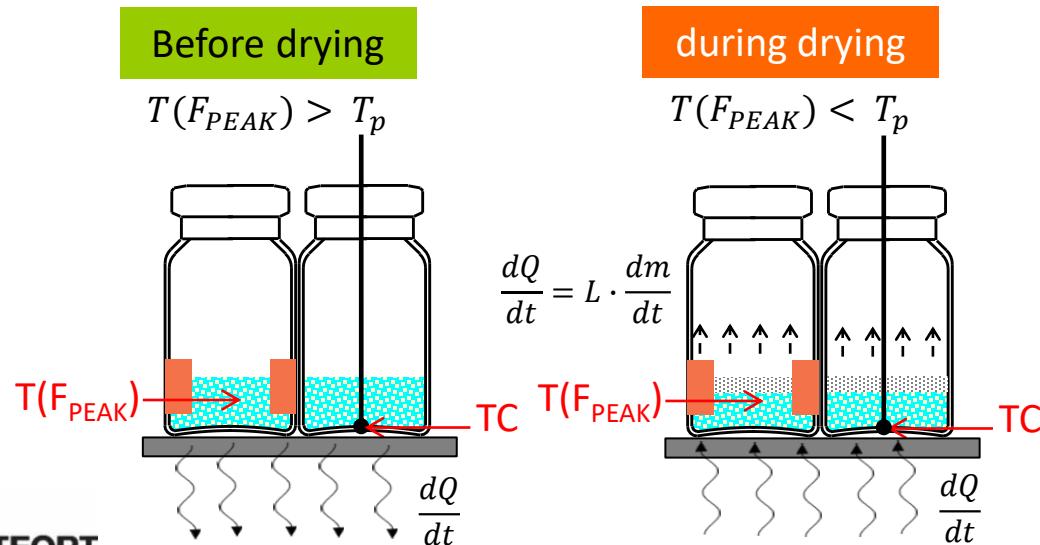
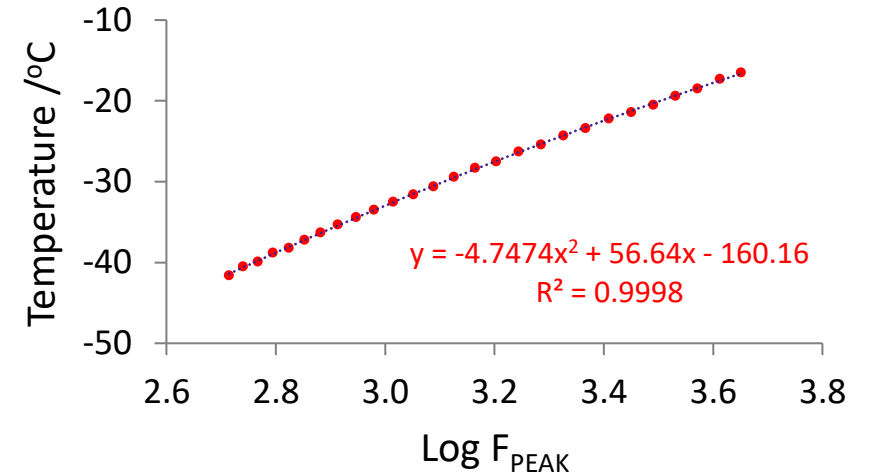
Temperature Calibration

- F_{PEAK} profile during annealing has 'similar' profile with product temperature.
- Assuming thermal equivalence between the thermocouple (TC) vial and TVIS vial, then the temperature calibration from annealing might be employed for the prediction of temperature during primary drying



Temperature Prediction in Primary Drying

- Temperature calibration curve selected for temperature prediction in primary drying : $T(F_{PEAK})$
- Good agreement between product temperature (by TC) and $T(F_{PEAK})$

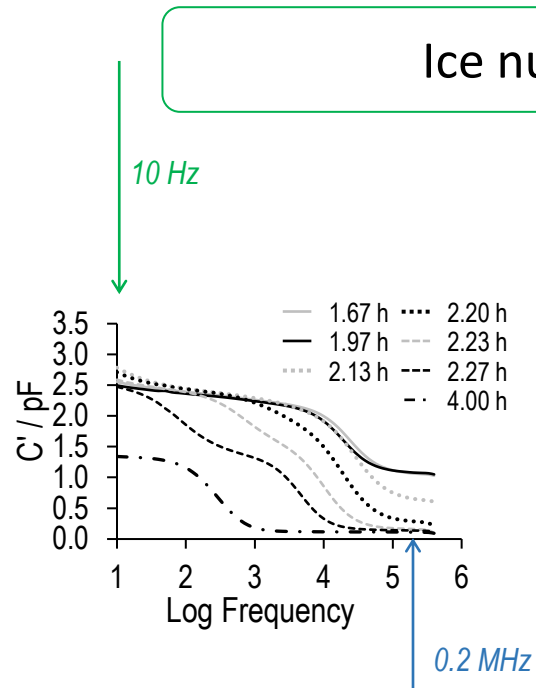


CASE STUDY 2

Ice solidification end point



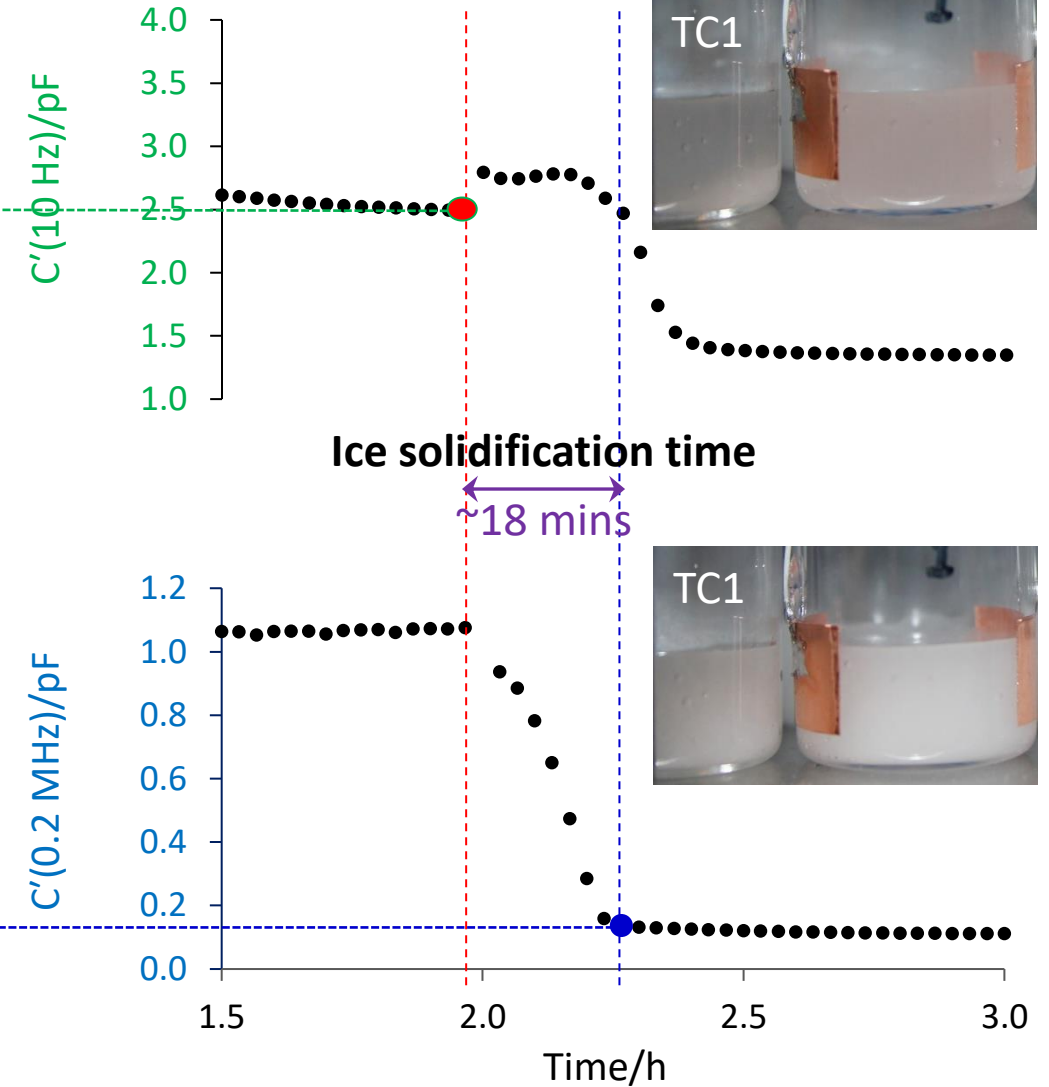
Solidification period



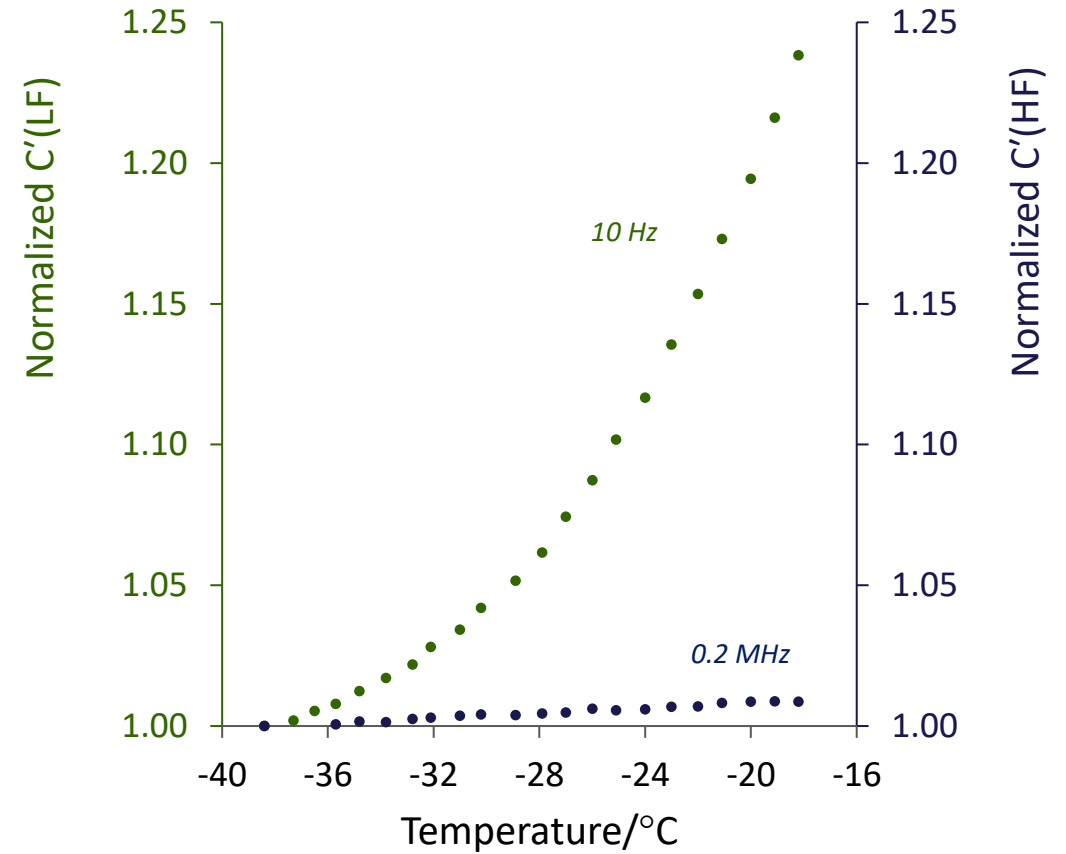
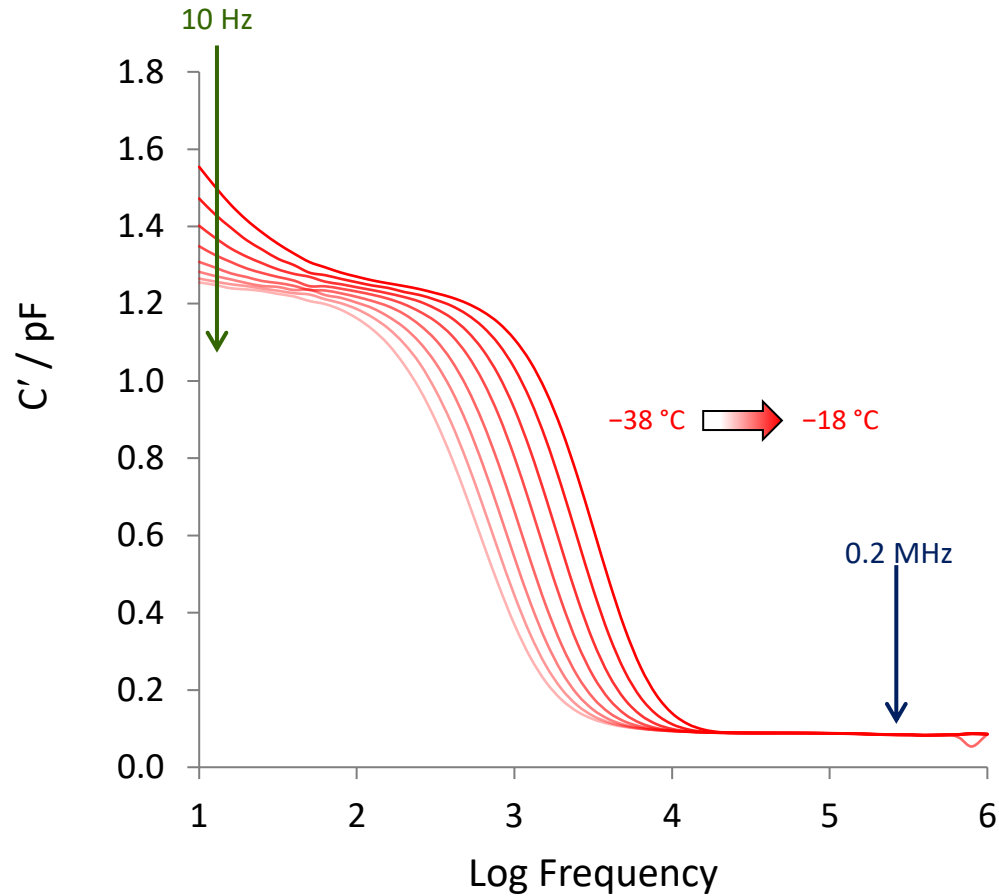
Ice nucleation

- The difference between these two times if the ice solidification time
- Knowing the height of the product in the vial one can then estimate an average solidification rate

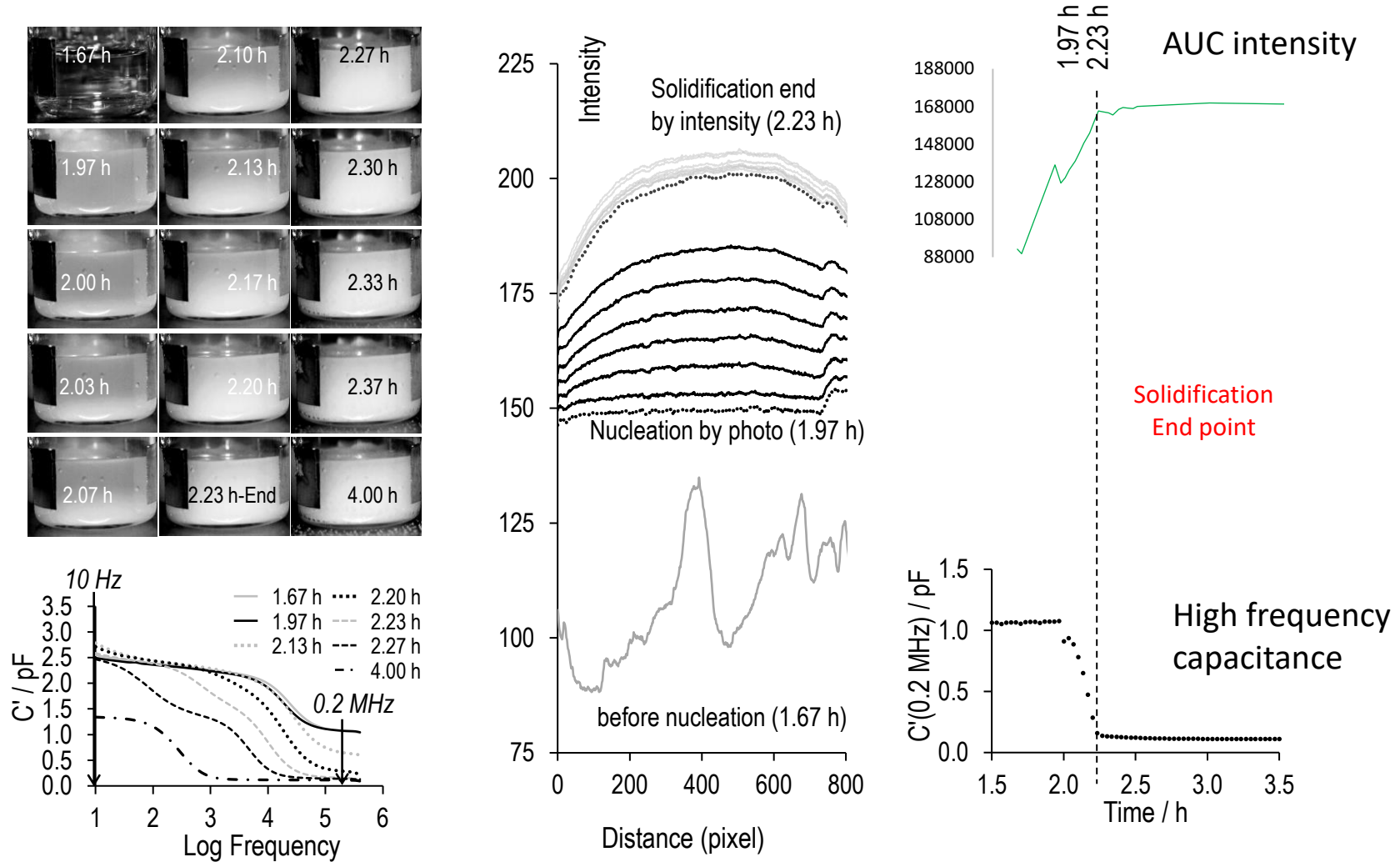
Solidification end point



Temperature Profile of Frozen Water



Solidification end point 5% sucrose



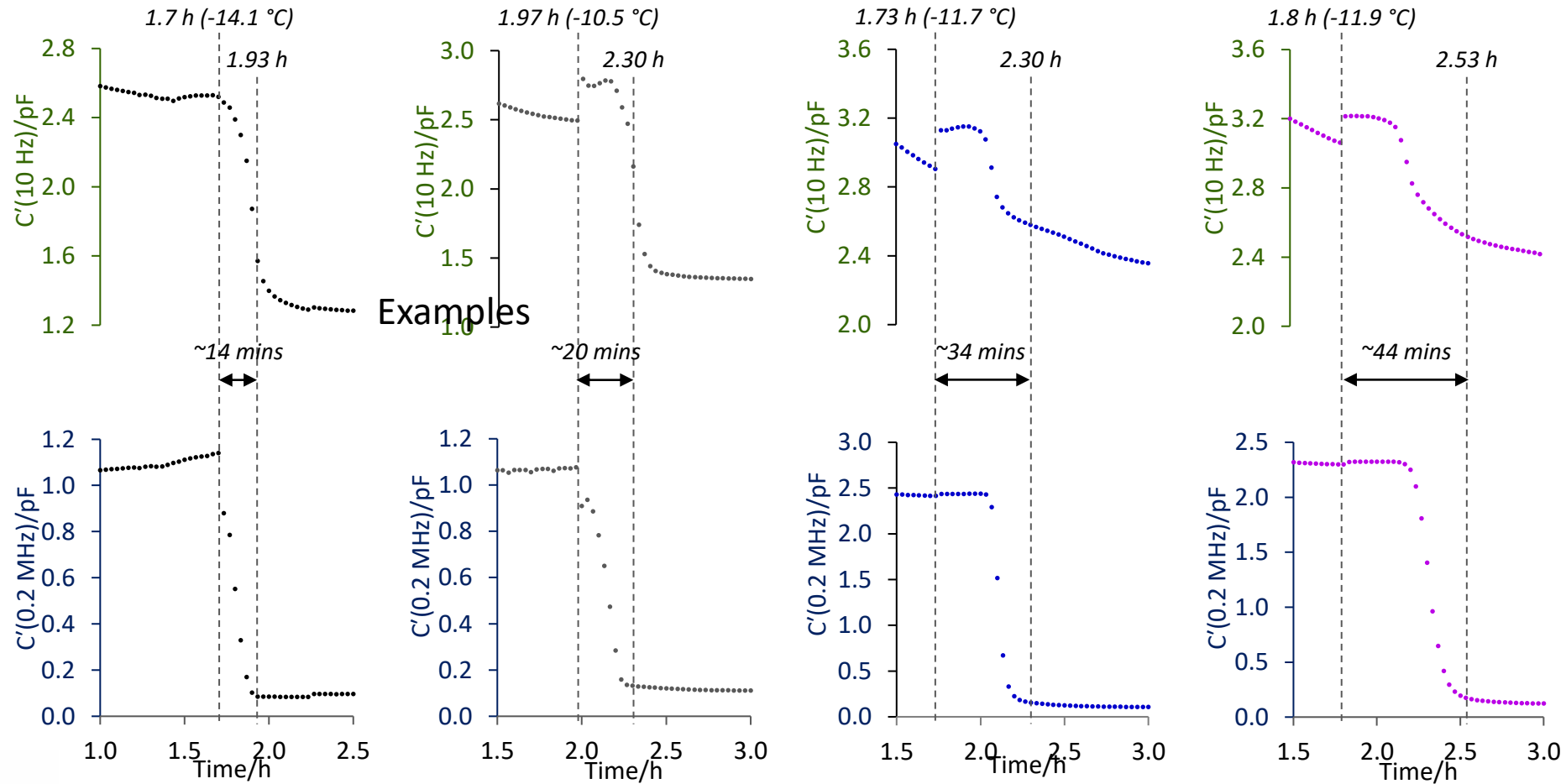
Water

5% Sucrose

5% Sucrose in
0.26% NaCl

5% Sucrose in
0.55% NaCl

Solidification time for ice formation increases with the salt concentration

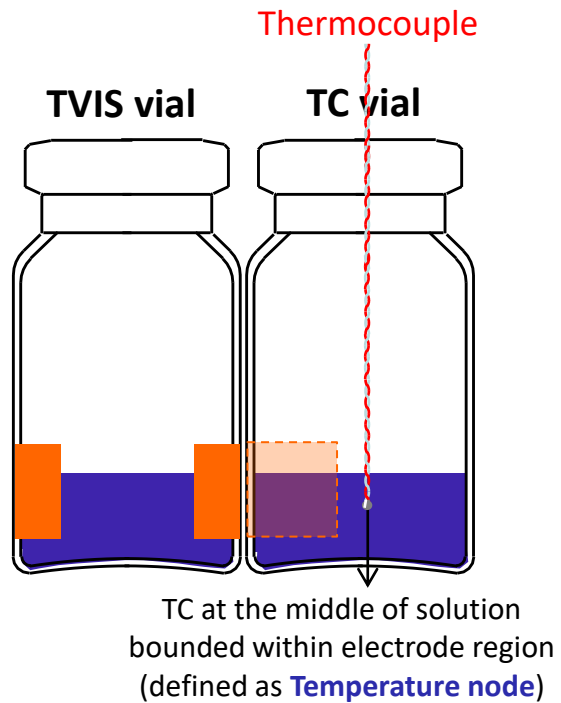


CASE STUDY 3

Determination of in-vial Glass Transition temperature (T'_g)

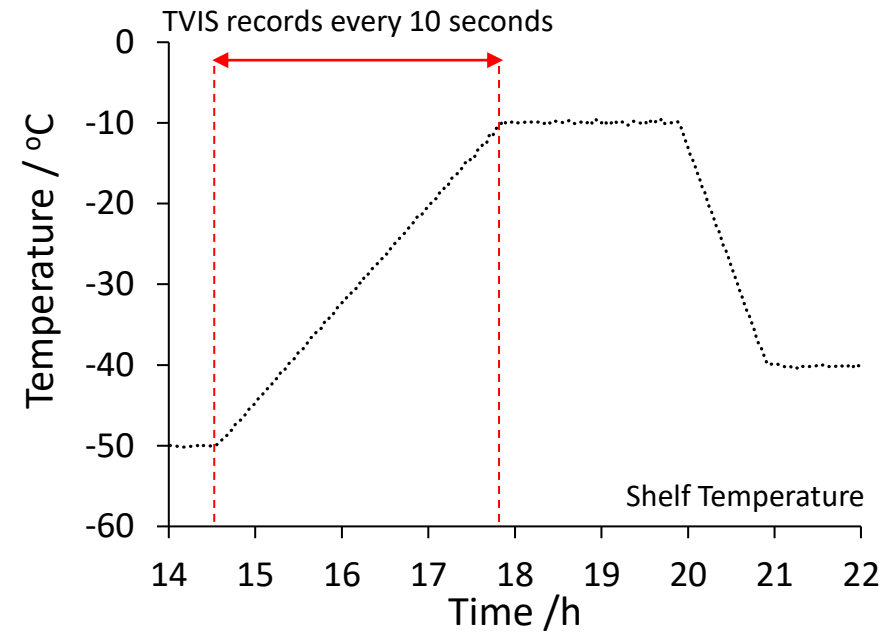


Glass Transition Temperature



Thermocouple position

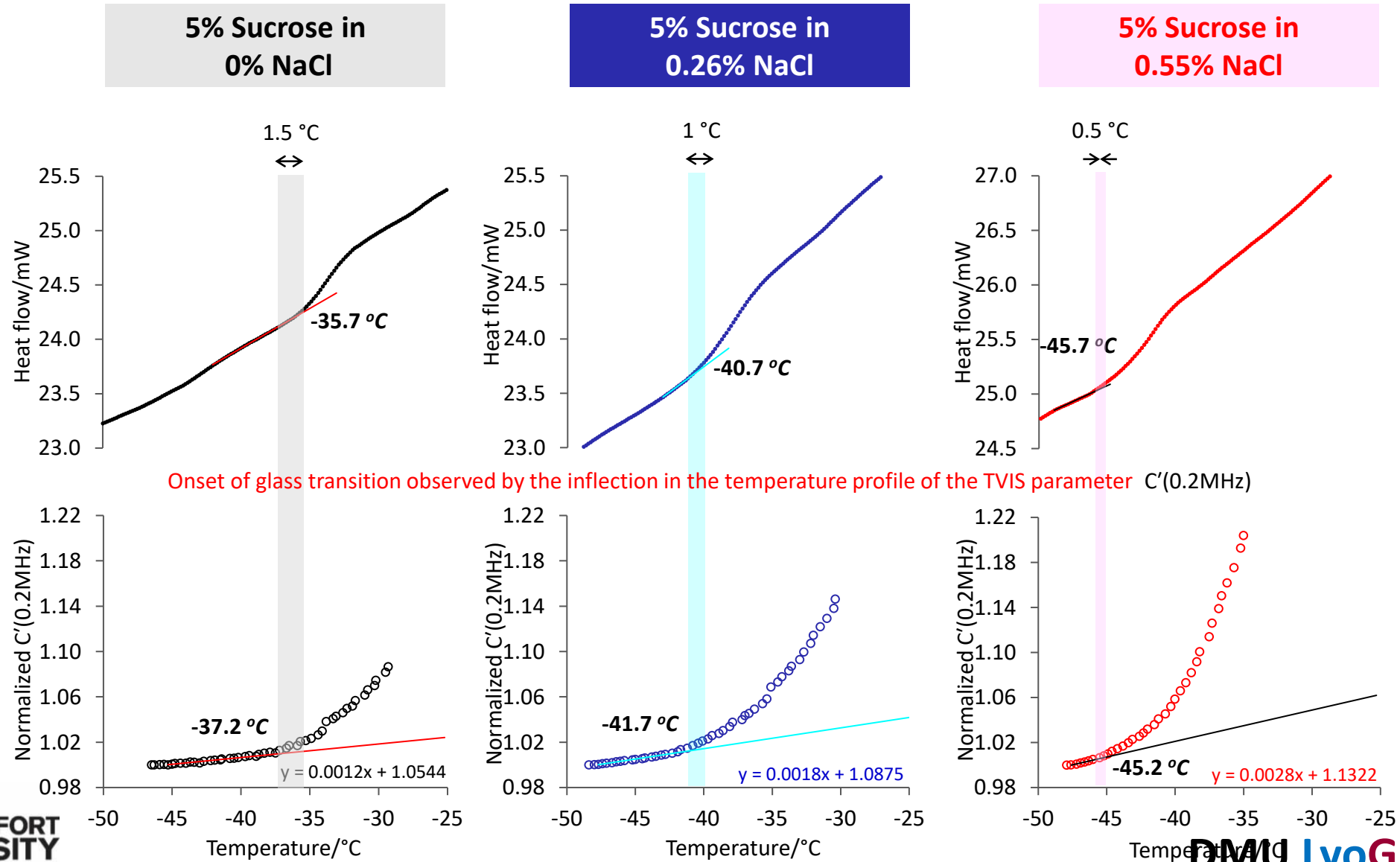
Annealing



Ramp from -50 °C to -10 °C with 0.2 /min



C' (0.2 MHz) during Re-heating

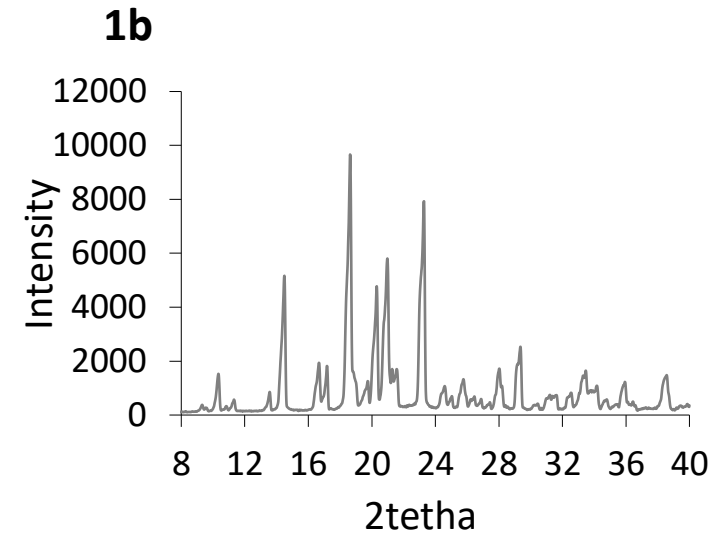
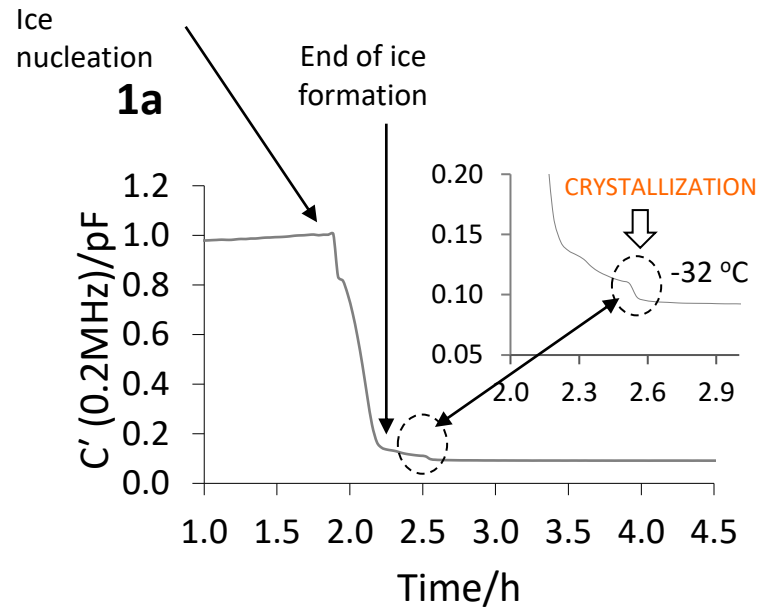


CASE STUDY 4

Freezing and annealing of mannitol solution



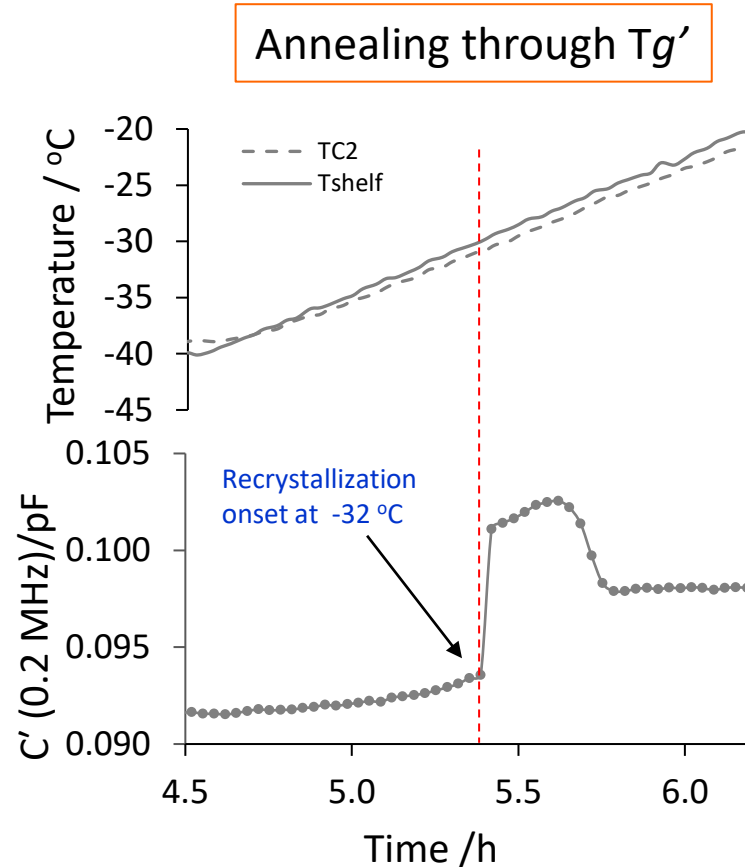
Crystallization events at freezing step for 5% mannitol



TVIS and XRD results for solution of 5% mannitol showing crystallization detected by TVIS



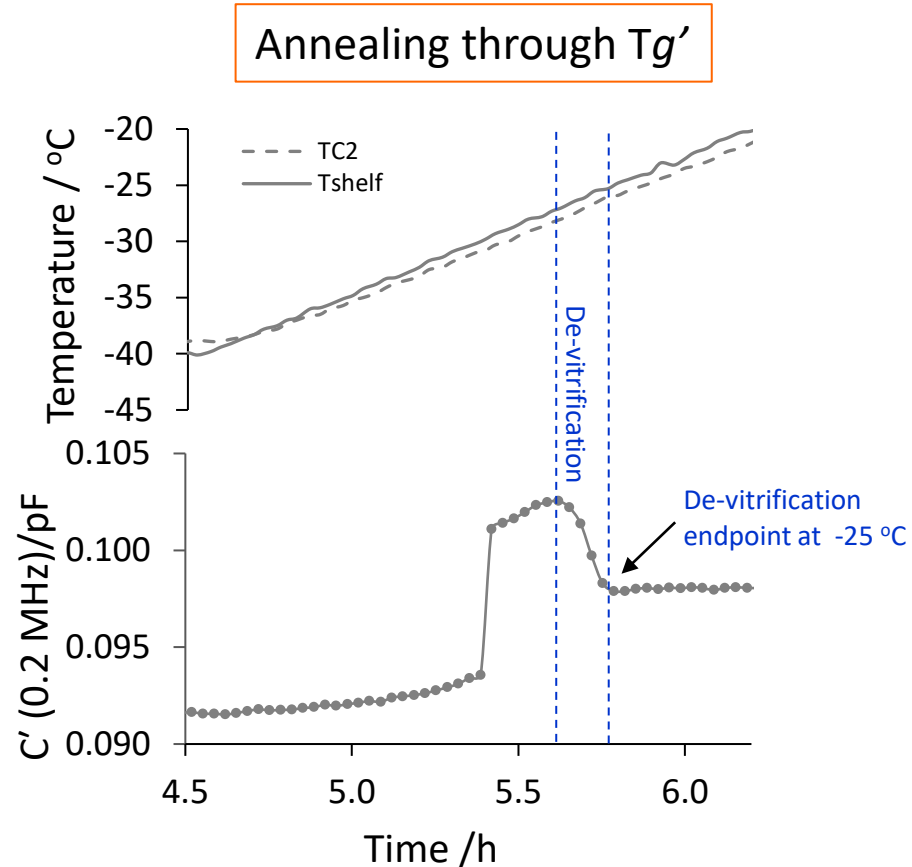
Recrystallization (dehydration) and de-vitrification? on annealing of 5% mannitol



On annealing to -20°C
 C' (0.2 MHz) shows
mannitol recrystallization with
onset at -32°C
A devitrification end point at -25°C



Recrystallization (dehydration) and de-vitrification? on annealing of 5% mannitol



On annealing to -20 °C
 C' (0.2 MHz) shows
mannitol recrystallization with
onset at -32 °C
A devitrification end point at -25 °C

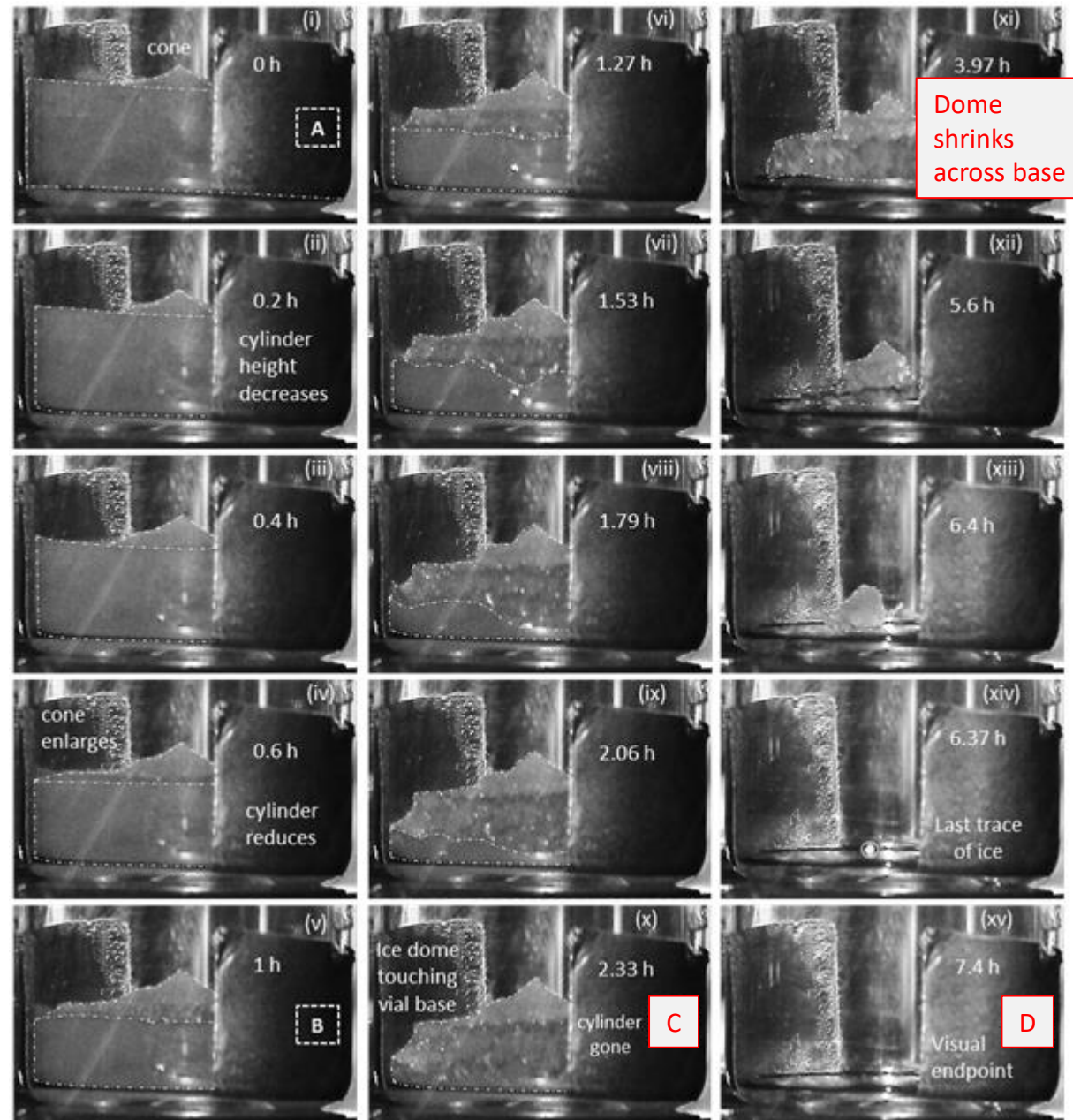
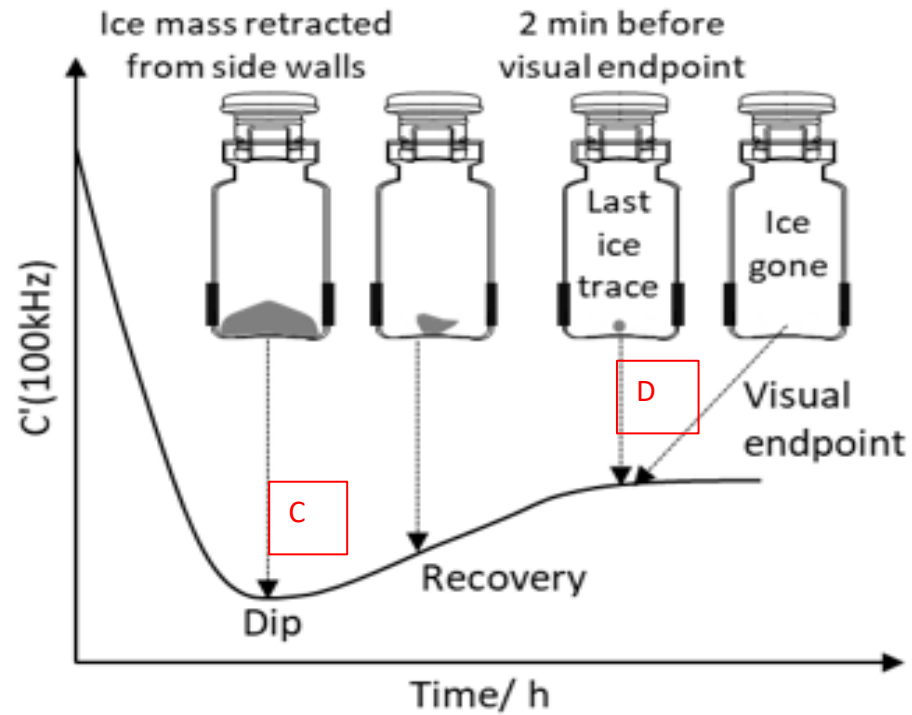


CASE STUDY 5

Sublimation end point



Sublimation end point

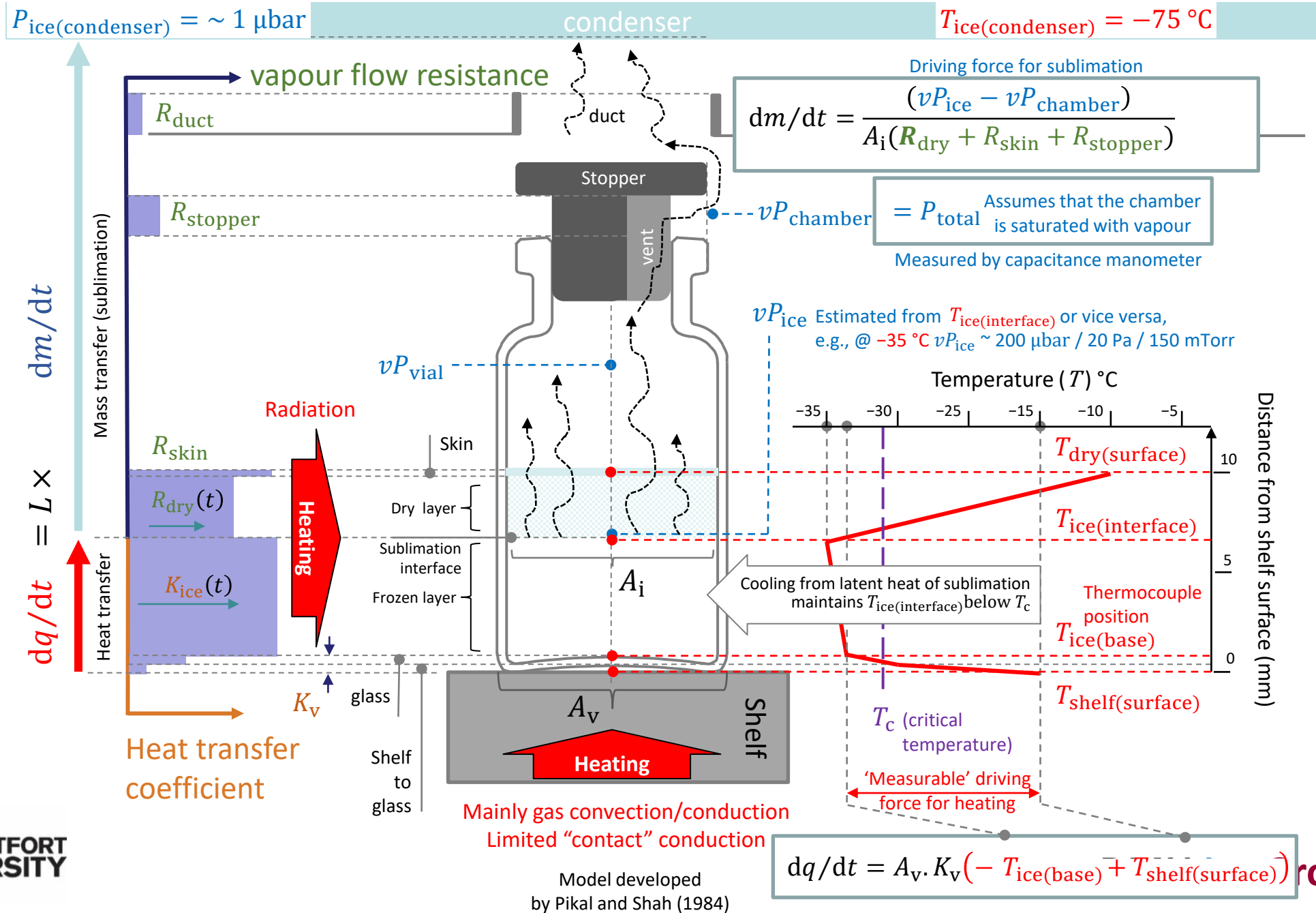


CASE STUDY 6

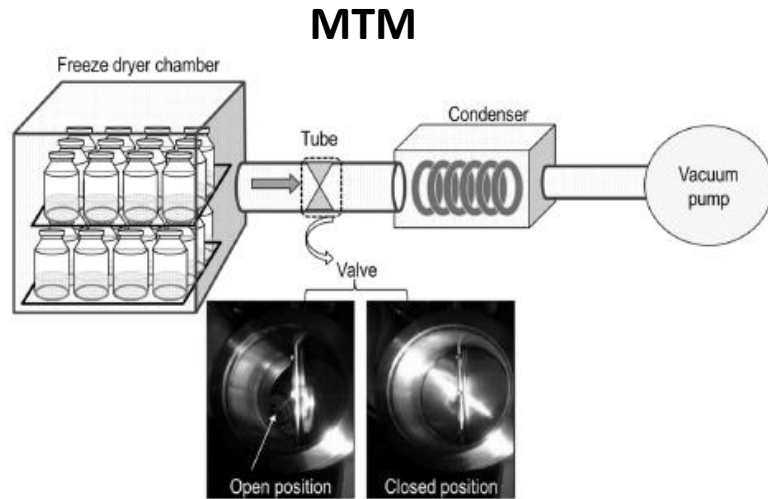
Primary drying rate models

Assumptions of a planar ice interface





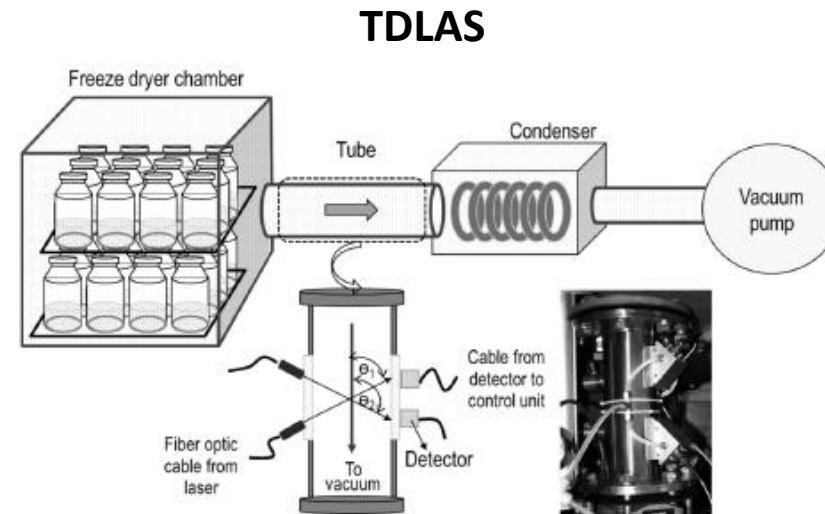
Drying rate based on vapour pressure measurement



- Increase the pressure in the freeze dryer (same as PRT)
- MTM combines the pressure rise data with a mathematical equation to predict drying rates

From drying rate calculate:

- (i) heat transfer coefficient; (ii) batch 'average' temperatures (@ ice front & ice base), (iii) drying endpoints, (iv) dry layer resistance



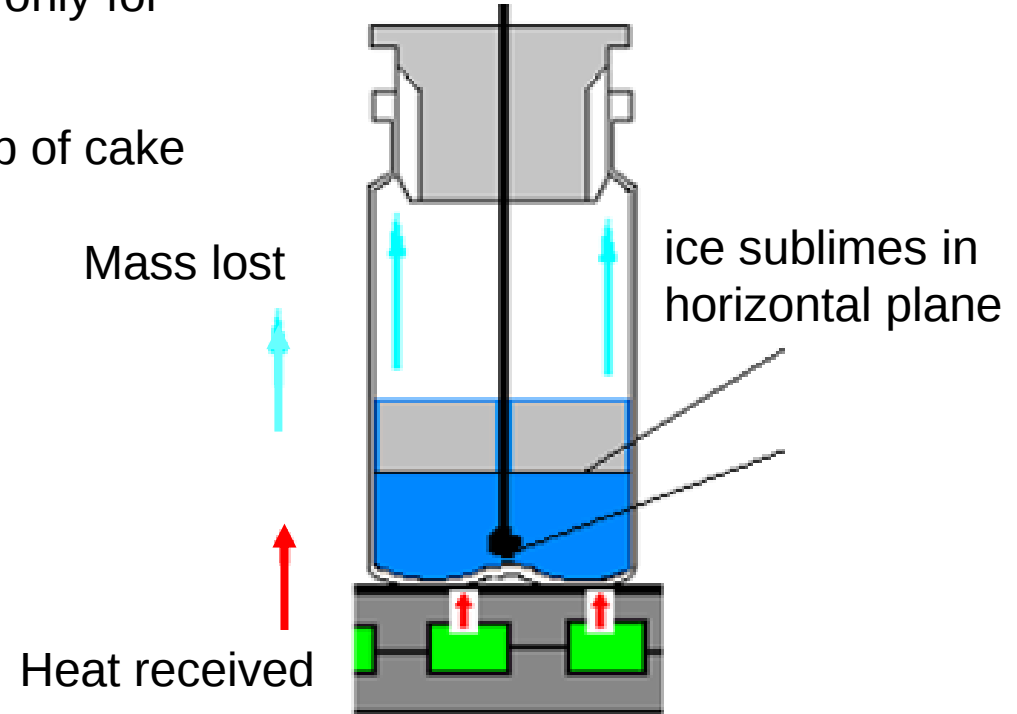
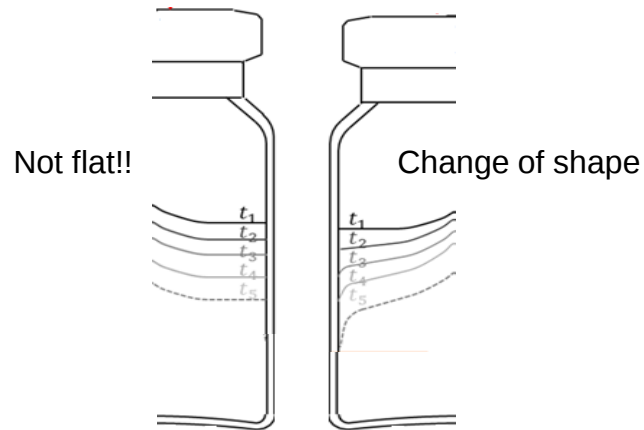
- Laser assembly tube is connected between chamber and condenser
- Drying rate determined from
 - Laser light absorbed is proportional to the concentration of gas/water vapour
 - Doppler effect used to determine velocity of the vapour



Heat and Mass Balance: Assumptions

1. All heat received by product is used only for sublimation of water.
2. Sublimation front moves from the top of cake parallel to the vial bottom

Reality:



3. The contribution of radiation component to the vial heat transfer coefficient is constant within entire operation temperature range

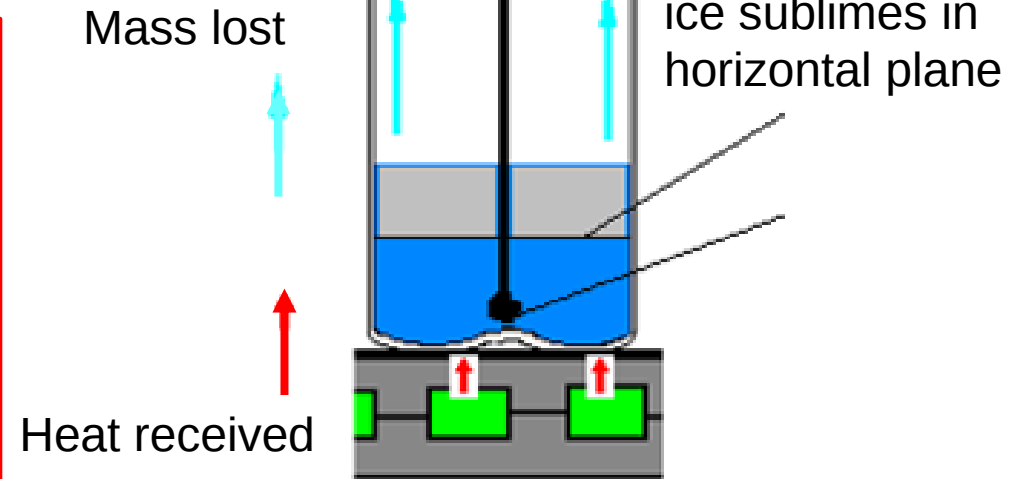
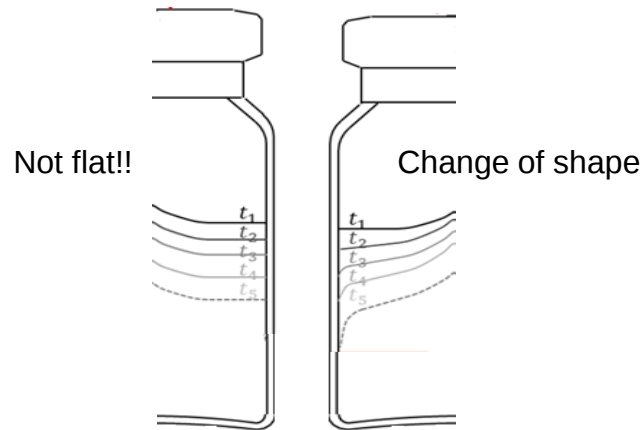
Pikal et al. (1984) J Pharm Sci 73:1224
Temperature measurements have to be completed before 15% of the ice mass is removed before the assumption of a planar ice-surface interface is seriously violated



Heat and Mass Balance: Assumptions

1. All heat received by product is used only for sublimation of water.
2. Sublimation front moves from the top of cake parallel to the vial bottom

Reality:



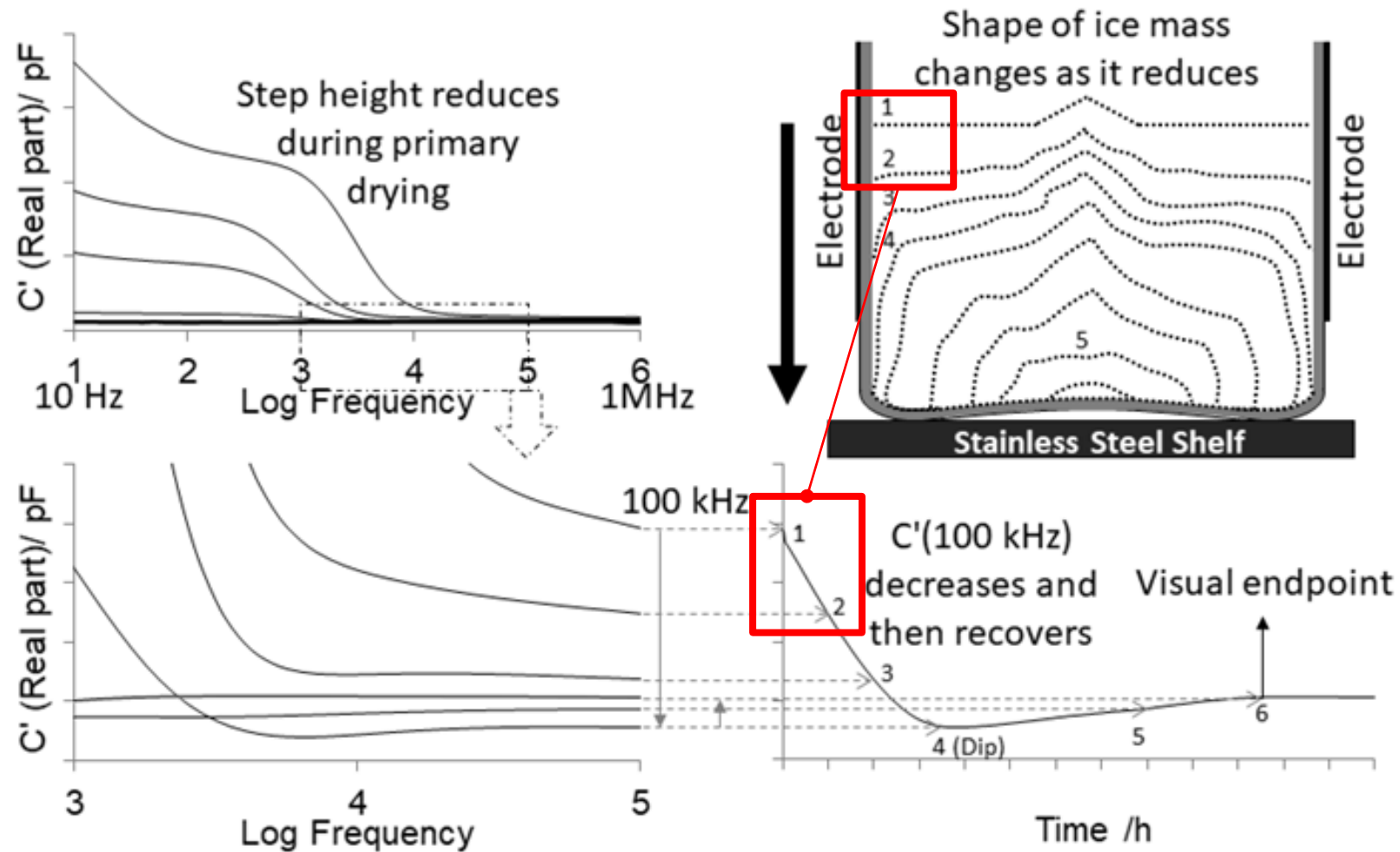
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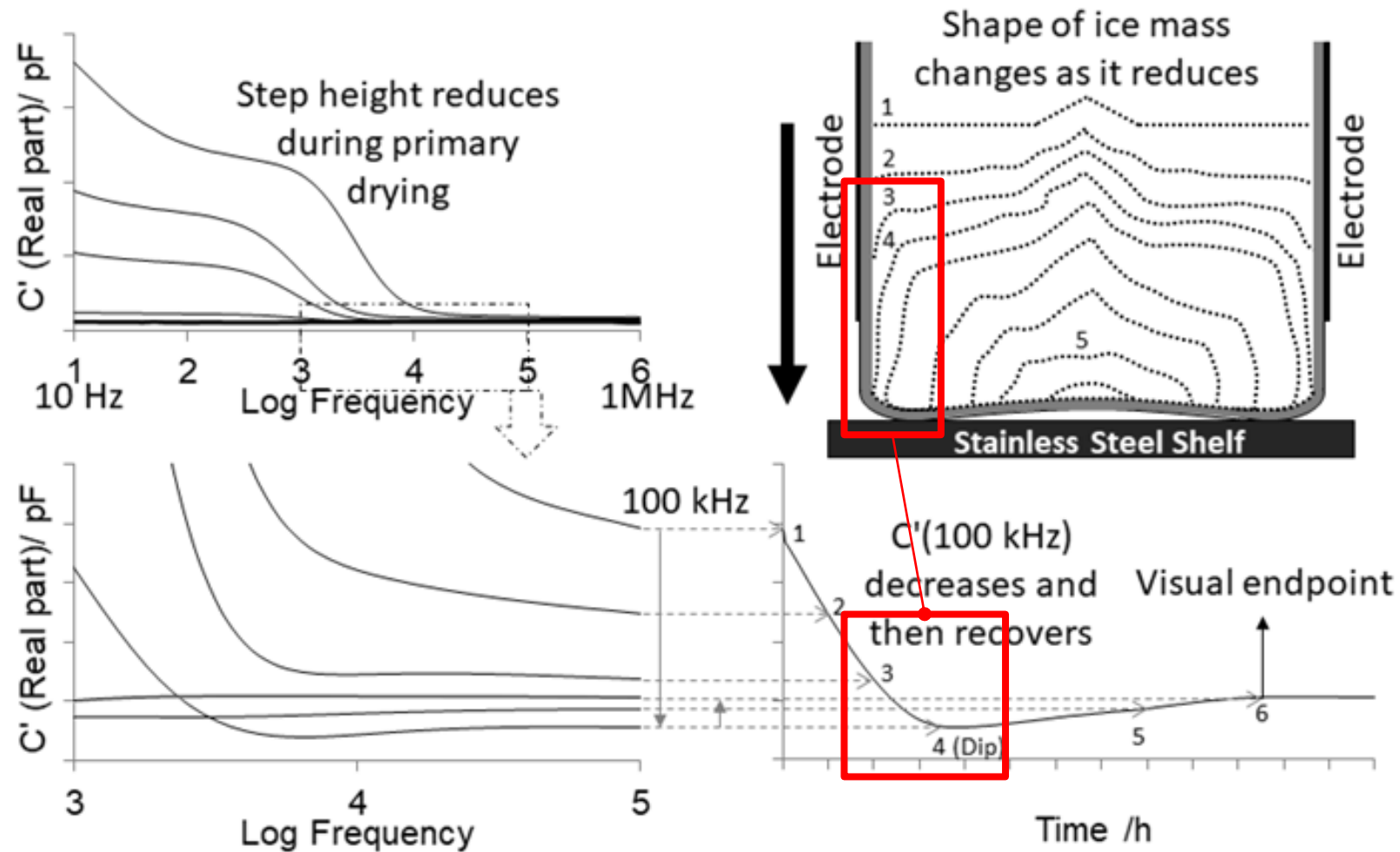
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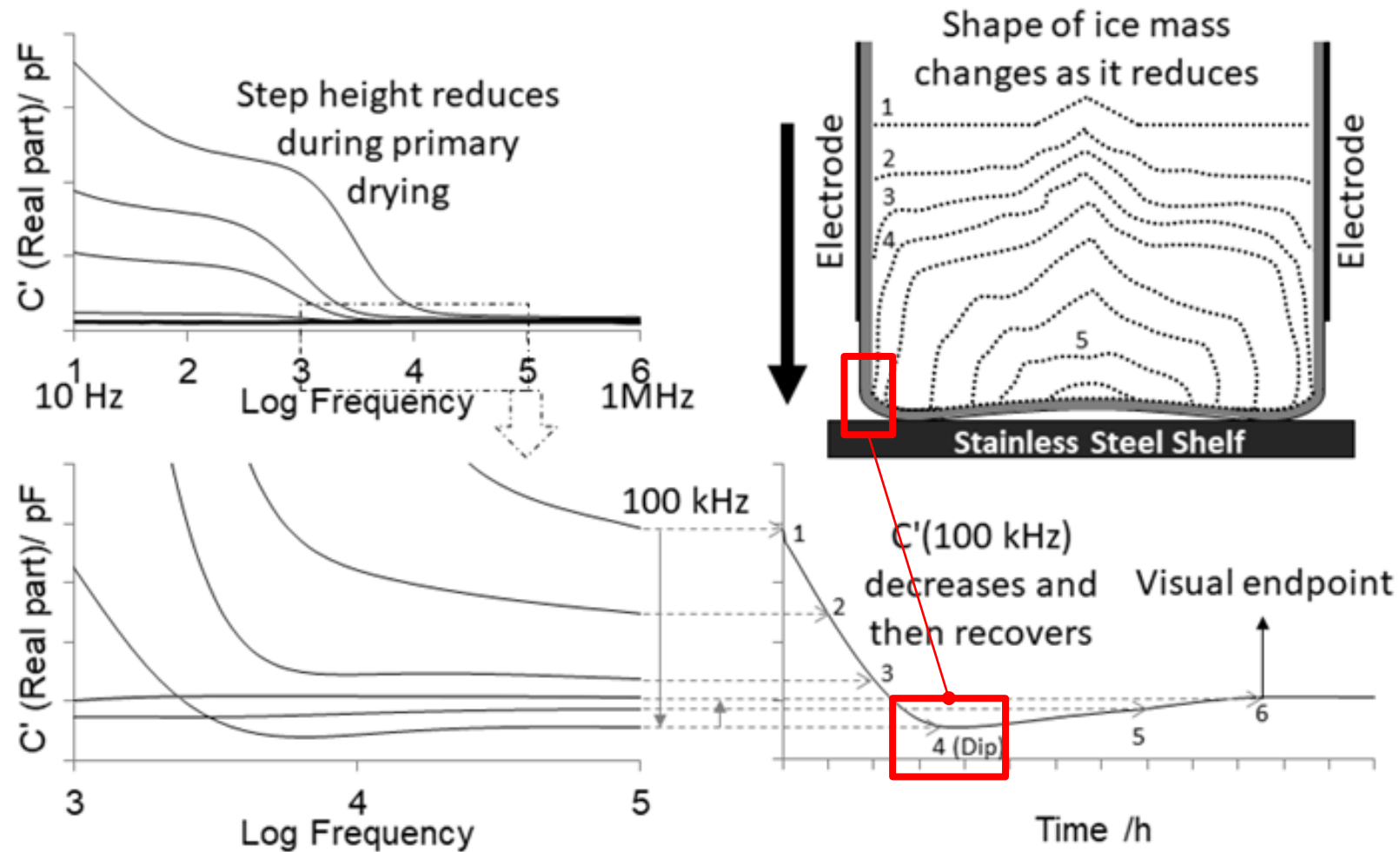
TVIS application in studying ice mass shape



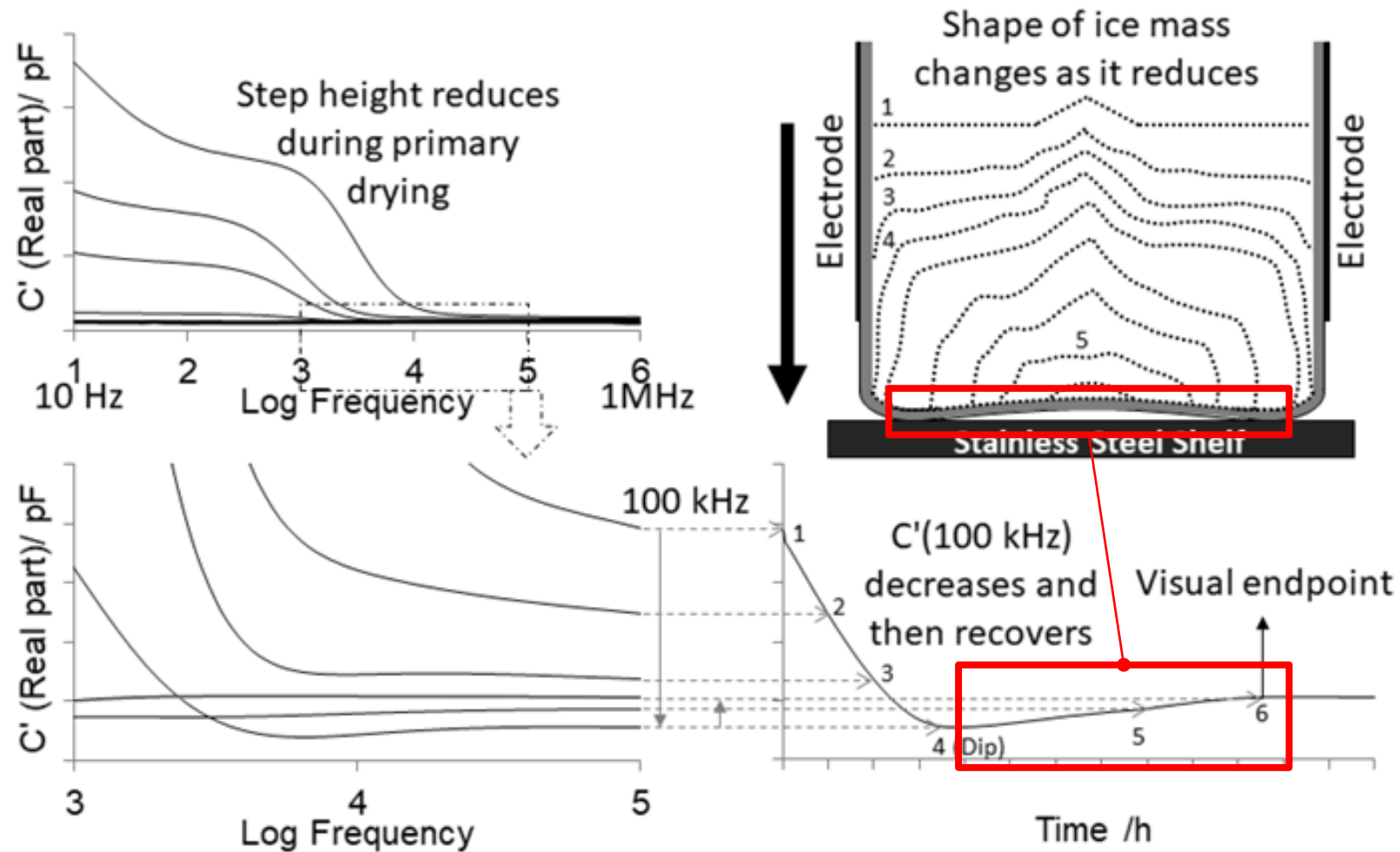
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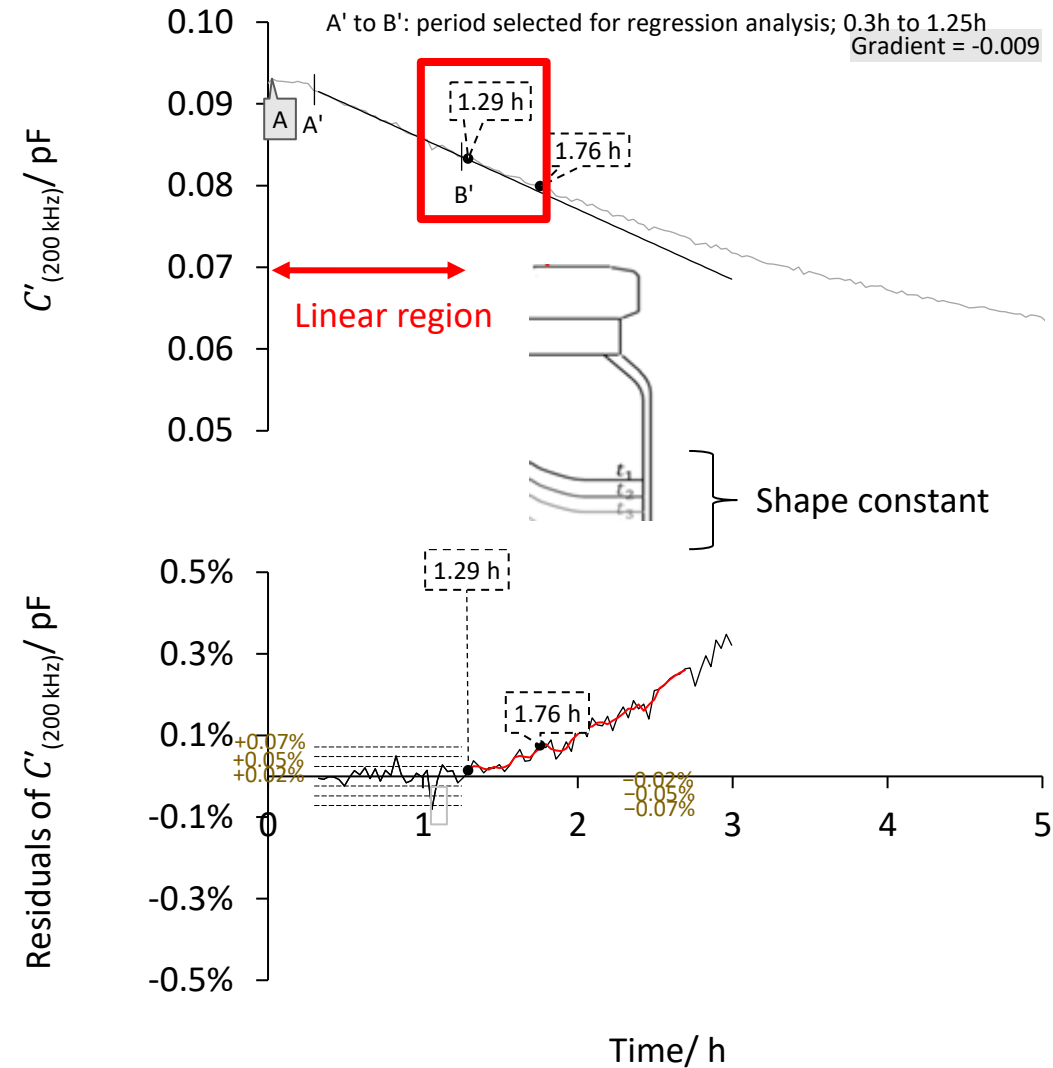
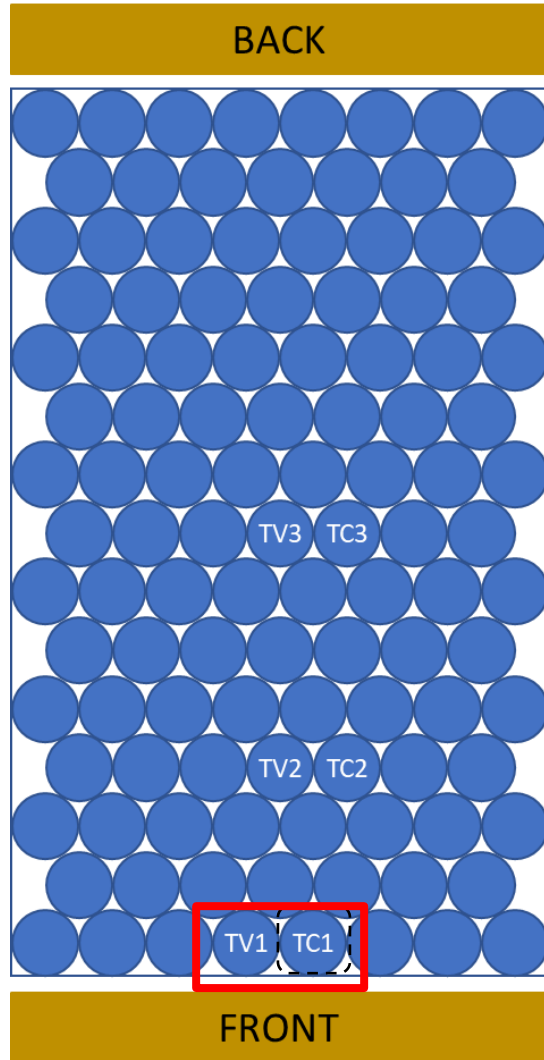
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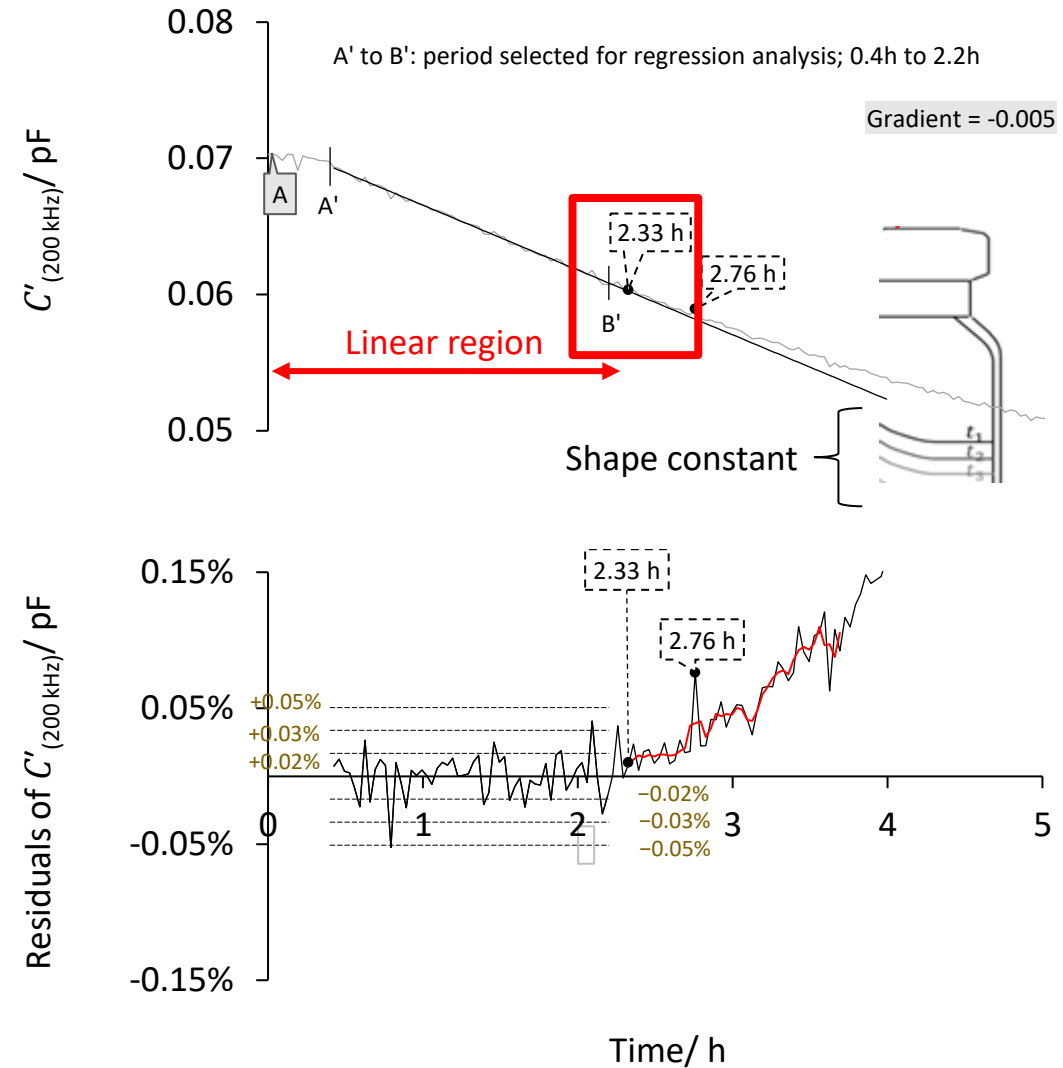
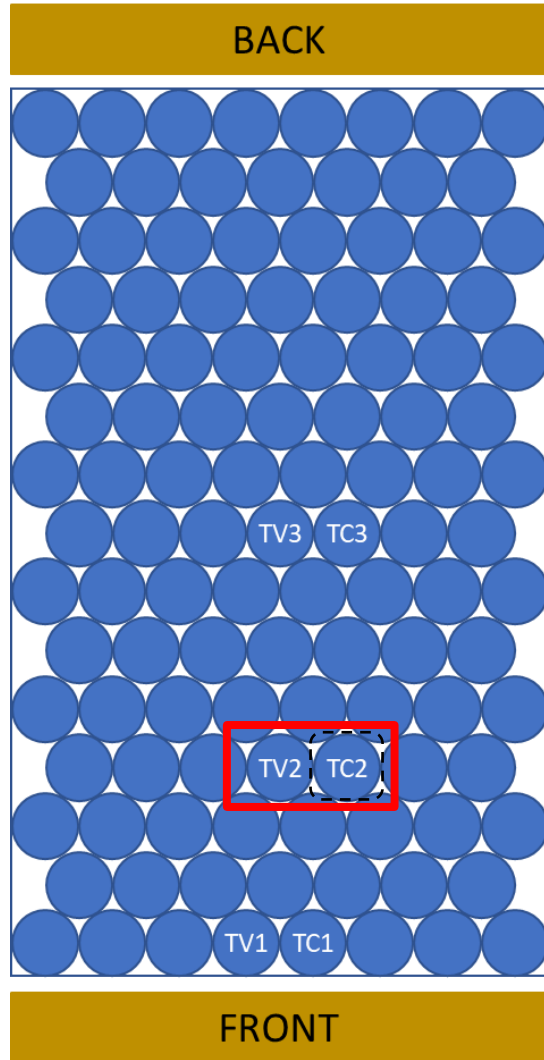
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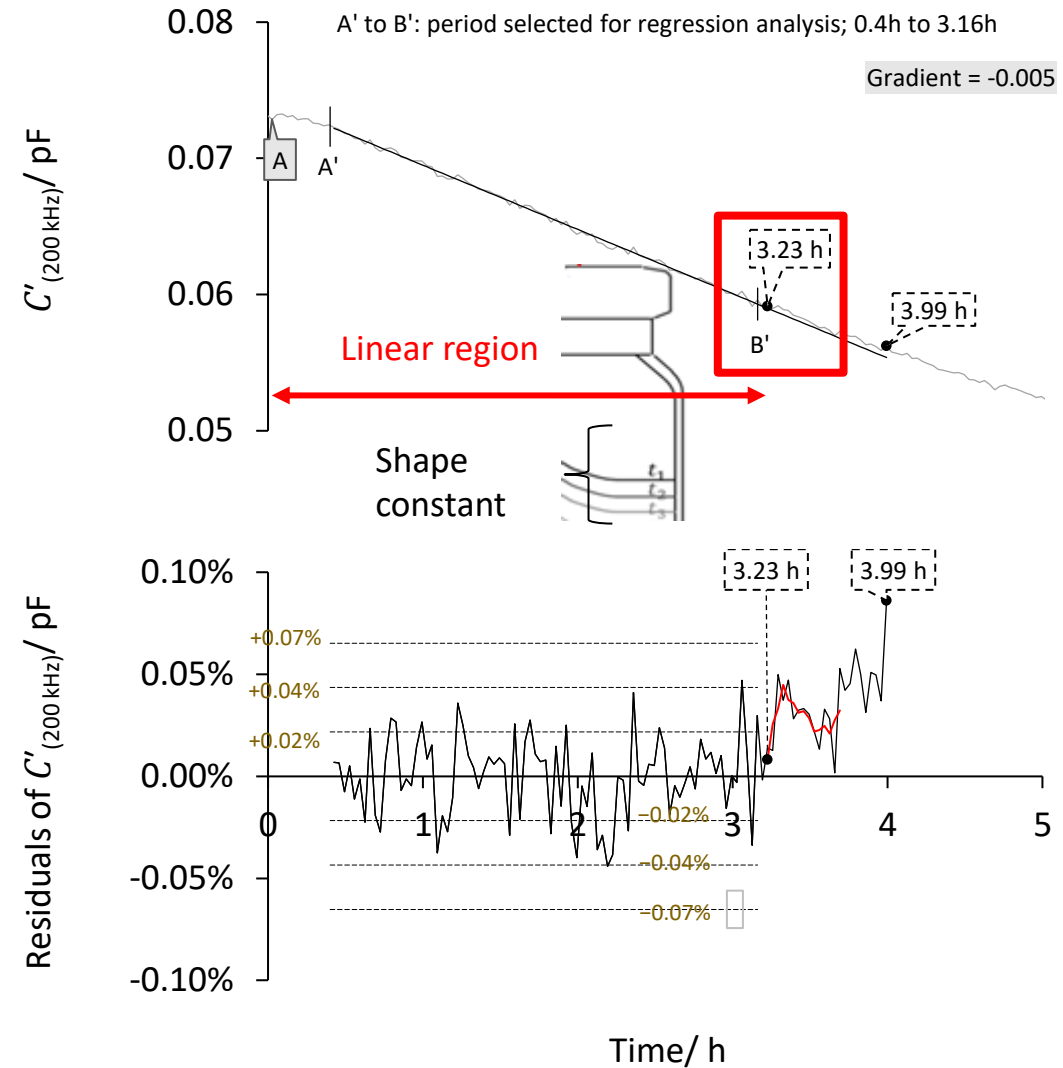
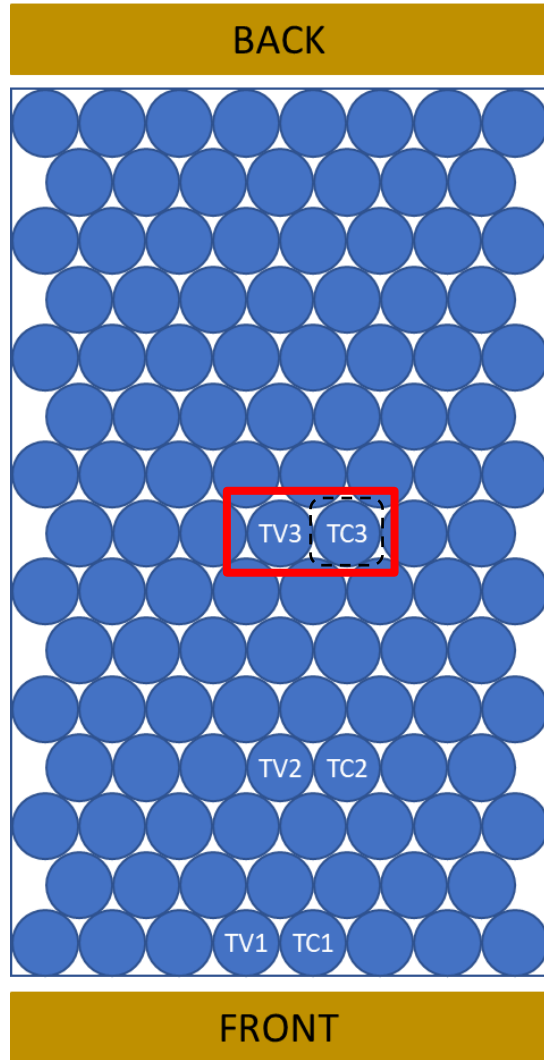
Front of shelf: linear for 1.29 h



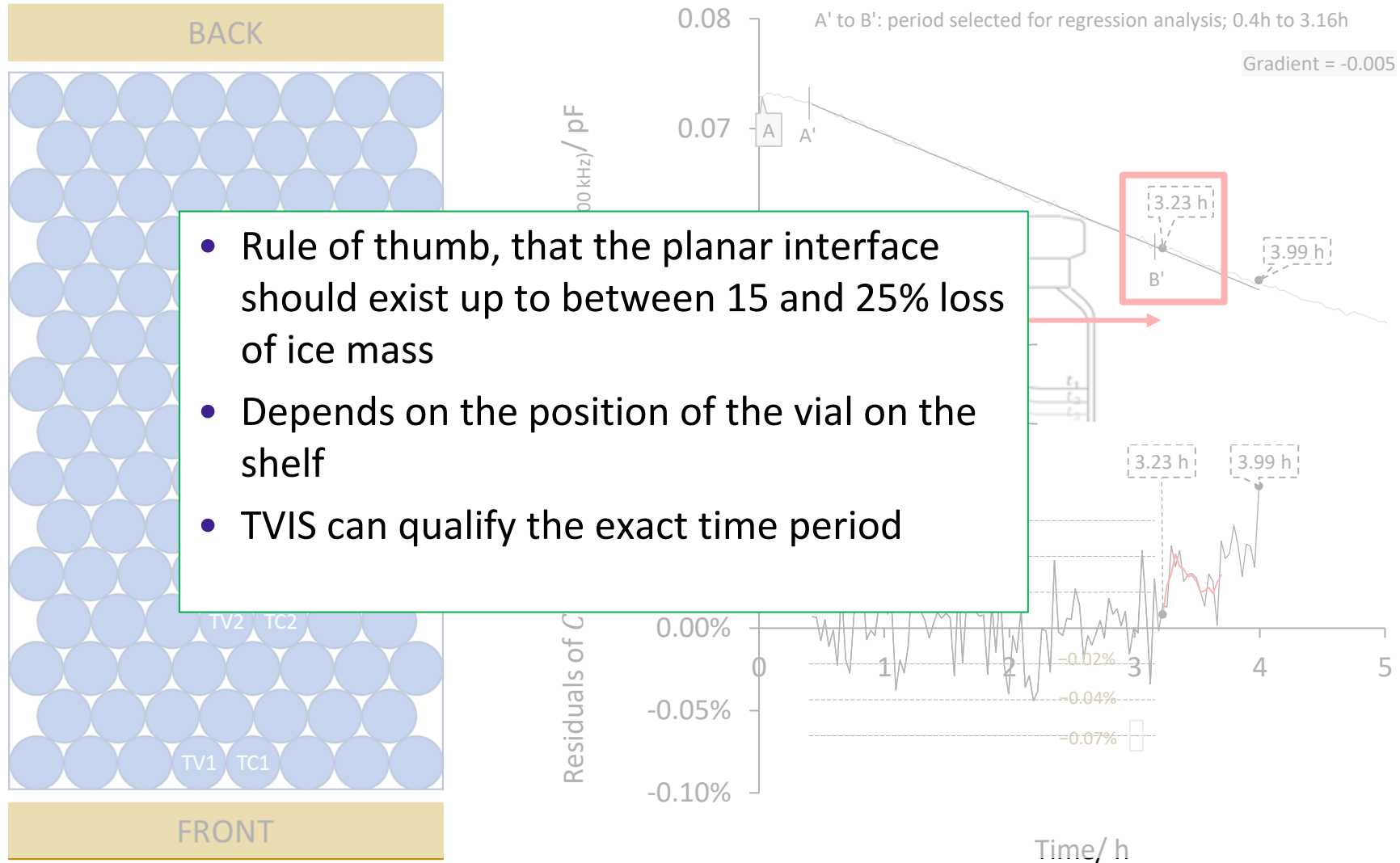
Mid-way to the centre : linear for 2.33 h



Centre of the shelf : linear for 3.23 h



Centre of the shelf : linear for 3.23 h



Take home messages from this talk

TVIS provides

- Temperature prediction for primary drying
- Non-invasive determination of ice nucleation temperature (and ice solidification end point)
- Identification of primary drying end point
- Qualification of batch process models (MTM, TDLAS) in terms of the assumptions in the model (planar ice interface)



Summary

Dielectric loss peak		Dielectric constant	
Log peak frequency (F_{PEAK})	Temperature calibration (ice phase)	Low frequency (100 Hz)	Ice nucleation onset time and temperature
	Spatial measurements of ice temperature possible with multiple nodes		
Peak amplitude (C''_{PEAK})	Ice mass & sublimation rate	High frequency (100 kHz)	Ice solidification end point
	Annealing end-point		
			Glass transition temperature
			Devitrification
			Sublimation end point



Acknowledgements



Evgeny Polygalov
Physicist and Inventor of TVIS
1952-2020



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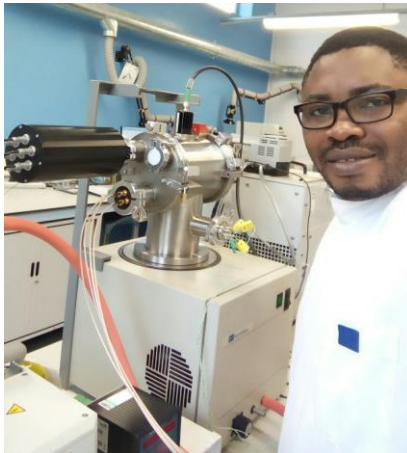
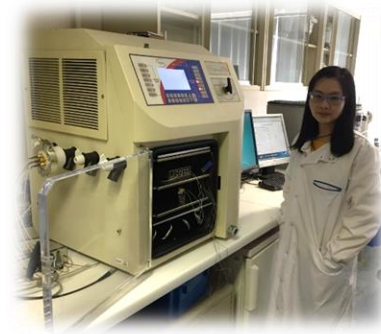
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References

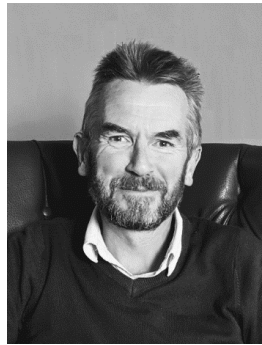
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Thank you for listening!



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